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Spongy texture in mantle clinopyroxene records decompression-induced melting

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ABSTRACT

Spongy texture in clinopyroxene is found in mantle xenolith suites worldwide while its origin commonly is unclear. Detailed petrologic observations and major- and trace-element analysis were conducted on peridotite xenoliths with spongy clinopyroxene, from the central part of the Great Xing'an Range, in the eastern part of the Central Asian Orogenic Belt, to investigate the origin of the texture. Spongy texture mainly occurs as rims of variable thickness on clean cores and occasionally extends into the inner parts of clinopyroxene or even totally covers the whole grain; there is no obvious relationship between the thickness of the rims and their proximity to the host basalt. The boundaries between spongy-textured clinopyroxenes and surrounding minerals are sharp. Spongy domains consist of secondary clinopyroxenes, glasses, yugs and traces of olivine. Relative to the primary ones, clinopyroxenes in spongy domains have low Na₂O, Al₂O₃ and Al^{VI}/Al^{IV}, and high CaO, Mg[#] and Cr[#]. The estimated pressures of spongy domains are 0.34-0.95 Gpa, significantly lower than those of primary peridotite phases (2.10–2.44 Gpa). Glasses in spongy domains are rich in Na₂O and Al₂O₃ and depleted in MgO and FeO, containing a small amount of K2O. Some glasses have low Mg[#] (down to 43.9), and are not in equilibrium with associated clinopyroxenes and olivines. Based on these observations, we suggest that the spongy texture in the studied peridotites was recently formed by decompression-induced low-degree partial melting. By comparing our petrological and chemical data with those of previous studies, textural and chemical criteria are developed to differentiate various possible origins of spongy texture in mantle clinopyroxene.

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1. Introduction

Peridotite xenoliths are an essential source of information about physicochemical properties and evolution of the deep lithosphere from which they were derived (Griffin et al., 1998; Zheng et al., 2007; Zheng. 2009; Tang et al., 2013a). Spongy texture, also called sieved texture (Nelson and Montana 1992; Shaw et al., 2006; Shaw and Dingwell 2008), is common in many mantle xenolith suites in the world, such as the Western Qinling (Su et al., 2011), NW Syria (Ma et al., 2015), Northern Tanzania (Dawson 2002), Hessian Depression (Carpenter et al. 2002), Kenya Rift (Kaeser et al., 2007) and Dariganga lava plateau (Ionov et al., 1994). The texture is frequently glass-bearing and characterized by porous rims or coronae of variable thickness on host minerals in mantle peridotite (mainly on clinopyroxene and spinel, occasionally on orthopyroxene and olivine; Streck 2008).

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During the past decades, the origin of spongy texture in mantle clinopyroxene has been the subject of some controversy. Some authors have attributed spongy texture to metasomatism by fluid/melt penetration before entrainment in the host magma (Bonadiman et al., 2005, 2008; Carpenter et al., 2002; Ionov et al., 1994; Liang and Elthon 1990). In these cases, melt pockets, interstitial glasses or modal metasomatic minerals were observed with the occurrence of spongy-textured clinopyroxene (Bonadiman et al., 2005; Ionov et al., 1994, 2005). Glasses associated with spongy texture have high $Na_2O + K_2O$ (up 15.5 wt%), but low Na₂O/K₂O (<1), CaO (<3 wt%), MgO (<1.63 wt%) and FeO (<1.44 wt%; Bonadiman et al., 2005). In contrast, other authors have argued that interaction between primary orthopyroxene in peridotite xenolith and host magma during transport could produce a Si- and alkali-rich melt, which causes the incongruent dissolution of clinopyroxene and finally results in the formation of spongy texture (Shaw et al., 2006; Shaw and Dingwell 2008). Spongy-textured clinopyroxene, as well as reacted orthopyroxene, are particularly concentrated near contacts with the host magma (Shaw et al., 2006, Shaw and Dingwell 2008). Spongy clinopyroxenes have higher Ca and Cr and lower Na and Al contents, compared with the primary ones





(Shaw et al., 2006, Shaw and Dingwell 2008). Glasses within spongy domain have high CaO (up to 20 wt%), MgO (up to 15.7 wt%) and Mg[#] (78.4–91.4; Shaw et al., 2006, Shaw and Dingwell 2008). Recently, Su et al., (2011) presented petrographical and chemical data on spongy-textured clinopyroxene in peridotites from the West Qinling area. In that study, spongy rims on clinopyroxene are enriched in Ca and depleted in Na and Al. More importantly, Al^{IV}/Al^{VI} of spongy rims is higher than that of intact cores. The authors argued that spongy texture in clinopyroxene developed from decompression-induced partial melting (Su et al., 2011). However, no chemical composition of associated glass was reported.

It thus appears that the origin of spongy texture in mantle clinopyroxene could be related to different processes. Correctly understanding the origin of spongy textures in peridotite minerals is important to reveal the processes operating in the upper mantle and avoid misinterpretations of the mineralogy and chemical compositions of the mantle imposed by post-entrainment lava-xenolith interaction (Ma et al., 2015; Shaw et al., 2006; Su et al., 2011). Therefore, it is necessary to identify criteria to differentiate the origins of these similar spongy textures in mantle clinopyroxene.

Cenozoic basalts are widely distributed in NE China (Fig. 1a), and contain abundant peridotite xenoliths. Previous studies on these xenoliths mainly focused on their formation ages and geochemical characteristics, and discussed the nature and evolution of the lithospheric mantle beneath NE China (Pan et al., 2013, 2015; Wu et al., 2003; Xu et al., 1998; Yu et al., 2009; Zhang et al., 2000, 2011, 2012; Zhou et al., 2007,

Wuchaqou-Halaha Volcanic Fields

2009). Spongy-textured clinopyroxenes in peridotites were recognized in some locations such as Aershan, Shuagliao and Yitong (Fig. 1a; Yu et al. 2007; Zhou et al., 2009; Liu et al., 2014), but little attention was paid to them. In this study, we present detailed petrological and major- and trace-element data on peridotite xenoliths with spongy textures, from the central part of the Great Xing'an Range, in the eastern part of the Central Asian Orogenic Belt. These data allow us to investigate the origin of spongy texture and further develop textural and chemical criteria to differentiate various origins of spongy texture in mantle clinopyroxene, through comparison of our petrological observations and chemical data with those in previous studies.

2. Geological background

120°45'E

The Central Asian Orogenic Belt (called the Altaids in some references), which links the Siberia Craton to the north and the North China Craton to the south, is one of the largest Phanerozoic accretionary orogenic belts on Earth (Sengör and Burtman 1993; Jahn et al., 2000; Kröner et al., 2007; Windley et al., 2007; Wilhem et al., 2012). Tectonically, the Great Xing'an Range is located in the eastern part of the orogen (Fig. 1a). During the Paleozoic, the region was controlled by the Paleo-Asian Ocean tectonic regime and characterized by the amalgamation of several terranes (Wu et al., 2002, 2011; Xiao et al., 2003; Xu et al., 2013). Subsequently, since early Mesozoic, this region has been strongly overprinted by subduction of the Paleo-Pacific and Mongol-Okhotsk Oceans (Wang et al., 2006; Xu et al., 2013; Pan et al., 2014;

121°45′E

121°15′E



Fig. 1. (a) Tectonic framework of eastern China and the location of the studied area (modified after Xu et al. 2004). (b) Sketch map shows distributions of the Wuchagou and Halaha volcanic field (host lava) and the sample locations (modified after Ho et al., 1991). WEK, Wudalianchi-Erkeshan-Keluo; DTGL, Daxing'anling-Taihangshan gravity lineament; TLFZ, Tanlu translithosperic fault zone.

Table 1			
Microstructure and minera	mode of the Aershan	n peridotite xenoliths,	NE China

Sample	Rock	Ol-Mg [#]	Microstructure	Modes (wt%)				
				01	Орх	Срх	Sp	Symplectite
AS03	lherzolite	90.3	porphyroclastic	66	25	6		3
AS04	lherzolite	90.4	porphyroclastic	72	18	7	3	
AS06	lherzolite	88.9	protogranular	58	26	6		10
HLH10	lherzolite	90.1	porphyroclastic	65	20	11		4
DHG04	harzburgite	90.1	porphyroclastic	69	23	5		3
DHG10	lherzolite	89.9	protogranular	63	22	14		1
DHG13	harzburgite	90.9	porphyroclastic	73	22	4	1	

Mineral modes of samples are determined by point counting. Ol, olivine; Opx, orthopyroxene; Cpx, clinopyroxene; Sp, spinel.

Tang et al., 2014). The basement of the range is composed of Paleozoic to Mesozoic strata, including low-grade metamorphosed volcano-sedimentary associations, limestones and clastic rocks, which are intruded by voluminous volcanic rocks and granitoids (Ge et al., 2005; Wu et al., 2001, 2011; Zhang et al., 2010). The formation ages of some previously assumed Precambrian metamorphic complexes are still controversial (Ge et al., 2011; Miao et al., 2004; Shi et al., 2003; Sun et al., 2013a). Recently, Paleoproterozoic (1741–1854 Ma) granitic gneisses and Neoproterozoic (737–851 Ma) granitoids were recognized in the northern part of the Great Xing'an Range (Sun et al., 2013b).

The Cenozoic volcanic field in the central part of the Great Xing'an Range can be divided into two domains, defined as the Wuchagou volcanic field in the south and the Halaha volcanic field in the north (Fig. 1b). The latter contains abundant peridotite xenoliths, which are rarely found in the former. The Halaha volcanic rocks are alkali basalts and accumulated mainly above the valleys of local rivers, forming a low lava platform. The alkali basalts, with K—Ar ages of 2.0–0.2 Ma, represent the melting products of convective mantle (Ho et al., 2013). The studied peridotites were collected from the valley of the Halaha River, near the city of Aershan (Fig. 1b).

3. Petrography

The studied xenoliths are fresh, rounded to sub-angular but small, predominantly <5 cm in diameter. All the peridotites are anhydrous and contain no metasomatic phases (e.g. amphibole, mica or apatite). The modal compositions of these peridotites have been determined by point-counting with >500 points in each thin section. A summary of their textures and mineral assemblages is given in Table 1.

Spongy-textured clinopyroxene grains occur widely and are randomly distributed across the thin sections. They do not show preferential development (i.e. thicker spongy rims) toward the contact between the peridotite and host basalt (Fig. 2). Most spongy-textured clinopyroxenes are composed of clean cores and spongy rims with thickness from 20 to 250 μ m (Fig. 3a). Occasionally, spongy domains extend into the inner part of the host clinopyroxene (Fig. 3b) or even totally cover the whole grain. It appears that the development of spongy domains is not controlled by cracks across clinopyroxene (Fig. 3c). The boundaries between spongy-textured clinopyroxenes and surrounding mineral grains are sharp. Spongy-textured clinopyroxenes also display well-defined contacts between the spongy rims and the intact cores, and the latter do not show any zoning pattern in back-scattered electron images (Fig. 3d and e).

Spongy domains are mainly composed of secondary clinopyroxenes, glasses and vugs. Clinopyroxene in the spongy part is brighter than the corresponding intact core in back-scattered electron images (Fig. 3a, d and e). They share optical continuity under cross-polarized light (Fig. 3f). Glasses are dark grey. Vugs are always empty and vary from round to worm-like shapes, or even form net-channels (Fig. 3a, c, d and e). Traces of olivine (<3 vol% of the spongy domain), are anhedral to euhedral and accompanied by glasses or vugs (Fig. 3d and e), and are ubiquitous in spongy domains of most samples. Samples DHG13 and AS04 are spinel-bearing harzburgite and lherzolite, respectively. Other samples (AS03, AS06, DHG04, DHG10 and HLH10) contain symplectite (Fig. 4a) which consists of tiny spinels, orthopyroxenes and quenched glasses (Fig. 4b). The relationship between symplectite and spongy-textured clinopyroxene is hard to investigate because



Fig. 2. Scanned thin section of xenolith (sample AS06). OI: olivine; Opx: orthopyroxene; Cpx: clinopyroxene. Numbers in brackets are average thickness (nm) for each spongy rim on clinopyroxene.



Fig. 3. BSE micrographs (a, b, c, d and e) and photomicrograph (f) of spongy-textured clinopyroxene in the Aershan peridotites. (a) Spongy rim of clinopyroxene in AS04 showing variable thickness. (b) Spongy domain extends into the inner part of host clinopyroxene in DHG04. (c) Cracks crosscut a clinopyroxene grain in DHG04. (d) and (e) Spongy domain consists of secondary olivines, clinopyroxenes, glasses and variable-shaped vugs. (f) A spongy-textured clinopyroxene grain in DHG04 under cross-polarized light. OI: olivine; Cpx: clinopyroxene.

they are not connected with each other. The inner parts of some symplectites are composed of ultra-fine mineral grains ($<1 \mu m$).

4. Analytical methods

Major-element analyses of primary minerals and different materials in spongy domains were carried out at the State Key Laboratory of Geological Processes and Mineral Resources, China University of Geosciences (Wuhan), with a JEOL JXA-8100 Electron Probe Micro Analyzer equipped with four wavelength-dispersive spectrometers (WDS). The samples were coated with a thin conductive carbon film prior to analysis. The precautions suggested by Zhang and Yang (2016) were used to minimize the difference of carbon film thickness between samples and obtain an approximately uniform coating ca. 20 nm thick.



Fig. 4. (a) BSE micrograph of a symplectite in DHG04. (b) Enlarged image of the rectangle in (a), showing the symplectite consists of Opx, Sp and glass. OI: olivine; Opx: orthopyroxene; Sp: spinel.



Fig. 5. Plots of Mg[#] vs (a) CaO and (b) NiO in primary olivines and secondary ones in spongy area. The compositional boundary between mantle olivine and cumulate olivine is from Thompson and Gibson (2000).

During the analysis, an accelerating voltage of 15 kV, a beam current of 20 nA and a 5 μ m spot size were used to analyze minerals. Data were corrected on-line using a modified ZAF (atomic number, absorption, fluorescence) correction procedure. The peak counting time was 10 s for Na, Mg, Al, Si, K, Ca, Fe, Cr, Ni and 20 s for Ti and Mn. The background counting time was one-half of the peak counting time on the high- and low-energy background positions. The following standards were used: Sanidine (K), Pyrope Garnet (Fe, Al), Diopsode (Ca, Mg), Jadeite (Na), Rhodonite (Mn), Olivine (Si), Rutile (Ti), Chromium oxide (Cr).

Trace-element analyses of primary clinopyroxenes and orthopyroxenes were obtained by LA-ICP-MS at the State Key Laboratory of Geological Processes and Mineral Resources, China University of Geosciences, Wuhan. Laser sampling was performed using a GeoLas2005 with laser beam diameter of 44 µm. The ICPMS is an Agilent 7700. Each analysis incorporated a background acquisition of approximately 20–30 s (gas blank) followed by 50 s data acquisition from the sample. The NIST 610 glass and the USGS glasses were used as external standards. No internal standard was applied for correction. Additional descriptions of instrument operating conditions, calibration values, detection limits, and error analysis for the laser microprobe are reported by Liu et al., (2008).

5. Analytical results

5.1. Major-element compositions

5.1.1. Primary minerals

The primary minerals are homogeneous, based on several core and rim analyses in each sample. Their major-element compositions are listed in Appendix 1.

The primary olivine in the studied peridotites has high NiO and low CaO contents (Fig. 5). Its Mg[#] varies from 88.9 to 90.9 (Fig. 5). The Al₂O₃ contents of orthopyroxene range from 3.99 to 5.47 wt% and broadly decrease with increasing Mg[#]. Spinel only occurs in samples DHG13 and AS04, with Cr[#] of 26.5–41.8 and Mg[#] of 67.0–69.3. All primary clinopyroxenes are Cr-diopside with Cr₂O₃ contents of 0.98–1.60 wt%, displaying a negative correlation of Mg[#] vs Al₂O₃ but a positive correlation of Mg[#] vs CaO (Fig. 6).



Fig. 6. Plots of Mg[#] vs Na₂O (a), Al₂O₃ (b), CaO (c) and Cr[#] (d) in primary clinopyroxenes and secondary ones in spongy domain. CPXP: primary clinopyroxene; CPXS: secondary clinopyroxene in spongy domain; Cpx in Mls: clinopyroxene in melt inclusions. Data sources: Shaw CPXP and Shaw CPXS are from Shaw et al. (2006) and Shaw and Dingwell. (2008); Bonadiman CPXP and Bonadiman CPXP and Bonadiman CPXS are from Bonadiman et al. (2005); Su CPXP and Su CPXS are from Su et al. (2011); Cpx in Mls are from Ionov et al. (2011) and Bénard et al. (2016).

Table 2

Electron microprobe analyses (wt%) of materials in spongy domain of clinopyroxene of the Aershan peridotites.

Sample	AS03			AS04		AS06	AS06		HLH10	
	Срх	Glass	Ol	Срх	Glass	Срх	Glass	Ol	Срх	Glass
Number	10	6	3	11	5	10	5	3	10	5
Na ₂ O	0.42	5.02	0.02	0.50	5.46	0.39	6.07	0.03	0.39	5.14
K ₂ O	0.01	0.38	b.d.	0.01	0.52	0.01	0.64	0.00	0.00	0.57
FeO	3.09	0.48	10.20	3.31	0.44	3.58	0.48	11.75	3.33	0.76
MgO	17.60	0.27	46.92	17.25	0.27	16.99	0.21	47.11	17.85	0.60
CaO	20.20	11.05	0.32	19.62	10.01	19.65	9.28	0.35	20.31	10.66
Al ₂ O ₃	3.20	27.16	0.05	4.25	26.44	4.09	26.41	0.07	3.82	27.30
SiO ₂	52.59	53.94	40.81	52.05	55.62	51.45	56.71	39.42	51.94	54.48
Cr_2O_3	1.14	0.06	0.07	1.41	0.05	1.80	0.06	0.19	1.39	0.02
TiO ₂	0.43	0.10	0.01	0.62	0.19	0.56	0.15	0.03	0.43	0.12
MnO	0.05	0.04	0.31	0.07	0.02	0.07	0.01	0.27	0.09	0.03
NiO	0.06	0.04	0.15	0.10	0.05	0.09	0.00	0.13	0.03	0.01
Total	98.79	98.53	98.84	99.18	99.08	98.69	100.01	99.34	99.58	99.69
Mg [#]	91.1	50.4	89.2	90.4	52.1	89.5	43.8	87.8	90.6	58.8
Cr [#]	19.1			18.0		22.6			19.4	
^{IV} Al	0.05			0.09		0.10			0.09	
^{VI} Al	0.09			0.09		0.08			0.08	
^{VI} Al/ ^{IV} Al	1.62			0.95		0.78			0.84	
C	111110	DUCCA			DUCIO			DUCID		
Sample	HLHIU	DHG04			DHG10			DHG13		
Sample	Ol	Cpx	Glass	Ol	Cpx	Glass	Ol	Cpx	Glass	
Number	Ol 3	 Срх 12	Glass 7	Ol 8	Cpx 11	Glass 6	0l 4	Cpx 12	Glass 3	
Number Na ₂ O	Ol 3 0.02	DHG04 Cpx 12 0.47	Glass 7 5.09	Ol 8 0.03	Cpx 11 0.35	Glass 6 4.50	0l 4 0.03	DHG13 Cpx 12 0.32	Glass 3 4.92	
Number Na ₂ O K ₂ O	Ol 3 0.02 0.00	DHG04 Cpx 12 0.47 0.01	Glass 7 5.09 0.22	Ol 8 0.03 0.01	DHG10 Cpx 11 0.35 0.01	Glass 6 4.50 0.17	Ol 4 0.03 b.d.	DHG13 Cpx 12 0.32 0.01	Glass 3 4.92 0.47	
Number Na2O K2O FeO	Ol 3 0.02 0.00 13.44	DHG04 Cpx 12 0.47 0.01 3.84	Glass 7 5.09 0.22 0.57	Ol 8 0.03 0.01 10.81	DHG10 Cpx 11 0.35 0.01 3.71	Glass 6 4.50 0.17 0.66	Ol 4 0.03 b.d. 10.92	DHG13 Cpx 12 0.32 0.01 3.12	Glass 3 4.92 0.47 0.44	
Number Na2O K2O FeO MgO	0l 3 0.02 0.00 13.44 45.50	DHG04 Cpx 12 0.47 0.01 3.84 19.41	Glass 7 5.09 0.22 0.57 0.42	Ol 8 0.03 0.01 10.81 46.56	DHG10 Cpx 11 0.35 0.01 3.71 18.61	Glass 6 4.50 0.17 0.66 0.52	Ol 4 0.03 b.d. 10.92 46.56	DHG13 Cpx 12 0.32 0.01 3.12 18.27	Glass 3 4.92 0.47 0.44 0.19	
Number Na ₂ O K ₂ O FeO MgO CaO	01 3 0.02 0.00 13.44 45.50 0.31	DHG04 Cpx 12 0.47 0.01 3.84 19.41 17.79	Glass 7 5.09 0.22 0.57 0.42 10.89	Ol 8 0.03 0.01 10.81 46.56 0.28	DHG10 Cpx 11 0.35 0.01 3.71 18.61 18.14	Glass 6 4.50 0.17 0.66 0.52 11.90	Ol 4 0.03 b.d. 10.92 46.56 0.27	DHG13 Cpx 12 0.32 0.01 3.12 18.27 19.62	Glass 3 4.92 0.47 0.44 0.19 11.02	
Number Na2O K2O FeO MgO CaO Al2O3	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03	DHG04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11	DHG10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98	Class 3 4.92 0.47 0.44 0.19 11.02 27.59	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87	DHG10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09	DHG04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64	Glass 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02	DHG04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35	Glass 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16	
Number Na20 K20 Fe0 Mg0 Ca0 Al203 Si02 Cr203 Ti02 Mn0	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30	DHG10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08	Glass 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00	
Number Na2O K2O FeO MgO CaO Al2O3 SIO2 Cr2O3 TiO2 MnO NiO	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.02 0.24 0.16	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03 0.04	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09	Glass 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂ MnO NiO Total	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24 0.16 99.67	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.9 0.03 0.04 99.61	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07 99.13	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂ MnO NiO Total Mg [#]	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.24 0.16 99.67 85.9	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33 90.1	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40 56.8	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26 88.6	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10 90.0	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.03 0.04 99.61 58.5	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22 88.5	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45 91.3	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07 99.13 44.0	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂ MnO NiO Total Mg [#] Cr [#]	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24 0.16 99.67 85.9	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33 90.1 18.3	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40 56.8	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26 88.6	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10 90.0 17.8	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03 0.04 99.61 58.5	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22 88.5	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45 91.3 26.7	Glass 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.07 99.13 44.0	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂ MnO NiO Total Mg [#] Cr [#] I ^V Al	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24 0.16 99.67 85.9	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33 90.1 18.3 0.11	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40 56.8	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26 88.6	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10 90.0 17.8 0.08	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03 0.04 99.61 58.5	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22 88.5	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45 91.3 26.7 0.04	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07 99.13 44.0	
Number Na2O K2O FeO MgO CaO Al2O3 SiO2 Cr2O3 TiO2 MnO NiO Total Mg [#] Cr [#] I'VAI V ¹ AI	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24 0.16 99.67 85.9	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33 90.1 18.3 0.11 0.05	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40 56.8	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26 88.6	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10 90.0 17.8 0.08 0.09	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03 0.04 99.61 58.5	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22 88.5	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45 91.3 26.7 0.04 0.08	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07 99.13 44.0	
Number Na ₂ O K ₂ O FeO MgO CaO Al ₂ O ₃ SiO ₂ Cr ₂ O ₃ TiO ₂ MnO NiO Total Mg [#] Cr [#] I'V Al V ^I Al V ^I Al	ALHIO Ol 3 0.02 0.00 13.44 45.50 0.31 0.03 39.86 0.09 0.02 0.24 0.16 99.67 85.9	DHC04 Cpx 12 0.47 0.01 3.84 19.41 17.79 3.88 51.99 1.32 0.42 0.09 0.10 99.33 90.1 18.3 0.11 0.05 0.47	Glass 7 5.09 0.22 0.57 0.42 10.89 27.07 54.91 0.05 0.09 0.04 0.05 99.40 56.8	Ol 8 0.03 0.01 10.81 46.56 0.28 0.11 40.87 0.09 0.02 0.30 0.18 99.26 88.6	DHG 10 Cpx 11 0.35 0.01 3.71 18.61 18.14 3.96 52.50 1.29 0.36 0.08 0.09 99.10 90.0 17.8 0.08 0.09 1.07	Glass 6 4.50 0.17 0.66 0.52 11.90 27.72 53.95 0.04 0.09 0.03 0.04 99.61 58.5	Ol 4 0.03 b.d. 10.92 46.56 0.27 0.04 40.82 0.04 0.02 0.38 0.16 99.22 88.5	DHC13 Cpx 12 0.32 0.01 3.12 18.27 19.62 2.98 52.97 1.64 0.35 0.08 0.09 99.45 91.3 26.7 0.04 0.08 1.88	Class 3 4.92 0.47 0.44 0.19 11.02 27.59 54.27 0.00 0.16 0.00 0.07 99.13 44.0	

Ol: olivine; Opx: orthopyroxene; Cpx: clinopyroxene; Sp: spinel; b.d.: below detection limit.

5.1.2. Spongy domains in clinopyroxene

Major-element compositions of different materials (clinopyroxene, glass and olivine) in the spongy domain of clinopyroxene are given in Table 2. Relative to the primary one, secondary olivine in spongy domains has apparently low Mg[#] and NiO but high CaO (Fig. 5). Compared with the primary one, secondary clinopyroxene in spongy domains has obviously low Al₂O₃ and Na₂O but high CaO (Figs. 6 and 7). The SiO₂ and TiO₂ concentrations are similar and do not show systematic differences. Glasses in spongy domains show a narrow range in Al₂O₃ (26.4–27.7 wt%), CaO (9.3–11.9 wt%) and SiO₂ (53.9–56.7 wt%). It is poor in MgO (0.19–0.52 wt%), FeO (0.44–0.76 wt%) and TiO₂ (0.09–0.19 wt%) but rich in Na₂O (4.50–6.07 wt%), with a small amount of K₂O (0.17–0.64 wt%). The Mg[#] of glass is low and varies from 43.9 to 58.8.

5.2. Trace-element compositions

Trace-element compositions of primary clinopyroxene and orthopyroxene are listed in Appendix 2. Chondrite-normalized REE patterns for primary clinopyroxene are convex-upward, with peaks at Sm, Nd and Eu (Fig. 8a). The extended trace-element plots show that the primary clinopyroxene of sample DHG13 has strong negative anomaly in Ti and low Ti/Eu (856). Other peridotites have high Ti/Eu, >4410. All clinopyroxenes have low LREE/HREE ((La/Yb)_N < 5). REE patterns of orthopyroxenes, with positive anomalies in HFSE, show flat HREE to

MREE patterns and continuously decreasing normalized abundances from MREE to LREE (Fig. 8b).

6. Discussion

6.1. Early enrichment processes

Clinopyroxene is the major carrier for most trace elements in anhydrous peridotite peridotite, and its trace-element compositions can be used to reveal mantle processes, such as partial melting and metasomatism (Zangana et al., 1999). Recent studies suggested that partition coefficient of REE between orthopyroxene and clinopyroxene (DREE Opx/Cpx) is negatively correlated to cation radius but positively correlated to equilibrium temperature of peridotite (Witt-Eickschen and O'Neill, 2005; Lee et al., 2007). Although the measured partition coefficient of REE between primary clinopyroxene and orthopyroxene broadly increase with decreasing radius (from La to Lu), they are highly variable and do not show positive correlations with equilibrium temperatures (Appendix 2), implying the REE concentrations of primary clinopyroxene are disturbed and can hardly be used to estimate the degree of melt extraction experienced by the studied peridotites.

As no hydrous phases were identified in the studied xenoliths, moderate enrichment of incompatible elements, such as LREE and Sr, is consistent with a cryptic metasomatism. In particular, the convex-upward REE patterns of clinopyroxene, with the peak at Sm, Nd or Eu (Fig. 8a),



Fig. 7. Compositional mapping of spongy-textured clinopyroxene grain in Fig. 3c. Color bar in the right indicates the relative contents of the selected elements.

are diagnostic of equilibrium with LREE-rich melts (Bodinier et al., 1990; Navon and Stolper 1987). All primary clinopyroxenes, except for those of sample DHG13, with low (La/Yb)_N (< 5) and high Ti/Eu (> 4410), record the signatures of silicate-melt related metasomatism (Coltorti et al., 1999). Mg[#] of clinopyroxene in samples AS06 and DHG10 is <89.0, possibly reflecting the addition of iron after partial melting. Clinopyroxene of sample DHG13 has low Ti/Eu of 856 and (La/Yb)_N of 0.23. The patterns are similar to those of clinopyroxene in peridotites from West Tianshan, which have been interpreted to be produced by a metasomatic agent transitional between carbonatitic and

silicate melt, or overprinted by multiple episodes of metasomatism (Zheng et al., 2006). Fig. 8a shows that the melts in equilibrium with primary clinopyroxenes had lower REE abundances than the host basalts. We therefore consider that the studied xenoliths had experienced enrichment processes before entrainment.

6.2. Origin of spongy texture in the Aershan peridotites

The origin of spongy texture in mantle clinopyroxene has long been controversial and attributed to interaction between peridotite and host magma (Qi et al., 1995; Shaw et al., 2006; Shaw and Dingwell 2008), decompression-induced melting (Su et al., 2011) and mantle metasomatism by fluid/melt penetration before entrainment (Bonadiman et al., 2005, 2008; Carpenter et al., 2002; Ionov et al., 1994). Based on petrological observations and chemical analysis, decompression-induced partial melting is considered as the most possible mechanism for the formation of spongy texture in the studied peridotites. The evidences are as below:

1. Most studied xenoliths contain symplectite which consists of tiny spinels, orthopyroxenes and quenched glasses. Such symplectites are generally interpreted as the products of breakdown of previous garnet due to pressure decrease (Godard and Martin 2000; Morishita and Arai 2003; Obata 2011; Obata and Ozawa 2011; Špaček et al., 2013). The low CaO content (17.5–20.4 wt%) of the primary clinopyroxene is also considered as reflecting equilibrium with the garnet phase (Bonadiman et al., 2005).

2. Secondary clinopyroxene in spongy domain is characterized by lower Al_2O_3 and Na_2O , but higher CaO, Cr[#] and Mg[#] than the primary one (Fig. 6). The decrease in Al_2O_3 and increase in Mg[#] and Cr[#] of clinopyroxene are frequently interpreted as effective indicators of degree of partial melting (e.g. Frey and Prinz 1978; Zheng et al., 2001). Therefore, secondary clinopyroxene in spongy domain could represent residues after partial melting of primary clinopyroxene. The relative proportions of Al^{IV} and Al^{VI} in clinopyroxene can provide useful information on pressure differences (Wass 1979). Primary clinopyroxene (for example, the intact core) has systematically higher Al^{VI} , and more significantly, higher Al^{VI}/Al^{IV} ratios than that of secondary clinopyroxene in spongy domains (1.35–2.57 vs 0.47–1.88, Table 2 and Appendix 1), implying a decompression process. The lower content of Na in secondary clinopyroxene in spongy domains also supports the lower-pressure origin (e.g. Ma et al., 2015).

3. Several thermometers have been adopted to estimate the equilibrium temperatures of the studied peridotites. These thermometers give similar temperatures for each sample (Table 3), with a discrepancy of 3–58 °C, which can generally be reconciled with the uncertainties of used thermometers (for example, 70 °C for Mg-Fe-two-pyroxene thermometer, Wells 1977). Recently, Fan et al., (2008) reported a fresh garnet peridotite xenolith collected near the Chaoer River in the Halaha

able 3							
Estimated temp	eratures (°C) and	pressures ((Gpa)	for t	the Aershan	peridotites.

Sample	Primary	phases			Spongy domain		
	T _{BK-ca}	T _{DE}	T _{Wells}	Р	T _{Putirka-32d}	P _{Putirka-32a}	
AS03	1036	1052	1063	2.10	1075	0.95	
AS04	1067	1085	1112	2.16	1105	0.57	
AS06	1160	1136	1152	2.35	1095	0.58	
HLH10	1073	1119	1061	2.18	1098	0.34	
DHG04	1173	1201	1181	2.38	1227	0.50	
DHG10	1186	1244	1221	2.40	1219	0.49	
DHG13	1201	1198	1204	2.44	1127	0.50	

 T_{BK-Ca} : Ca in Opx thermometer, Brey and Köhler (1990); T_{DE} : Al in Ol thermometer, De Hoog et al. (2010).

 $T_{Wells}:$ Mg-Fe-two-pyroxene thermometer, Wells (1977). P for primary phases are estimated from T_{BK-Ca} assuming the studied xenoliths shared the same geothermal gradient with those from Fan et al. (2008).

P_{Putirka-32a} and P_{Putirka-32d}: single clinopyroxene thermobarometer, Eqs 32a and 32d from Putirka, (2008).



Fig. 8. Chondrite-normalized REE patterns (a) and primitive-mantle-normalized traceelement diagrams for primary clinopyroxenes in the studied xenoliths (b). Melt in equilibrium with primary clinopyroxene was calculated using partition coefficients of Hart and Dunn (1993). Host basalt is from Ho et al. (2013). Normalizing values are from McDonough and Sun (1995).

volcanic field (about 35 km away from our sample location; Fig. 1b), with a temperature-pressure estimate of 1164 °C and 2.36 Gpa. Assuming that the studied xenoliths shared the same geothermal gradient, and accepting their T_{BK-Ca} (1036–1201 °C) as equilibration temperatures, the pressures of origin would be 2.10–2.44 Gpa. Temperature and pressure for spongy domains can be estimated using the clinopyroxene-only thermobarometer (Eqs 32a and 32d in Putirka 2008), which gives slightly higher temperatures (1175 °C to 1292 °C) but significantly lower pressure (0.34–0.95 Gpa) than those for primary phases. Although pressure estimation for peridotites employing the iterative method of Putirka (2008) carries some uncertainty (Ma et al., 2015) and the estimated pressures for spongy clinopyroxene show large errors (up to 0.6 Gpa), we still suggest that the systematically lower pressures are indicative.

Glasses in spongy domains probably were produced by low-degree melting of primary clinopyroxene. They are rich in Na₂O, SiO₂ and Al₂O₃, similar to low-degree melts obtained by experiments on anhydrous peridotite at pressure of 1.0–1.5 Gpa (Baker et al., 1995; Draper and Green 1997; Robinson et al., 1998). Low-degree melts of anhydrous peridotite at high pressure (e.g. 2.5 Gpa) is characterized by low SiO₂ (e.g. < 50 wt%) and alkali contents because the activity of SiO₂ decreases due to more compatibility of alkali oxide in clinopyroxene at greater depth (Hirschmann et al., 1998; Schiano et al., 2000). Compared with interstitial glasses, melt inclusions in mantle minerals are considered to be preserved in closed systems and devoid of post-entrapment modification. Therefore, homogenized melts inclusions can be used to determine primitive magmas percolating the lithosphere (Bénard et al., 2016; Ionov et al., 2011; Schiano et al., 1998). Glasses studied here

have extremely low MgO and FeO contents (both <1 wt%), apparently deviating from the field defined by low-degree melts of anhydrous peridotite and homogenized melt inclusions from Avacha volcano (LD melts and GL-MIs in Fig. 9). The low MgO and FeO contents may be due to crystallization of olivine (presence of olivine traces in spongy domain). More importantly, glasses in some studied xenoliths have low Mg[#] (down to 43.8), out of equilibrium with coexisting secondary clinopyroxenes and olivines, assuming the KCpx/lid D Fe/Mg and KOl/ lid D Fe/Mg to be 0.28 and 0.30, respectively (Putirka 2008). The lack of equilibrium, combined with ubiquitous presence of glasses, implies that the spongy texture was formed quickly and recently. The decompression event possibly was related to the tectonic extension which resulted in the eruption of the widely distributed Cenozoic alkali basalts across the NE China (e.g. the host magma). Since Late Cretaceous time, the NE China has experienced a period of magmatic relaxation until the eruption of the basalts (Zhang et al., 2000).

6.3. Differentiating various origins of spongy texture in mantle clinopyroxene

As mentioned above, three processes have been considered to be responsible for the formation of spongy texture in mantle clinopyroxene, including interaction between peridotite and host magma (Shaw et al., 2006; Shaw and Dingwell 2008); metasomatism by fluid/melt penetration (Bonadiman et al., 2005); decompression-induced melting (Su et al., 2011; this study). Here, we have summarized the petrologic and chemical differences of spongy-textured clinopyroxene formed by the three processes (Table 4). The comparison allows us to establish textural and chemical criteria to identify the origins of these similar spongy textures in mantle clinopyroxene.

1. Petrologic differences: The interaction between orthopyroxene and host magma during transport could produce a Si- and alkali-rich melt, which is a necessary precursor to formation of spongy texture in clinopyroxene (Shaw et al., 2006; Shaw and Dingwell 2008). In those studies, spongy-textured clinopyroxene, as well as reacted orthopyroxene, is particularly concentrated near contacts with the host magma (Shaw et al., 2006; Shaw and Dingwell 2008). In contrast, orthopyroxenes in peridotites studied here are generally well-preserved (Appendix 3). Spongy-textured clinopyroxenes are randomly distributed in the thin section and do not exhibit preferential development toward the contacts between peridotite and host basalts (Fig. 2). This is consistent with peridotite xenoliths from the Western Qinling area, in which the spongy texture was interpreted as products of decompression-induced melting of clinopyroxene (Su et al., 2011). Peridotite xenoliths from Sal Island contain spongy-textured clinopyroxenes, which are accompanied with abundant melt pockets, interstitial glasses and modal metasomatic minerals (Bonadiman et al., 2005; Ionov et al. 2005). Kimberlite-like melt infiltration was considered as the mechanism responsible for the spongy texture (Bonadiman et al., 2005). All peridotites studied here are anhydrous and contain no modal metasomatic phases. Spongy domains do not appear to preferentially develop along cracks cross clinopyroxenes (Fig. 3c). These features argue against the infiltration of melt/fluid. Although melt pockets were observed in peridotites from the West Qinling area, the authors interpreted those melt pockets as formed after spongy clinopyroxene (Su et al., 2010).

2. Chemical differences: Secondary clinopyroxene in spongy domain has lower Al_2O_3 and Na_2O , but higher CaO, $Cr^{\#}$ and $Mg^{\#}$, than the primary one, regardless of the interpreted origin (Figs. 6). Their plotted areas overlap with each other, and therefore cannot be used to constrain the origin of spongy clinopyroxene. Glasses are commonly observed associated with spongy-textured clinopyroxene (Bonadiman et al., 2005; Carpenter et al., 2002; Ionov et al., 2005; Shaw et al., 2006). They have different major-oxide contents which can be used to differentiate the origins of spongy texture in mantle clinopyroxene. Glasses within spongy domains studied here (decompression-induced melting of



Fig. 9. Plots (recalculated on an anhydrous basis) of SiO₂ vs (a) CaO, (b) Al_2O_3 , (c) $Na_2O + K_2O$ (d) Na_2O/K_2O , (e) MgO, (f) FeO, (g) TiO₂ and (h) Mg[#] in glasses associated with spongy clinopyroxene of various origins. Shaw and Bonadiman: glass associated with spongy-texture clinopyroxene; GL-MIs: homogenized (heated) glass inclusions from subduction zones; LD melts: low-degree melts produced by experiments on anhydrous peridotite. Data sources: Shaw are from Shaw et al. (2006) and Shaw and Dingwell (2008); Bonadiman are from Bonadiman et al. (2005); GL-MIs are from lonov et al. (2011) and Bénard et al. (2016); LD melts are from Baker et al., (1995), Draper and Green. (1997) and Robinson et al. (1998). Note K₂O contents of LD melts are generally not reported.

clinopyroxene) show narrow ranges of major-element compositions (Fig. 9). They have obviously higher Al_2O_3 (26.4–27.8 wt%) and Na_2O/K_2O (up to 26.4), compared with those related to spongy textures of other origins (Shaw and Bonadiman in Fig. 9). The Al_2O_3 contents of the studied glasses are even higher than those of low-degree melts of anhydrous peridotites and most Al-rich primitive melts percolating the lithosphere (LD melts and GL-MIs in Fig. 9). Glasses in peridotites from Sal Island (metasomatism by melt infiltration, Bonadiman in Fig. 9) contain large amounts of Na₂O and K₂O (up to 15 wt%) and have low CaO (< 3 wt%) and Na₂O/K₂O (< 1), which cannot be produced by melting of clinopyroxene and exotic alkali-rich agent must be evoked. Strong penetration of kimberlite-like melt was considered as the mechanism responsible for the formation of spongy texture (Bonadiman et al.,

2005). Glasses within spongy-textured clinopyroxene, which were formed by host magma-peridotite interaction (Shaw in Fig. 9), are characterized by high MgO and FeO, as well as $Mg^{\#}$. They also have low Na₂O/K₂O, due to the involvement of alkali-rich melts released by dissolution of orthopyroxene (Shaw et al., 2006; Shaw and Dingwell 2008).

7. Concluding remarks

1. The peridotite xenoliths from the eastern Central Asian Orogenic Belt contain spongy-textured clinopyroxene. The spongy domains are variable in thickness and consists of clinopyroxenes, glasses, vugs and traces of olivine.

Table 4

Petrologic and geochemical differences of spongy texture in mantle clinopyroxene with various origins.

Origin	Petrologic differences	Chemical differences (associated glasses)	References
Decompression-induced partial melting	No melt pocket or modal metasomatic mineral; no preferential development of spongy texture toward host magma and along cracks; orthopyroxene well preserved	High Al ₂ O ₃ and Na ₂ O/K ₂ O; low MgO, FeO and Mg [#]	This study
Host lava-peridotite interaction	Presence of reacted orthopyroxene; preferential development of spongy texture toward host magma	High MgO, FeO and Mg [#] ; low Na ₂ O/K ₂ O	Shaw et al. 2006; Shaw and Dingwell 2008
Metasomatism by melt infiltration	Presence of melt pockets, glass patches and modal metasomatic minerals	High Na ₂ O + K ₂ O; low CaO and Na ₂ O/K ₂ O	Bonadiman et al. 2005

2. Spongy texture in the studied peridotites was formed by decompression-induced melting of clinopyroxene.

3. The differences in petrologic characteristics and major-element compositions of associated glasses can be used to identify origin of spongy texture in mantle clinopyroxene.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi. org/10.1016/j.lithos.2018.08.035.

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