

Petrogenesis of plio-quaternary intra-plate continental alkaline lavas from the İskenderun Gulf (Southern Turkey): Evidence for metasomatized lithospheric mantle

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ABSTRACT

The Quaternary alkaline volcanic field of Southern Turkey is characterized by intra-continental plate-type magmatic products, exposed to the north of the İskenderun Gulf along a NE-SW trending East Anatolian Fault, to the west of its intersection with the N-S trending Dead Sea Fault zone. The İskenderun Gulf alkaline rocks are mostly silica-undersaturated with normative nepheline and olivine and are mostly classified as basanites and alkaline basalts with their low-silica contents ranging between 43 and 48 wt.% SiO₂. They display Ocean Island Basalt (OIB)-type trace element patterns characterized by enrichments in large-ion-lithophile elements (LILE) and light rare earth element (LREE), and have (La/Yb)_N = 8.8–17.7 and (Hf/Sr)_N = 0.9–1.6 similar to those of basaltic rocks found in intraplate suites. The basanitic rocks have limited variations Sr-Nd isotopic ratios (⁸⁷Sr/⁸⁶Sr = 0.70307–0.70324, ¹⁴³Nd/¹⁴⁴Nd = 0.512918–0.521947), whereas the alkali basalts display more evolved Sr-Nd isotopic ratios (⁸⁷Sr/⁸⁶Sr = 0.70346–0.70365, ¹⁴³Nd/¹⁴⁴Nd = 0.512887–0.521896). The İskenderun Gulf alkaline rocks also display limited Pb isotopic variations with ²⁰⁶Pb/²⁰⁴Pb = 18.75–19.09, ²⁰⁷Pb/²⁰⁴Pb = 15.61–15.66 and ²⁰⁸Pb/²⁰⁴Pb = 38.65–39.02, indicating that they originated from an enriched lithospheric mantle source. Calculated fractionation vectors indicate that clinopyroxene and olivine are the main fractionating mineral phases. Similarly, based on Sr-Nd isotopic ratios, the assimilation and fractional crystallization (AFC) modeling shows that the alkali basalts were affected by AFC processes (*r* = 0.2) and were slightly contaminated by the upper crustal material.

The high TiO₂ contents, enrichments in Ba and Nb, and depletions in Rb can likely be explained by the existence of amphibole in the mantle source, which might, in turn, indicate that the source mantle has been affected by metasomatic processes. The modeling based on relative abundances of trace elements suggests involvement of amphibole-bearing peridotite as the source material. İskenderun Gulf alkaline rocks can thus be interpreted as the products of variable extent of mixing between melts from both amphibole-bearing peridotite and dry peridotite.

1. Introduction

Although effusive, low-volume ocean islandbasalt-type volcanism on continent is widespread worldwide, its origin and source region are still controversial. The studies carried out in some volcanic areas with low or no continental contamination may provide information about mantle composition of OIB-type magmas. It can be said that these magmas display large compositional variations, which indicates that they are the products of melt generation from a number of compositionally different mantle reservoirs, mixing between them at various proportions and variable melting degrees (e.g. Aldanmaz et al., 2006).

The magmas of intra-continental OIB type volcanism may be derived from hydrous sub-continental lithospheric mantle (SCLM; e.g.

Hawkesworth and Gallagher, 1993) or from the asthenospheric mantle with slight contributions from SCLM (e.g. McKenzie and Bickle, 1988; White and McKenzie, 1989). When evaluating the contribution of the lithospheric mantle, care must be taken that the nature and composition of the lithospheric mantle are quite variable and sometimes cannot be easily distinguished from deep mantle sources. Lithospheric extensions, transtensional decompression or asthenospheric plumes may cause intra-continental alkaline volcanism (Turcotte and Emerman, 1983; White and McKenzie, 1989; Wilson and Downes, 1991). In such environments, the metasomatized, heterogeneous lithospheric mantle with enriched incompatible elements and low solidus temperatures is proposed to play a significant role for melt generation (Foley, 1992).

The re-fertilization of the mantle with low-degree metasomatic

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melts is a mechanism for elucidating elevated abundances of incompatible trace elements and diagnostic isotope compositions of primitive alkaline rocks. Mantle metasomatism related compositional variations can be determined from mantle xenoliths gathered from the sub-continent lithospheric mantle (e.g. Kempton et al., 1988; O'Reilly et al., 1991; Ackerman et al., 2013). According to geochemical and isotopic studies, the trace element and isotope concentrations of OIB can be reflected by mixing of various mantle components such as fertile mantle, recycled oceanic lithosphere, or recycled sub-continent lithospheric mantle (e.g. Hofmann and White, 1982; White, 1985; Hart, 1988; Hart et al., 1992). Additionally, recycled oceanic crust preserved as eclogite, pyroxenite or amphibolite can constitute a part of the lithospheric mantle (Sobolev et al., 2005, 2007). The presence of such components in the mantle and their involvement in subsequent melt production result in variable trace element and isotope compositions.

Alkaline rocks of the İskenderun Gulf, which are the product of intra-continental volcanism in southern Turkey, are exposed through the Eastern Mediterranean Region (Fig. 1). These mafic volcanic rocks in the region are alkaline basalts and basanites with OIB character (Polat et al., 1997; Parlak et al., 1997; Yurtmen et al., 2000). The origin of these continental volcanic rocks with OIB-like geochemical signatures have previously been attributed to the lithospheric thinning caused by strike-slip faults and the partial melting due to decompression of the metasomatized asthenosphere (Polat et al., 1997; Bağcı et al., 2011).

Basaltic magmas erupted in intra-plate settings are commonly subjected to low-pressure fractional crystallization at different degrees. Fractionation strongly affects the bulk-rock composition and therefore only primitive rocks can be used for interpretation of characteristics of magma sources (Aldanmaz, 2002). In this study, the geochemical characteristics and modal mineralogy of the mantle source will be determined by using whole rock major and trace elements concentrations, and Sr, Nd and Pb isotope data of İskenderun Gulf intra-continental alkaline rocks. In addition to this, the possible effects, such as fractional crystallization and crustal contamination of mantle-derived magma and the melting processes forming alkaline magma will be examined.

2. Regional geology and characteristics of volcanic rocks

The İskenderun Gulf in the Eastern Mediterranean Region is located between the African and Arabian plates, and the Anatolian collision zone (Fig. 1). Since the Early Pliocene the plate movements in the Eastern Mediterranean formed a triple joint known as "Maras Triple Junction" in Kahramanmaraş, where the Dead Sea Fault, the Eastern Anatolian Fault and the northeast extension of the Cyprus Arch intersect (Fig. 1a, b; Sengör and Yilmaz, 1981; Karig and Kozlu, 1990; Kozlu, 1987; Fyitkas et al., 1984). The movements of the Arabian and African plates towards the Anatolian plate formed the fault zones and deformations in the region (McKenzie, 1972; Sengör and Yilmaz, 1981; Fyitkas et al., 1984; İzzeldin, 1987; Bayer et al., 1988; Le Pichon and Gaulier, 1988; Yilmaz et al., 1988; Yürür and Chorowicz, 1998; Albora et al., 2006).

The İskenderun Gulf area is located between the Misis Range in the north and the Amanos Range in the east (Fig. 1a). The Misis Range is an olistostromic complex tectonically emplaced onto Miocene to Early Pliocene marine sediments (Kelling et al., 1987). The Amanos Range in the east is the fragment of the continental crust, which is separated from the Arabian Platform by the NE-SW left-lateral Karatas-Osmaniye Fault Zone (KFZ) and usually included in the African plate (Adiyaman and Chorowicz, 2002).

The Plio-Quaternary intra-continental volcanic rocks in the İskenderun Gulf area (Arger et al., 2000), outcrop along the northeast-southwest KFZ (Fig. 1a; KOFZ; Saroğlu et al., 1992). Deli Halil Hill, Üçtepeler, Toprakkale, Gertepe, and some smaller morphologically preserved cinder cones represent eruptive centers of the studied basaltic lavas. Dali Halil Hill represents a small, but rather complex volcano,

with a central cone, several parasitic craters and lava flows. The cone is built of basaltic scoria (Yurtmen et al., 2000). Olivine phenocrysts and ellipsoidal vesicles are apparent in basaltic lava with blocky surface (aa-type). Columnar jointing with thin columns (40–50 cm) can be seen where inner section of the lava is exposed. The cones consist of alternating layers of black and red (oxidized) scoriae with clast-supported structure, suggesting fall deposition. The scoriae contain lava shreds and bombs (up to 30 cm). The character of deposits implies a Strombolian style eruption (Walker, 1973; Cas and Wright, 1987). The scoria cones of Gertepe and Üçtepeler located to the north of Deli Halil Hill are elongated approximately E-W. The lava flows are intercalated with alluvial fan deposits of Quaternary age (Yurtmen et al., 2000). The ages of the volcanic rocks containing alkali olivine basalt and basanitic rocks were determined to be younger than 2.25 Ma according to bulk-rock K-Ar ages (Arger et al., 2000).

3. Petrography

Petrography of the alkali basalts and basanites from the İskenderun Gulf was described by Yurtmen et al. (2000) and Bağcı et al. (2011). The majority of alkali basalts have intergranular, interstitial texture and some of them have a vesicular structure. The phenocryst size is about 1.5 mm and proportion of phenocrysts to matrix varies between 15 and 25%. Olivines are euhedral and subhedral, mostly not altered, but occasionally iddingsitized along margins. Skeletal and embayment textures are also found in some olivine phenocrysts. Clinopyroxene in the form of microphenocrysts is subhedral to euhedral. Plagioclase, olivine, clinopyroxene and opaque minerals constitute the matrix.

Basanites display interstitial, pilotaxitic and weakly porphyric textures, with common euhedral and subhedral olivine phenocrysts. The phenocryst size is about 2 mm and the proportion of phenocrysts to matrix is < 10%. There are also melt inclusions in the olivines, where skeleton and embayment textures are observed. Some olivines are fractured and iddingsitized along the crystal margins and fractures. The matrix is composed of plagioclase, olivine, clinopyroxene, nepheline and opaque minerals. The opaque minerals in these series were defined as titanomagnetite by Yurtmen et al. (2000).

4. Analytical techniques

Twenty-five alkaline volcanic rocks from the İskenderun Gulf were sampled for major and trace-element analyses. Major and trace element concentrations were determined by inductively coupled plasma emission spectrometry (ICP-AES) and inductively coupled plasma mass spectrometry (ICP-MS) at Analytical Laboratories (ACME, Canada). Analyses of major and trace elements were made with the 4 A + 4B package method. 0.2 g of rock-powder was fused with 1.5 g LiBO₂ and then dissolved in 100 ml 5% HNO₃. The loss on ignition (LOI) values were obtained by heating the dried samples at 1000 °C. The detection limits range from 0.01 to 0.1 wt.%, for major oxides, 0.1 to 10 ppm for trace elements and 0.01 to 0.5 ppm for the rare earth elements. The major and trace element contents and the sample locations are given in Table 1.

For Sr-Nd isotope analysis, 9 samples were selected from alkali basalt and basanitic rocks based on geochemical characteristics. Sr-Nd isotopic analyses were performed at the Middle East Technical University (Radiogenic Isotope Laboratory) Ankara, Turkey, using analytical techniques described in Romer et al. (2001). Isotope ratios were measured using TIMS (Triton Multi-Collector Thermal Ionization Mass Spectrometer) and analytical uncertainties were given at 2 sigma levels. Ratios of ⁸⁷Sr/⁸⁶Sr are normalized to ⁸⁶Sr/⁸⁸Sr = 0.1194 (Nier, 1938), and ¹⁴³Nd/¹⁴⁴Nd data are normalized to ¹⁴³Nd/¹⁴⁴Nd = 0.7219 (O'Nions et al., 1979). During the course of the measurement, NBS 987 and Nd LaJolla international standards were measured with ⁸⁷Sr/⁸⁶Sr and ¹⁴³Nd/¹⁴⁴Nd of 0.710244 ± 8 (n = 3) and 0.511852 ± 2 (n = 2), respectively. The USGS reference material BCR-1 gave ⁸⁷Sr/⁸⁶Sr = 0.705014 ± 5 and ¹⁴³Nd/¹⁴⁴Nd = 0.512638 ± 4.

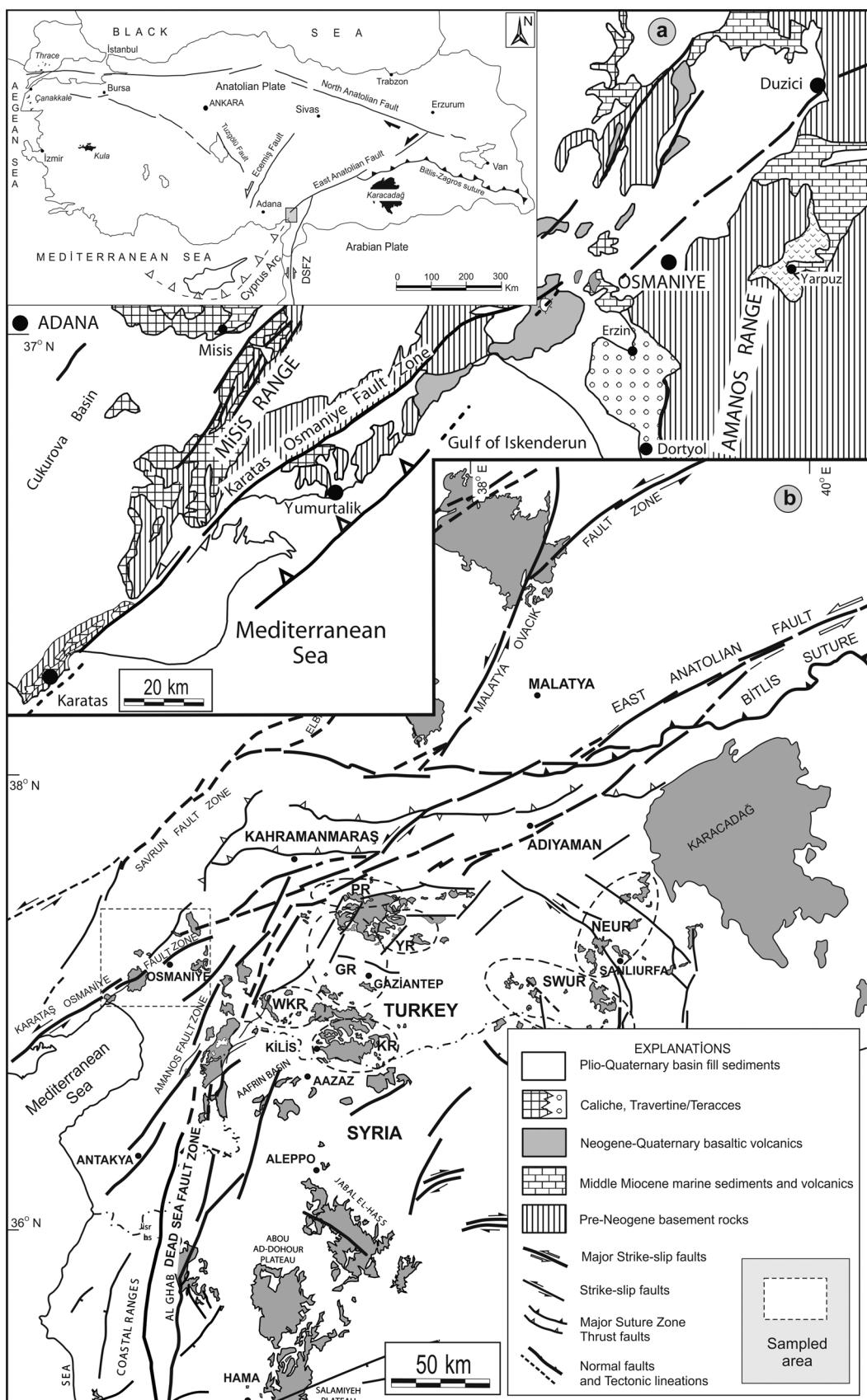


Fig. 1. (a) Geological map of the İskenderun Gulf and its surroundings showing major faults and volcanic outcrops, modified from Gürsoy et al., 2003 (based on the 1:100,000 geological map of Turkey). Inset figure show the alkaline volcanic suits (black areas) of Kula, Çanakkale, Thrace and Karacadağ, respectively. (b) Simplified map of southern Turkey and northwestern Syria highlighting the distribution of Cenozoic volcanic rocks (Gürsoy et al., 2009; Ma et al., 2013). Abbreviations: GR: Gaziantep Region; KR: Kilis Region; PR: Pazarcık Region; YR: Yavuzeli Region; WKR: Western Kilis Region; SWUR: SW Urfa Region; NEUR: NE Urfa Region.

Table 1

Whole-rock major and trace element data for the representative samples from İskenderun Gulf alkaline volcanic rocks.

Sample No	O-1	O-2	O-3	O-4	O-5	O-6	O-7	O-9	O-10	O-11	O-12	O-14	
Rock Type	Basanite	Basanite	Basanite	Basanite	Al. Bas.	Basanite	Al. Bas.	Al. Bas.	Al. Bas.	Al. Bas.	Al. Bas.	Al. Bas.	
Longitude	36° 04'	36° 04'	36° 05'	36° 04'	36° 05'	36° 08'	36° 08'	36° 08'	36° 07'	36° 07'	36° 08'	36° 03'	
Latitude	16° E	15° E	14° E	16° E	06° E	09° E	08° E	24° E	55° E	59° E	01° E	19° E	
	37° 02'	37° 02'	37° 02'	37° 02'	37° 03'	37° 03'	37° 02'	37° 01'	37° 01'	37° 01'	37° 01'	37° 06'	
	09° N	10° N	15° N	10° N	06° N	04° N	28° N	37° N	41° N	53° N	56° N	32° N	
												MDL	
SiO ₂	43.22	43.38	46.13	46.34	48.02	43.67	47.32	47.10	45.48	47.18	46.67	45.34	
TiO ₂	2.90	2.90	2.52	2.54	1.94	2.89	1.96	2.06	1.84	2.02	2.07	2.47	
Al ₂ O ₃	14.38	14.28	15.99	16.01	15.15	14.64	14.95	14.75	14.38	14.78	14.72	15.26	
Fe ₂ O ₃	13.27	13.17	11.15	11.38	12.46	13.18	12.77	12.98	12.45	12.80	12.78	12.17	
MnO	0.18	0.18	0.18	0.19	0.16	0.18	0.15	0.17	0.17	0.17	0.17	0.16	
MgO	9.65	9.49	7.17	7.30	7.92	9.29	7.42	8.58	8.80	8.56	8.72	8.79	
CaO	9.46	9.29	7.58	7.64	9.70	10.26	9.72	9.79	11.14	9.81	9.79	10.12	
Na ₂ O	4.18	3.71	5.19	4.98	3.19	3.59	2.96	3.36	3.03	3.27	3.08	3.26	
K ₂ O	1.63	1.91	2.72	2.45	0.73	1.42	0.68	0.89	0.73	0.82	0.87	1.12	
P ₂ O ₅	0.98	1.04	0.71	0.72	0.33	0.86	0.32	0.38	0.32	0.36	0.40	0.42	
L.O.I.	0.3	0.2	0.3	0.1	0.1	0.4	1.4	0.4	1.4	0.1	0.4	0.6	
Total	99.61	99.62	99.67	99.68	99.76	99.65	99.73	99.74	99.75	99.74	99.73	99.71	
Sc	21	21	18	19	23	24	23	24	24	23	26	1	
Cr	246.3	246.3	109.5	116.3	266.9	239.5	266.9	287.3	307.9	287.3	287.3	205.2	
V	208	203	163	166	193	235	203	204	195	203	203	8	
Ni	195	189	114	117	112	164	120	152	151	146	151	20	
Co	50.7	48.1	37.4	35.5	44.6	48.9	44.3	47.7	49.1	48.5	47.4	50.1	
Cu	58.5	64.8	37.7	33.7	37.7	52.9	43.5	43.6	40.1	41.4	31.2	43.3	
Zn	90	93	65	55	62	70	73	75	70	69	56	60	
Ga	19.8	20.3	20.2	19.6	18.5	19.6	19.8	19.5	18.6	19.5	17.9	18.9	
Rb	18.7	19.5	31.4	33.1	6.8	13.4	4.2	9.1	7.2	7.8	8.7	9.1	
Sr	1046	1054	845.8	817.3	484.6	953.7	797.8	551.7	498.1	518.7	531.1	670.4	
Y	22.3	23.0	24.7	24.3	18.2	22.2	18.3	18.4	16.1	17.7	17.7	19.5	
Zr	219.5	221.9	324.5	315.5	107.1	169.4	103.5	117.9	102.2	113.2	119.8	138.2	
Nb	58.0	58.1	60.9	60.7	16.0	42.2	15.6	19.8	16.1	18.9	20.2	24.6	
Ba	358	355	346	324	166	309	171	195	153	169	196	201	
La	40.3	42.6	44.3	43.8	19.1	34.7	18.9	20.8	17.4	19.1	20.7	21.8	
Ce	84.8	86.0	83.6	83.3	37.3	69.5	36.1	42.4	34.6	39.4	41.8	44.8	
Pr	10.10	10.33	9.44	9.27	4.53	8.67	4.42	4.99	4.14	4.69	4.92	5.33	
Nd	40.5	43.5	37.1	33.5	18.0	35.8	19.6	20.8	17.0	19.7	21.5	22.3	
Sm	7.70	8.04	6.78	6.58	4.00	6.88	4.10	4.53	3.69	4.24	4.42	4.90	
Eu	2.55	2.61	2.17	2.17	1.45	2.32	1.48	1.58	1.36	1.56	1.67	0.02	
Gd	6.71	6.93	5.90	6.12	4.25	6.11	4.39	4.47	4.14	4.53	4.47	4.63	
Tb	0.97	0.99	0.90	0.93	0.69	0.95	0.68	0.72	0.64	0.69	0.71	0.75	
Dy	4.91	5.17	5.03	4.93	3.70	4.97	3.86	3.95	3.47	3.81	3.98	3.97	
Ho	0.82	0.85	0.94	0.92	0.69	0.89	0.70	0.70	0.63	0.68	0.69	0.74	
Er	2.13	2.21	2.57	2.48	1.76	2.19	1.85	1.91	1.76	1.86	1.79	2.03	
Tm	0.27	0.29	0.35	0.34	0.24	0.30	0.26	0.25	0.23	0.24	0.26	0.27	
Yb	1.63	1.70	2.35	2.21	1.54	1.74	1.58	1.60	1.41	1.47	1.56	1.62	
Lu	0.24	0.24	0.35	0.32	0.23	0.25	0.22	0.23	0.20	0.23	0.22	0.01	
Hf	5.5	5.5	7.4	6.8	2.8	4.2	3.4	2.9	3.0	3.2	3.4	0.1	
Ta	3.4	3.8	4.1	4.4	1.0	2.5	1.0	1.2	1.0	1.1	1.1	0.1	
Pb	2.3	2.9	1.2	1.2	1.2	1.3	1.7	1.5	1.3	1.1	1.3	0.1	
Th	3.9	4.2	7.6	8.1	2.6	3.0	2.0	2.3	2.2	2.4	2.2	0.2	
U	1.2	1.4	2.2	2.3	0.6	0.9	0.2	0.6	0.4	0.6	0.6	0.1	
Mg#	61.6	61.3	58.6	58.8	58.3	60.0	56.1	59.3	60.9	59.6	60.0	61.4	
Fractionation corrected values													
Rb	14.80	15.43	22.97	23.89	4.91	10.34	2.82	6.75	5.55	5.86	6.62	7.20	
Sr	884.2	850.7	635.3	606.3	359.5	752.0	554.3	419.6	392.7	399.3	413.8	541.1	
K	10693	12530	16491	14650	4365	9080	3784	5469	4668	5107	5491	7347	
La	32.1	33.9	32.7	31.9	13.9	26.9	12.8	15.5	13.5	14.4	15.8	17.3	
Nb	45.8	45.9	44.4	43.7	11.5	32.5	10.4	14.6	12.4	14.1	15.3	19.4	
Dy	4.11	4.33	3.97	3.85	2.89	4.09	2.86	3.15	2.85	3.07	3.24	3.33	
Yb	1.35	1.41	1.83	1.71	1.19	1.41	1.15	1.26	1.14	1.17	1.25	1.34	
Th	3.1	3.3	5.6	5.8	1.8	2.3	1.3	1.7	1.8	1.6	1.5		
Sample No	O-16	O-17	O-18	O-19	O-21	O-22	O-23	O-24	O-26	O-27	O-28	O-30	O-31
Rock Type	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Al. Bas.	Basanite	Al. Bas.	Basanite
Longitude	36° 04'	36° 04'	36° 04'	36° 04'	36° 05'	36° 05'	36° 05'	36° 05'	36° 05'	36° 59'	36° 09'	36° 08'	36° 11'
Latitude	26° E	47° E	44° E	55° E	17° E	29° E	36° E	30° E	35° E	59° E	20° E	00° E	06° E
	37° 06'	37° 06'	37° 06'	37° 06'	37° 07'	37° 06'	37° 07'	37° 02'	37° 02'	36° 05'	37° 04'	37° 01'	37° 06'
	52° N	58° N	56° N	50° N	01° N	55° N	15° N	14° N	20° N	42° N	10° N	47° N	30° N
SiO ₂	44.62	44.76	45.82	45.00	44.43	45.32	44.39	45.30	45.04	47.68	44.18	47.22	43.9
TiO ₂	2.86	2.79	2.62	2.80	2.80	2.73	2.49	2.70	2.76	1.84	3.46	2.03	2.29
Al ₂ O ₃	14.93	14.88	15.66	14.77	14.88	15.73	15.31	15.38	15.04	14.66	14.83	14.78	13.62
Fe ₂ O ₃	11.37	11.74	11.13	11.66	11.90	11.20	10.72	11.49	11.51	12.61	12.67	12.84	12.79
MnO	0.17	0.17	0.17	0.17	0.17	0.17	0.16	0.17	0.17	0.17	0.17	0.17	0.17

(continued on next page)

Table 1 (continued)

Sample No	O-16	O-17	O-18	O-19	O-21	O-22	O-23	O-24	O-26	O-27	O-28	O-30	O-31
Rock Type	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Basanite	Al. Bas.	Basanite	Al. Bas.	Al. Bas.	Basanite
Longitude	36° 04'	36° 04'	36° 04'	36° 04'	36° 05'	36° 05'	36° 05'	36° 05'	36° 05'	36° 05'	36° 05'	36° 08'	36° 11'
Latitude	26° E	47° E	44° E	55° E	17° E	29° E	36° E	30° E	35° E	59° E	20° E	00° E	06° E
	37° 06'	37° 06'	37° 06'	37° 06'	37° 07'	37° 06'	37° 07'	37° 02'	37° 02'	36° 05'	37° 04'	37° 01'	37° 06'
	52° N	58° N	56° N	50° N	01° N	55° N	15° N	14° N	20° N	42° N	10° N	47° N	30° N
MgO	9.02	9.31	7.67	9.09	9.38	8.08	8.03	8.62	9.04	8.56	6.26	8.07	12.28
CaO	8.50	8.51	7.31	8.57	8.67	7.83	7.38	8.03	8.50	9.73	11.42	9.82	8.57
Na ₂ O	4.57	4.49	5.05	4.49	4.40	4.97	4.64	4.84	4.34	3.29	3.47	3.16	3.69
K ₂ O	2.38	2.15	2.77	2.27	2.24	2.55	2.47	2.50	2.11	0.70	1.51	0.81	1.51
P ₂ O ₅	0.73	0.76	0.72	0.78	0.78	0.73	0.63	0.74	0.76	0.41	0.89	0.36	0.86
L.O.I.	0.4	0.0	0.7	0.0	0.1	0.3	3.4	0.2	0.3	0.0	0.8	0.4	0.2
Total	99.64	99.64	99.66	99.64	99.63	99.66	99.67	99.65	99.65	99.72	99.67	99.73	99.60
Sc	21	21	18	21	19	18	20	21	22	25	23	20	
Cr	212.1	225.8	136.9	232.6	225.8	164.2	198.41	184.7	218.9	342.1	102.6	287.3	335.2
V	236	231	187	222	217	198	174	202	204	200	263	216	192
Ni	160	177	130	178	184	138	148	159	175	176	54	139	322
Co	47.6	50.2	42.8	48.3	48.6	42.3	47.8	46.5	46.8	53.2	42.6	52.6	62.9
Cu	39.3	41.1	32.6	45.1	44.4	37.1	100.9	36.9	41.9	39.0	46.6	42.2	46.2
Zn	59	60	59	60	59	56	57	61	59	85	71	72	68
Ga	21.2	21.5	22.7	19.3	20.0	20.3	17.9	21.1	18.4	18.8	20.6	19.4	17.8
Rb	24.6	24.3	29.0	22.8	22.8	27.7	26.0	26.0	23.0	9.5	13.8	7.6	16.8
Sr	917.9	910.8	881.1	945.6	938.7	888.3	802.9	888.6	894.9	602.8	1060	550.4	907.3
Y	23.7	24.0	24.7	24.3	23.5	24.2	21.5	24.1	23.0	19.6	24.1	19.7	20.8
Zr	274.1	271.9	327.9	266.5	264.3	306.3	291.8	298.3	258.2	130.3	194.5	123.1	202.8
Nb	55.8	52.8	60.4	53.4	51.8	56.1	52.3	54.9	49.4	22.7	55.5	19.6	44.5
Ba	325	319	317	324	313	313	271	309	300	250	407	283	316
La	41.4	40.6	39.8	39.3	38.9	40.8	34.6	38.9	37.5	32.3	34.7	19.9	39.0
Ce	79.7	78.0	81.0	80.7	80.5	80.5	71.7	78.7	77.1	63.4	80.6	43.3	80.9
Pr	9.54	9.29	9.21	9.32	9.26	9.17	7.73	8.96	8.76	6.81	9.54	4.76	8.62
Nd	39.5	37.6	35.9	38.4	36.1	36.6	29.5	35.5	34.8	25.2	41.6	20.8	34.7
Sm	6.99	7.02	6.86	7.02	7.10	6.68	5.54	6.62	6.66	4.87	7.82	4.45	6.43
Eu	2.25	2.30	2.20	2.21	2.19	2.21	1.90	2.22	2.24	1.71	2.68	1.58	2.20
Gd	6.06	6.37	5.96	6.18	6.21	6.20	5.31	6.20	6.17	4.68	7.31	4.57	5.74
Tb	0.92	0.95	0.93	0.93	0.91	0.91	0.79	0.91	0.91	0.74	1.02	0.72	0.84
Dy	4.77	4.96	4.81	4.75	4.82	4.74	4.03	4.67	4.71	3.76	4.81	3.67	4.17
Ho	0.87	0.86	0.85	0.85	0.87	0.87	0.79	0.88	0.84	0.73	0.95	0.71	0.78
Er	2.26	2.38	2.51	2.39	2.29	2.36	2.03	2.37	2.26	1.92	2.27	1.83	1.98
Tm	0.33	0.34	0.35	0.32	0.33	0.35	0.32	0.33	0.32	0.27	0.29	0.26	0.28
Yb	1.94	1.96	2.13	1.87	1.89	2.18	1.83	2.17	1.93	1.66	1.85	1.62	1.65
Lu	0.29	0.29	0.31	0.28	0.28	0.30	0.26	0.29	0.27	0.24	0.25	0.22	0.22
Hf	5.6	5.9	7.0	6.0	6.0	6.2	6.3	7.1	5.8	3.2	4.9	2.8	4.1
Ta	3.6	3.5	3.9	3.1	3.3	3.5	3.3	3.5	3.3	1.4	3.0	1.0	2.6
Pb	2.2	1.0	2.4	1.8	1.0	1.7	1.8	1.1	1.0	0.8	1.4	1.6	1.1
Th	5.4	4.9	5.1	4.4	5.0	5.8	5.1	5.5	4.5	5.2	2.9	2.1	4.4
U	1.6	1.4	1.7	1.5	1.6	1.8	1.8	1.8	1.3	1.1	0.8	0.5	1.1
Mg#	63.6	63.6	60.3	63.2	63.4	61.3	62.2	62.3	63.3	59.9	52.1	58.0	67.9
Fractionation corrected values													
Rb	20.44	20.20	24.08	18.72	18.95	21.64	21.09	21.09	19.12	7.23	8.31	5.41	15.13
Sr	774.8	768.9	686.5	789.5	792.4	708.7	662.9	733.7	755.4	469.7	666.2	403.1	824.4
K	16403	14818	17483	15456	15438	16517	16614	16815	14542	4418	7526	4776	11283
La	34.6	33.9	30.5	32.4	32.5	32.1	28.2	31.7	31.3	24.8	21.2	14.3	35.2
Nb	46.3	43.8	45.9	43.8	43.0	43.7	42.3	44.4	41.0	17.2	33.3	13.9	40.0
Dy	4.15	4.31	3.92	4.09	4.19	3.93	3.44	3.99	4.09	3.06	3.28	2.84	3.85
Yb	1.67	1.64	1.71	1.60	1.63	1.79	1.55	1.84	1.66	1.33	1.24	1.24	1.51
Th	4.5	4.0	3.9	3.6	4.1	4.5	4.1	4.4	3.7	3.9	3.7	1.5	3.9

Pb isotope ratios were performed at the Laboratoire G-Time of the Université Libre de Bruxelles (Belgium). To remove alteration and any potential contamination, a leaching procedure was applied to powdered whole rock samples (250 mg), with repeated additions (5–6 times) of 6 N sub-boiled HCl followed by 30-minutes ultrasonic baths until the solution was clear. Lead separation was completed using successive acid elution on anionic resin column (e.g. Weis et al., 2006). Analytical uncertainties are given at 2-sigma level. Analyses of the SRM981 Pb standard solution yielded $^{208}\text{Pb}/^{204}\text{Pb} = 36.7174 \pm 47$, $^{207}\text{Pb}/^{204}\text{Pb} = 15.4978 \pm 13$, and $^{206}\text{Pb}/^{204}\text{Pb} = 16.9408 \pm 20$.

5. Results

5.1. Major and trace element characteristics

The mafic extrusive rocks are classified using the total alkali ($\text{Na}_2\text{O} + \text{K}_2\text{O}$) versus SiO_2 (TAS) in the diagram of Le Bas et al. (1986). According to this diagram, the rocks are all alkaline and classified as basanite and alkali basalt (Fig. 2). The rocks are silica-undersaturated, sodic ($\text{Na}_2\text{O}/\text{K}_2\text{O} = 1.82\text{--}4.70$) and have SiO_2 concentrations between 43 and 48%. The alkali basalts have moderate MgO (7–8 wt.%),

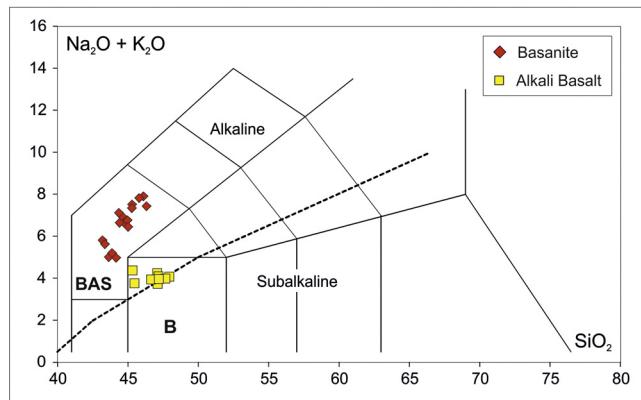


Fig. 2. Classification of the volcanic rocks from the İskenderun Gulf on a TAS diagram of Le Bas et al. (1986). Key to abbreviations: B: basalt; BAS: basanite.

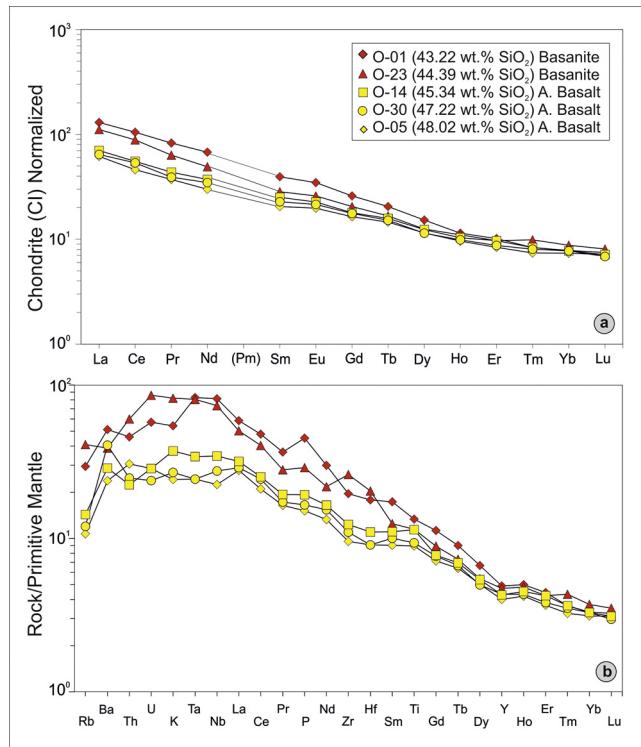


Fig. 3. (a) Chondrite-normalized REE element patterns for the İskenderun Gulf volcanic rocks. Chondrite normalising values are from Boynton (1984). (b) Primitive mantle normalized multi-element patterns for the Gulf of İskenderun volcanic rocks. Primitive mantle normalising values are from Sun and McDonough (1989).

whereas the basanites have a wide range of MgO contents (6–12 wt.%). Mg-number (Mg-number = 100Mg/(Mg/Fe)) varies from 58 to 60 for the alkali basalt samples and from 52 to 67 for the basanite samples. The rocks have high Al₂O₃ (14–16 wt.%) contents (Table 1). The CaO contents of the alkali basalt samples (9–11 wt.%) are generally higher than the basanite samples (7–11 wt.%). The Cr and Ni concentrations of the alkali basalts vary from 205 to 342 and from 112 to 176, respectively, while those of the basanite samples are between 102 and 335 ppm, and 54 and 322 ppm. In the chondrite-normalized (Boynton, 1984) REE diagram, the samples form a regular and semi-parallel patterns and there is common enrichment in LREE relative to HREE (Fig. 3a). The basanite samples are characterized by steeper REE patterns compared with alkaline basalts. The alkali basalts have (La/Yb)_N ratios from 8 to 13 while (La/Yb)_N = 13–17 in basanites. Primitive mantle normalized (Sun and McDonough, 1989) trace element diagrams for the İskenderun Gulf alkaline rocks is shown in Fig. 3b. On the primitive mantle-normalized multi-element diagrams, the basanite samples show positive Nb-Ta anomalies, with incompatible trace-element abundances higher than the alkali basalt samples.

5.2. Sr-Nd-Pb-isotopic systematics

The measured Sr, Nd and Pb isotopic ratios of volcanic rocks are given in Table 2. The Sr-Nd isotope ratios of alkaline basalts are more radiogenic than those of basanites. The ⁸⁷Sr/⁸⁶Sr isotope ratios of alkali basalts and basanites are 0.70346–0.70365 and 0.70307–0.70324, respectively. The ¹⁴³Nd/¹⁴⁴Nd isotope ratios are 0.512887–0.512896 in alkali basalts and 0.512918–0.512947 in basanites. For comparison, the diagrams include also the fields for MORB, OIB, CiMACI (the Circum-Mediterranean Anorogenic Cenozoic Igneous Province; Lustrino and Wilson, 2007) and several intra-plata lava suites, using data from a wider region (Fig. 4). In the ¹⁴³Nd/¹⁴⁴Nd vs ⁸⁷Sr/⁸⁶Sr correlation plot, İskenderun Gulf alkaline lavas in OIB area are similar to NW Anatolian alkaline lavas, Karacadag lavas and some NW Syria samples and basanites showing depleted-like pattern and alkali basalts being more enriched-like. The samples plot in the PREMA quadrangle relative to Bulk Silicate Earth and show a clear negative correlation (Fig. 4a).

The Pb isotope compositions of the samples are: ²⁰⁶Pb/²⁰⁴Pb = 19.02–19.08, ²⁰⁷Pb/²⁰⁴Pb = 15.64–15.66, ²⁰⁸Pb/²⁰⁴Pb = 38.98–39.02 for alkali basalts and ²⁰⁶Pb/²⁰⁴Pb = 18.75–19.09, ²⁰⁷Pb/²⁰⁴Pb = 15.60–15.63, ²⁰⁸Pb/²⁰⁴Pb = 38.65–38.88 for basanites. In the ¹⁴³Nd/¹⁴⁴Nd vs ²⁰⁶Pb/²⁰⁴Pb plot, all the samples fall in the OIB field and cluster between the Bulk Silicate Earth (BSE) and MORB fields (Fig. 4b). In the ²⁰⁷Pb/²⁰⁴Pb – ²⁰⁶Pb / ²⁰⁴Pb plot, the rocks fall on the right of geochron above the NHRL line ($\Delta^{7/4}$ up to 11.20; $\Delta^{8/4}$ up to 38.40; Table 2, Fig. 4c). There is a slight trend towards EMII in the alkali basalt samples. Similarly, in the ²⁰⁸Pb/²⁰⁴Pb – ²⁰⁶Pb / ²⁰⁴Pb diagram, the rocks are coherent to the above-mentioned OIB-type rocks, while alkali basalts form a trend towards the EMII composition (Fig. 4d). The alkali basalts are shifted towards a HIMU component in ¹⁴³Nd/¹⁴⁴Nd – ²⁰⁸Pb* / ²⁰⁶Pb* diagram (Fig. 4e).

Table 2

Sr, Nd and Pb isotope compositions for the İskenderun Gulf alkaline volcanic rocks.

Sample	⁸⁷ Sr/ ⁸⁶ Sr	2σ	¹⁴³ Nd/ ¹⁴⁴ Nd	2σ	²⁰⁶ Pb/ ²⁰⁴ Pb	2σ	²⁰⁷ Pb/ ²⁰⁴ Pb	2σ	²⁰⁸ Pb/ ²⁰⁴ Pb	2σ	$\Delta^{7/4}$	$\Delta^{8/4}$
O-2	0.703148	0.000012	0.512934	0.000005	18.7552	0.0017	15.6361	0.0015	38.6529	0.0044	11.21	35.09
O-3	0.703224	0.000012	0.512946	0.000007	19.0437	0.0014	15.6189	0.0012	38.8809	0.0035	6.35	23.00
O-16	0.703154	0.000015	0.512926	0.000003	18.9080	0.0037	15.6121	0.0030	38.7189	0.0085	7.14	23.21
O-21	0.703243	0.000018	0.512918	0.000002	19.0981	0.0031	15.6122	0.0026	38.8830	0.0062	5.10	16.63
O-24	0.703076	0.000009	0.512947	0.000003	19.0859	0.0016	15.6028	0.0013	38.8481	0.0038	4.28	14.63
O-9	0.703561	0.000017	0.512887	0.000003	19.0456	0.0049	15.6620	0.0044	39.0208	0.0106	10.64	36.77
O-12	0.703577	0.000006	0.512884	0.000001	19.0454	0.0016	15.6608	0.0011	39.0238	0.0033	10.53	37.10
O-14	0.703460	0.000009	0.512896	0.000002	19.0884	0.0054	15.6405	0.0043	38.9866	0.0103	8.03	28.17
O-30	0.703659	0.000015	0.512872	0.000002	19.0268	0.0057	15.6656	0.0039	39.0145	0.0109	11.21	38.40

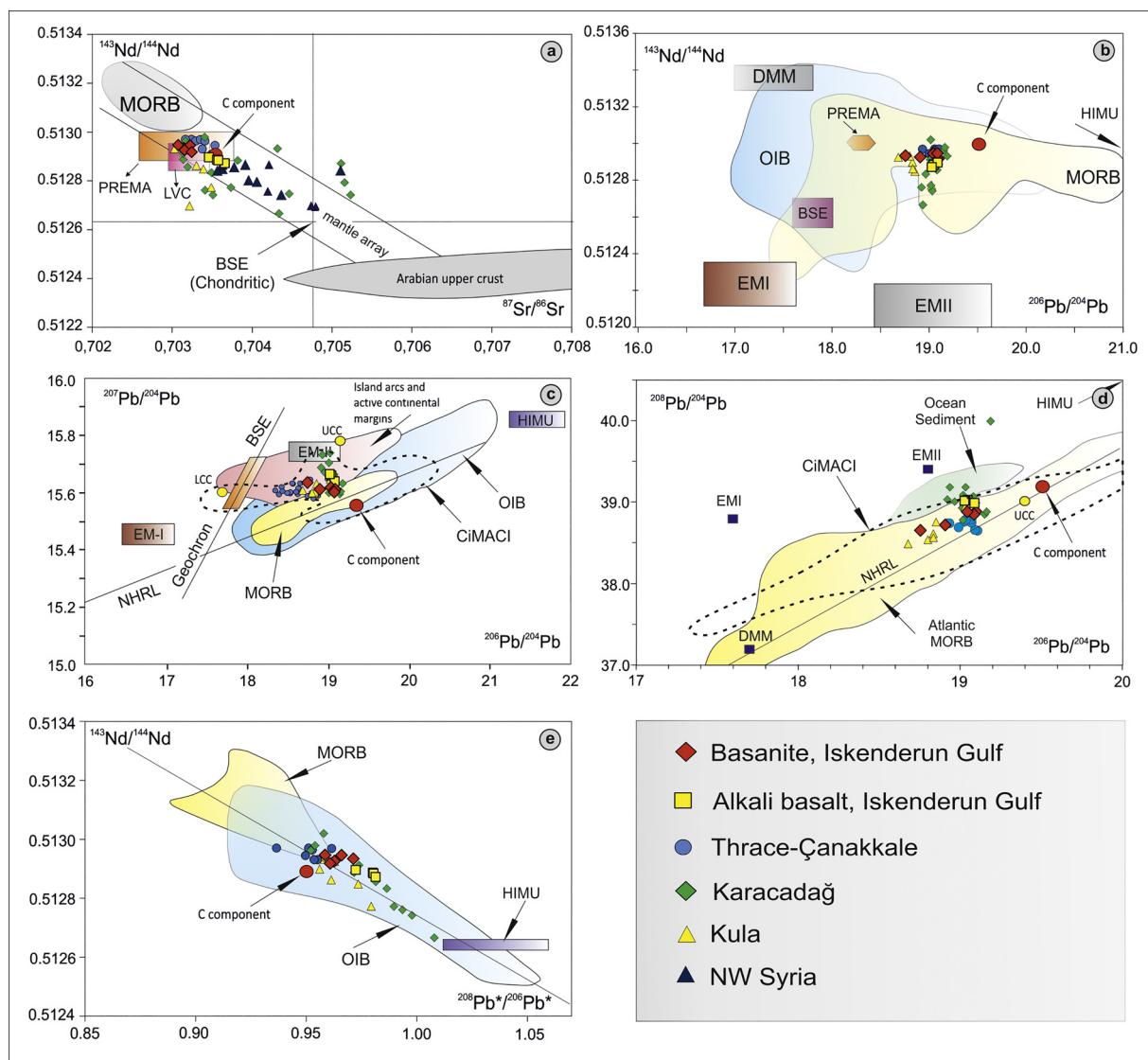


Fig. 4. (a) $^{87}\text{Sr}/^{86}\text{Sr}$ vs. $^{143}\text{Nd}/^{144}\text{Nd}$ and (b) $^{143}\text{Nd}/^{144}\text{Nd}$ vs. $^{206}\text{Pb}/^{204}\text{Pb}$ and (c) $^{207}\text{Pb}/^{204}\text{Pb}$ vs. $^{206}\text{Pb}/^{204}\text{Pb}$; and (d) $^{208}\text{Pb}/^{204}\text{Pb}$ vs. $^{206}\text{Pb}/^{204}\text{Pb}$; and (e) $^{143}\text{Nd}/^{144}\text{Nd}$ vs. $^{208}\text{Pb}^*/^{206}\text{Pb}^*$ diagrams for the alkaline lavas from Iskenderun Gulf. Data of the intra-continental volcanic rocks for a comparison is from Thrace and Çanakkale (Aldanmaz et al., 2006), Karacadağ (Keskin et al., 2012; Asan and Kurt, 2011; Şen et al., 2004), Kula (Aldanmaz et al., 2015; Alici et al., 2002; Innocenti et al., 2005), and NW Syria (Ma et al., 2013). Also plotted for comparison are: NHRL (Northern hemisphere reference line; Hart, 1984); common mantle component and the average composition of the upper continental crust (C, UCC; Hofmann, 1997); oceanic sediments (Ben Othman et al., 1989); Arabian upper continental crust (Hegner and Pallister, 1989; Jarrar et al., 2003); the Circum-Mediterranean Anorogenic Cenozoic Igneous Province (CiMACI; Lustrino and Wilson, 2007) and Atlantic MORB (Agranić et al., 2005; Hoernle et al., 2011). The parameter $^{208}\text{Pb}^*/^{206}\text{Pb}^*$ reflects radiogenic addition to the terrestrial lead defined as $[(^{208}\text{Pb}/^{204}\text{Pb}) \text{ measured} - 29.476]/[(^{206}\text{Pb}/^{204}\text{Pb}) \text{ measured} - 9.307]$ (Galer and O'Nions, 1985).

6. Discussion

The near primitive basaltic rocks that do not exhibit crustal contamination effect can accurately decipher the mantle composition, and in studies using such rocks the characteristics of the mantle and the processes of melting can be determined (e.g. Aldanmaz et al., 2015). In the mantle source studies using basaltic rocks, it is important to investigate the effects of shallow depth processes such as magma differentiation, contamination and mixing which are considered to change mantle trace element and isotopic signatures. For this reason, it is necessary to define whether intra-continental alkaline rocks of the Iskenderun Gulf are primitive in composition and to determine the possible effects that change the composition of the primary melts.

6.1. Primary melt composition and fractional crystallization

Basaltic magmas occurring in intra-plate environments are commonly subjected to fractional crystallization at different degrees. The magmas, in this case, may differ from primitive compositions. Magmas with primitive compositions should have high concentrations of compatible trace elements (e.g. Ni, Cr, V, Sc), Mg# in the range 68–76 (e.g. Frey et al., 1978) or the bulk rock Fe/Mg ratios within a range that is in equilibrium with mantle olivine (e.g. Fo91; Albaréde, 1992). Ni and Cr (alkali basalts: 112–176 ppm; basanites: 54–322 ppm and alkali basalts: 205–342 ppm; basanites: 102–335 ppm, respectively) contents are lower than expected, and Mg# is in the range of 58–60 for alkali basalts and 52–67 for basanites. Therefore, these rocks do not appear to reflect

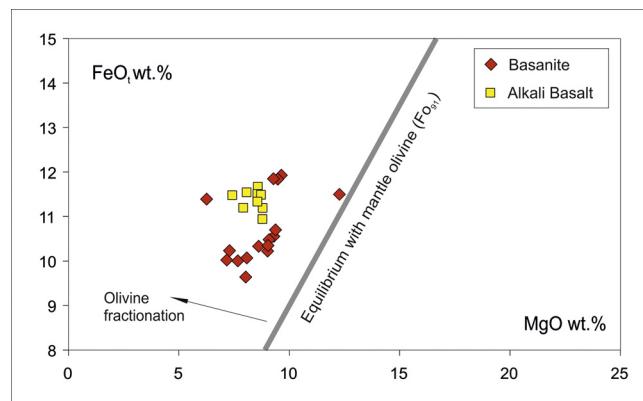


Fig. 5. Diagram of FeO_t versus MgO showing the effects of olivine fractionation on primary melt compositions. Supposing a K_D of 0.3, an $\text{Fe}_2\text{O}_3/\text{MgO}$ ratio of 0.9 can be inferred for the melt that is in equilibrium with the mantle olivine (Fo_{91}), and the Fe_2O_3 and MgO concentrations of liquid can be estimated.

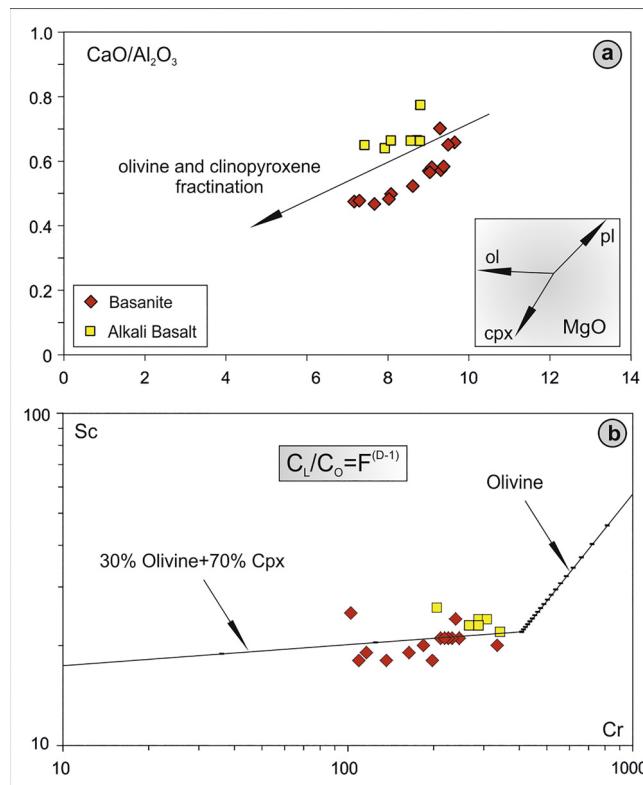


Fig. 6. (a) MgO vs $\text{CaO}/\text{Al}_2\text{O}_3$ diagram for volcanic rocks of the İskenderun Gulf. Diagram showing theoretical Rayleigh fractionation vectors for 50% crystallisation of the phase combinations are indicated in the inset (b) Diagram shows theoretical Rayleigh fractionation vectors for crystallization of the phase combinations. The starting composition of the vectors is taken as $\text{Sc} = 22 \text{ ppm}$, $\text{Cr} = 400 \text{ ppm}$. Thick marks on each vector are shown for 5% crystallization intervals. cpx = clinopyroxene.

primary melt compositions. For this reason, the rocks most likely underwent fractional crystallization at different degrees and fractionation correction is required for model studies.

To calculate the primary melt composition, the olivine control line, where the melt is in equilibrium with mantle olivine (e.g. with Fo₉₁; Albaréde, 1992), can be considered as the reference line (Fig. 5). In Fig. 5, assuming that $K_D = [(\text{Fe}/\text{Mg})_{\text{ol}}]/(\text{Fe}/\text{Mg})_{\text{melt}}$ is equal to 0.3, the $\text{FeO}_t/\text{MgO} = 0.9$ (Albaréde, 1992) value can be determined for the melt which is in equilibrium with mantle olivine. In the diagram, primary

melt composition for each sample was determined by adding olivine until the FeO_t/MgO ratio would be equal to 0.9. The amounts of olivine, F (mass fraction of liquid), fractionated from the primary melts were calculated using the determined melt compositions (where the rock compositions were used). F values were obtained by using the olivine control line and the Fe/Mg ratios of the samples. Furthermore, the changes in the MgO contents and the $\text{CaO}/\text{Al}_2\text{O}_3$ ratios of the samples indicate clinopyroxene and olivine fractionation; however, the fractionation of clinopyroxene is more dominant in basanites. (Fig. 6a). The genesis of the rocks can be modeled by 70% clinopyroxene + 30% olivine fractionation according to the mineral vectors attained by Rayleigh fractionation calculations using Cr, and Sc ($^{ol-lq}Kd_{\text{Cr}} \approx 0.700$, $^{cpx-lq}Kd_{\text{Cr}} \approx 34$, $^{ol-lq}Kd_{\text{Sc}} \approx 0.680$, $^{cpx-lq}Kd_{\text{Sc}} \approx 3.2$), which are hosted in olivine and clinopyroxene structures (Fig. 6b) and it is necessary to remove this effect. First, the bulk partition coefficient ($D_{70\%cpx+30\%ol}$) was calculated considering olivine and clinopyroxene fractionation. Then the primitive compositions were estimated according to Rayleigh fractionation law. The fractionation-corrected trace elements used for the melt modeling are given in Table 1.

6.2. Crustal contamination

The samples of the İskenderun Gulf intra-continental basaltic volcanism do not show negative Ta and Nb anomalies on the primitive mantle normalized diagram, indicating no significant crustal contamination (Fig. 3b). In addition, on the Th/Yb vs Ta/Yb diagram, the samples mostly plot on the diagonal mantle array indicating that the melts forming the volcanic suite have not been affected by any significant crustal contamination (Fig. 7). To test the effects of crustal involvement more precisely, the Sr-Nd isotope compositions can be used. For the modeling presented in Fig. 8 the most primitive sample was used as the initial composition, while the composition of the contaminant crust was taken as the average composition of the crustal

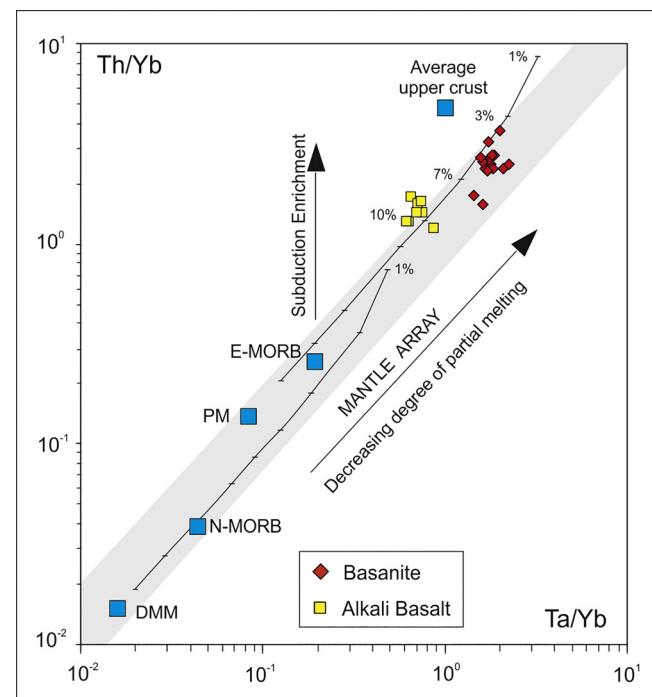


Fig. 7. Th/Yb versus Ta/Yb log-log diagram (after Pearce, 1983) for the İskenderun Gulf volcanic rocks. Some of the typical oceanic basaltic and mantle compositions including the depleted MORB mantle, primitive mantle (PM), N-MORB and E-MORB are also plotted for comparison. Average OIB and MORB values used for comparison are from Sun and McDonough (1989). The average upper crustal composition values are from Taylor and McLennan (1985).

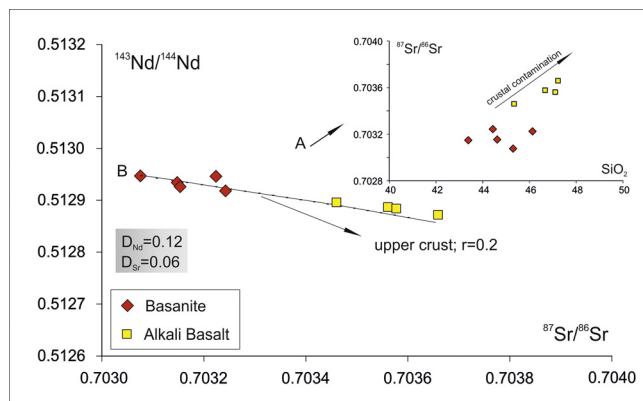


Fig. 8. Modelling of the AFC process based on the equations of DePaolo, (1981) for the İskenderun Gulf volcanic rocks, by using the Sr-Nd isotopic ratios. Crust composition is taken from the average of basement rocks ($\text{Sr} = 650 \text{ ppm}$, $\text{Nd} = 22.5 \text{ ppm}$, $^{87}\text{Sr}/^{86}\text{Sr} = 0.7064$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.5124$) the Arabian Plate in Jordan (Jarrar et al., 2003). Mantle composition is taken this study (O-24; $\text{Sr} = 888.6 \text{ ppm}$, $\text{Nd} = 35.5 \text{ ppm}$, $^{87}\text{Sr}/^{86}\text{Sr} = 0.703076$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.51294$). Bulk distribution coefficients are calculated using the modal mineralogy (70% cpx + 30% ol) of volcanic rocks. The “r” value denotes the rates of mass assimilation to mass crystallization. The inset diagram shows the variation of $^{87}\text{Sr}/^{86}\text{Sr}$ vs SiO_2 for the volcanic rocks.

rocks presented in Jarrar et al. (2003). According to the AFC curve generated from these values, the alkali basalts carry a modest level of contamination ($r = 0.2$). Similarly, increasing ratios of $^{87}\text{Sr}/^{86}\text{Sr}$ together with increased SiO_2 content in alkali basalt support this interpretation (Fig. 8).

6.3. Characteristics of the mantle source

The intra-continental alkaline rocks of the İskenderun Gulf are highly enriched in incompatible elements. These rocks have higher (La/Sm)_N of 2.8–4.2 for basanites and alkali basalts than average OIB (~ 2.4 Sun and McDonough, 1989), reflecting a highly enriched mantle source. Moreover, the enrichment in trace element abundances (e.g. Ta and Nb) for some of the samples compared to the primitive mantle compositions suggests melt derivation from a metasomatized mantle source. The Nd–Sr–Pb isotope compositions of these rocks indicate an origin by melting of an enriched lithospheric mantle source that was chemically modified by an OIB component (e.g. Witte et al., 2017).

Invasion of fluids from asthenosphere-derived melts or lenses of pyroxenite, amphibolite or eclogite may contribute to enrichment of LREE and LILE (Foley, 1992). Similarly, this element enrichment is associated with a metasomatized mantle such as hornblendite, carbonated peridotite and pyroxenite (e.g. Sobolev et al., 2005, 2011; Pilet et al., 2008). Geochemical and isotopic studies on OIB-like rocks point out that the possible composition of these rocks originates from a mixing of different mantle sources. For this reason, involvement of different mantle sources needs to be determined in petrogenetic studies of alkaline rocks (e.g. Sobolev et al., 2005; Niu, 2008).

The melting of metasomatized parts of the mantle can be due to either decompression through mantle upwelling or thermal perturbation in the lithosphere (Lloyd and Bailey, 1975; Wass and Rogers, 1980; Halliday et al., 1990; Pilet et al., 2008) or convective removal and delamination of the metasomatized lithosphere (e.g. in a plume; Halliday et al., 1995; McKenzie and O’Nions, 1995; Niu and O’Hara, 2003; Pilet et al., 2005). Therefore, alkaline magmas can be produced from hydrous metasomatic veins or their equivalents (Pilet et al., 2008).

Although peridotite produces alkaline melts by low degree partial melting at high pressure, a great majority of experimental mafic melts produced from peridotite do not resemble natural alkaline basalts. The experimental products have too low CaO and high Al_2O_3 compared to

natural alkaline basalts (e.g. Hirschmann et al., 2003; Dasgupta et al., 2007). However, it is suggested that low MgO contents of alkaline lavas are not produced solely by fractional crystallization processes and that there should be peridotitic material bearing hydrous phases in the source. The results of melting experiments run by Foley et al. (1999) on hydrous mineral bearing veins such as amphibolite in peridotite host rock have shown that the resulting products had low MgO, SiO_2 and high Al_2O_3 , TiO_2 contents. In general, experimental results obtained by Foley et al. (1999) indicate that lavas with high Ti concentrations may be produced by low-degree melting of the peridotites containing hydrous mineral phases.

In the last years, carbonated silica-poor eclogite and peridotites (Dasgupta et al., 2006, 2007) are thought to be possible sources of silica-undersaturated melts. Experimental studies supporting this idea have shown that the presence of CO_2 in mantle stabilizes orthopyroxene, and prevents it from melting and produces silica-deficient liquids. In the experiments, while the eclogite containing CO_2 produced very high FeO, CaO and $\text{CaO}/\text{Al}_2\text{O}_3$, the peridotite containing CO_2 produced very low FeO, SiO_2 and very high $\text{CaO}/\text{Al}_2\text{O}_3$ melts. In the same way, studies on both possible sources indicate that the K_2O content of melts is very low compared to the high alkali content of silica-poor mafic extrusive rocks, and that the compositions of melts produced by carbonated sources show variations according to the composition of OIB-type rocks (Dasgupta et al., 2007). The high Ti contents of most alkaline OIB can be explained by a metasomatized source that can increase the Ti and alkali concentration of the source (Prytulak and Elliott, 2007). For example, the geochemical composition of high Ti amphibole xenocrysts in the Rhön basalts of the CEVP (Central European Volcanic Province) resembles those of the amphibole veins in the lithospheric mantle and from this it follows that high-Ti alkaline rocks are likely generated from metasomatized mantle sources (Mayer et al., 2014).

Silica-poor pyroxenite is recognized as a source for mafic alkaline magmas, while SiO_2 -enriched pyroxenite is considered as a possible source for tholeiitic magmas (e.g. Hirschmann et al., 2003; Keshav et al., 2004; Zhang et al., 2009). The source lithologies of alkaline magmas are highly enriched in incompatible element such as K, Ba, Rb, and LREE relative to the peridotitic mantle. Silica-poor garnet pyroxenites do not have such enriched trace element concentrations and recycled oceanic crust cannot provide enrichment in this way (Stracke et al., 2003). Only metasomatic processes are expected for such enrichment (e.g. Nielson and Noller, 1987; Harte et al., 1993; Nielson and Wilshire, 1993).

The TiO_2 -rich mantle xenoliths within the alkaline volcanic rocks may occur in pyroxenite veins which are influenced by the involvement of basaltic melts (Witt-Eickschen et al., 1993). On the other hand, Pilet et al. (2008) proposed that the K_2O concentration of alkaline rocks cannot be clarified by silica-poor pyroxenite as a mantle source. Sobolev et al. (2007) indicated that pyroxenite-originated melts are rich in Si and Ni but depleted in Mg, Ca and Mn compared to peridotites. For this reason, such a source cannot explain the concentrations of trace elements and K_2O content in the alkaline basalts of the İskenderun Gulf.

Potassium and TiO_2 enrichment of the alkaline rocks can be explained by phases such as K and Ti-rich amphibole or phlogopite (Foley et al., 1996; Pilet et al., 2008) in the mantle source. The relative abundances of alkaline elements in the volcanic rocks can be used to determine the existence of amphibole or phlogopite in the mantle source as these elements have different partitioning behavior during partial melting. As a result of the incompatibility differences, the melts in equilibrium with phlogopite have considerably higher $\text{Rb}/\text{Sr} (> 0.1)$ and lower $\text{Ba}/\text{Rb} (< 20)$ than the amphibole-bearing mantle sources ($\text{Rb}/\text{Sr} < 0.06$; $\text{Ba}/\text{Rb} > 20$; Furman and Graham, 1999).

While Rb depletion in all samples is apparent, a significant enrichment in Ba (Fig. 3b) shows the effect of these potassic/hydrous phases in the source. The rocks have high Ce/Pb ratios (~ 26–80) from the presumed range for OIB ($\text{Ce}/\text{Pb} = 25 +/- 5$; Hofmann et al., 1986) due

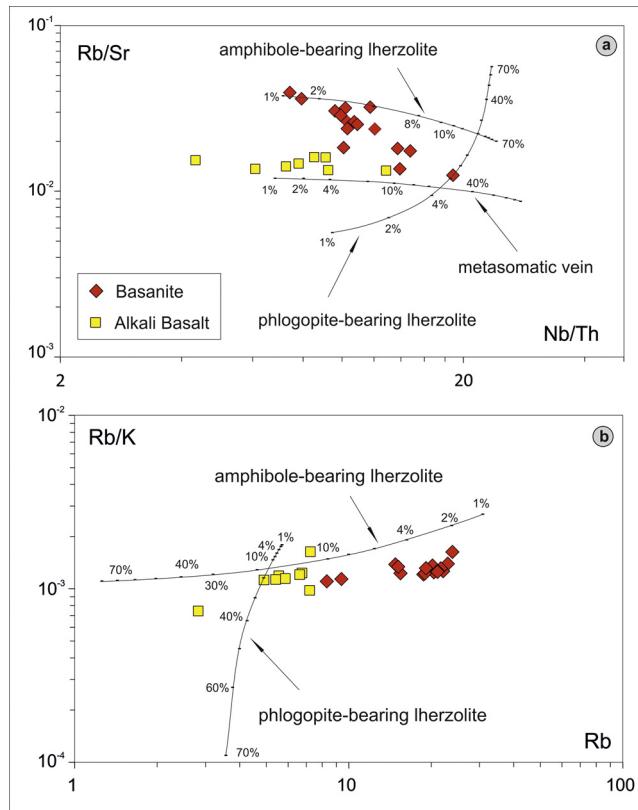


Fig. 9. (a) Nb/Th vs. Rb/Sr, (b) Rb vs. Rb/K covariation for the İskenderun Gulf alkaline rocks. Partial melting curves were calculated using the non-modal batch partial melting of Shaw (1970). Melt trajectories are drawn for amphibole-bearing Iherzolite (with mode and melt mode of $\text{ol}_{0.550} + \text{opx}_{0.19} + \text{cpx}_{0.07} + \text{gt}_{0.08} + \text{amp}_{0.11}$ and $\text{ol}_{0.05} + \text{opx}_{0.15} + \text{cpx}_{0.10} + \text{gt}_{0.3} + \text{amp}_{0.25}$; respectively; Witt-Eickschen et al., 1993); and phlogopite-bearing Iherzolite (with mode and melt mode of $\text{ol}_{0.564} + \text{opx}_{0.188} + \text{cpx}_{0.141} + \text{gt}_{0.047} + \text{phl}_{0.06}$ and $\text{ol}_{0.17} + \text{opx}_{0.19} + \text{cpx}_{0.27} + \text{gt}_{0.02} + \text{phl}_{0.35}$; respectively; Duggen et al., 2005) and amphibole metasomatic vein (with mode and melt mode of $\text{gt}_{0.05} + \text{amp}_{0.95}$ and $\text{cpx}_{0.14} + \text{gt}_{0.16} + \text{amp}_1$; respectively; Pilet et al., 2008). The source compositions calculated using the dynamic melting inversion method proposed by Zou and Zindler (1996); Nb = 2.57 ppm; Th = 0.11 ppm; Rb = 0.85 ppm; Sr = 39.47 ppm; K = 692 ppm. Partition coefficients used are from the compilation of McKenzie and O’Nions (1991, 1995), Green (1994) and LaTourrette et al. (1995).

to low Pb concentrations (0.8–2.9 ppm). It is generally supposed that amphibole and phlogopite are the main phases for Pb in the mantle (Rosenbaum, 1993). Thus, mantle melting that leaves amphibole or phlogopite-bearing residues would likely produce melts depleted in Pb. The low Rb/Sr (0.01–0.04) and high Ba/Rb (9.7–40.7) ratios of the mafic alkaline samples show that the main phase was amphibole in mantle source. In addition, Pilet et al. (2010) indicated that the melting temperature remarkably changes the Al_2O_3 and MgO abundance of the melt. Melting at low temperatures causes lower MgO but higher Al_2O_3 abundances in the melt, compared to common alkali basalts with similar SiO_2 and minor variations of TiO_2 contents (e.g. Mayer et al., 2013).

The Nb/Th vs. Rb/Sr variation suggested by Furman and Graham (1999) can also be used to test whether the hydrous phases in the mantle source is amphibole or phlogopite (Fig. 9a). Here, the Nb/Th ratios of the samples may be controlled by the hydrous phase as Nb is more compatible in amphibole than phlogopite (${}^{\text{amp}}\text{Kd}_{\text{Nb}} = 0.159$, ${}^{\text{phl}}\text{Kd}_{\text{Nb}} = 0.085$; LaTourrette et al., 1995). The Nb/Th ratio in most of the samples increases with decreasing Rb/Sr ratio. In Fig. 9a, the samples also plot between the melting curves defined for amphibole-

bearing Iherzolite and amphibolite vein and this may indicate the existence of residual amphibole in the mantle source during the melting process. Similarly, Rb/K vs. Rb variations may provide an additional constraint in defining the possible hydrous phase in the mantle residue. Since K is less compatible in amphibole than in phlogopite (${}^{\text{amp}}\text{Kd}_K = 0.58$, ${}^{\text{phl}}\text{Kd}_K = 3.67$; LaTourrette et al., 1995), melting of the amphibole-bearing Iherzolite would not much affect the Rb/K ratio, while this ratio is affected significantly during melting in the presence of residual phlogopite. The increase in Rb abundances and insignificant change in Rb/K ratios of the samples on the diagram, therefore, further support amphibole-bearing Iherzolite source rather than phlogopite-bearing source (Fig. 9b).

Combined together, it can be considered that hornblendite or amphibole-rich veins may be a more suitable source for alkaline rocks of the İskenderun Gulf. Pilet et al. (2008) used experiments in amphibole-bearing metasomatic veins at 1.5 GPa to suggest that partial melting of amphibole-rich parts of the mantle would reflect the geochemical properties of continental and oceanic alkaline magmas. These melting experiments have shown that amphibole-bearing veins represent the source which may produce large fraction of melt due to its low solidus temperature. These melts have high alkalis and TiO_2 and show high LREE/HREE (e.g. $(\text{La/Yb})_N > 5$) at reasonable level and intermediate MREE/HREE (e.g. $(\text{Dy/Yb})_N > 1.6$) along with Rb, K, Pb, and Zr anomalies in the primitive mantle normalized diagrams.

Partial melting of amphibole-rich veins, such as hornblendite, cpx-hornblendite and hornblendite + peridotite sandwiches, may generally produce mafic melts with high alkali content. Especially, the partial melting of the hornblendite + peridotite sandwich can produce mafic melts with high SiO_2 content (Dai et al., 2014). On the other hand, the pure hornblendite melts have very low content of SiO_2 in comparison to alkaline rocks of the İskenderun Gulf.

Using the trace element partition coefficients, the partial melting conditions of the source of parental İskenderun Gulf alkaline rocks can be modeled and the compositional variations of these alkaline magmas resulting in a mixture between amphibole-bearing peridotite and peridotite melts can be tested (e.g. Ma et al., 2011a). In the Dy/Yb vs La/Yb diagram, non-modal batch melting models were used to constrain the melting regime and the involvement of garnet in the source (Fig. 10). The peridotite and amphibole-bearing peridotite compositions and their modal compositions are given in Table 3. Primitive mantle compositions (Sun and McDonough, 1989) were used for peridotite and

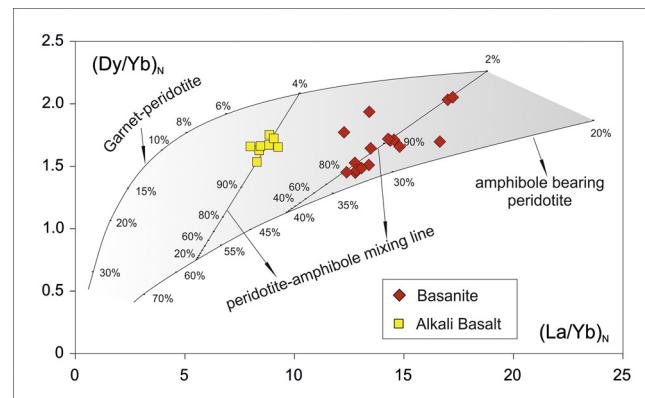


Fig. 10. La/Yb vs. Dy/Yb covariation for the İskenderun Gulf alkaline rocks. The continuous curves show non-modal batch melting models for peridotite (with a primitive mantle composition) and amphibole-bearing peridotite metasomatic vein as starting materials. It should be noted that most basalt samples fall in the grey-shaded fields, which define the possible melt compositions from mixing of small-degree ($F = 2\text{--}4\%$) peridotite melts and large-degree ($F = 40\text{--}60\%$) amphibole-bearing peridotite metasomatic vein melts. Source compositions and partition coefficient values are given in Table 3. F = degree of partial melting.

Table 3

Mineral-melt partition coefficients, source compositions and source modes used in partial melting modelling.

Partition coefficient							
	Ol	Opx	Cpx	Gt	Amp		
					Source (ppm)		
					PM vein		
La	0.0004	0.002	0.054	0.01	0.17	0.687	52.7
Yb	0.0015	0.049	0.28	4.03	0.59	0.493	2.54
Dy	0.022	0.33	0.055	1	0.78	0.737	7

Source Mode					
	Ol	Opx	Cpx	Gt	Amp
Garnet-bearing lherzolite					
Mineral mode	0.598	0.211	0.076	0.115	
Melting mode	0.05	0.20	0.30	0.45	
Garnet-bearing amphibole metasomatic vein					
Mineral mode	0.550	0.19	0.07	0.08	0.11
Melting mode	0.05	0.15	0.10	0.30	0.25

Ol, olivine; Opx, orthopyroxene; Cpx, clinopyroxene; Gt, garnet; Amph, amphibole; PM, primitive mantle of Sun and McDonough (1989); metasomatic vein, AG4 (Pilet et al., 2008). Garnet bearing lherzolite are taken from Thirlwall et al. (1994). Garnet-bearing amphibole is taken from (Witt-Eickschen et al., 1998); mineral/matrix partition coefficients are from the compilation of McKenzie and O'Nions (1991, 1995).

amphibole-bearing peridotite source components in the calculations. The use of the primitive mantle composition does not mean that the source is the primordial mantle. Compared to bulk earth, the samples having slightly decreased Sr, increased Nd isotopic ratios indicating long-term depletion in the source. Therefore, this primitive mantle composition should mirror a re-fertilized mantle source initially depleted but subsequently enriched by melts that are probably close to a primitive mantle composition or are enriched in incompatible elements (Ma et al., 2011a).

The Dy/Yb ratio of melt is normally controlled by garnet, because of the high compatibility of HREE in garnet. The Dy/Yb ratio can be decreased by the addition of amphibole due to the higher MREE compatibility in amphibole. Most of the samples plot between the peridotite and amphibole-bearing peridotite melting curves calculated by the melting model. They can be interpreted as the products of varying degrees of mixing between amphibole-bearing peridotite-derived (~40–60% melt fractions) and peridotite-derived melts (~2–4% melt fractions). According to this melting model, contribution of the amphibole-bearing metasomatic veins in the alkaline basalts of the İskenderun Gulf is approximately 40% or more.

6.4. Development of the metasomatised source region

Metasomatic veins rich in amphibole, which appear to be the main source of common alkaline magmas (Pilet et al., 2008, Pilet et al., 2011; Ma et al., 2011a, b) are usually considered to have formed in the lithospheric mantle during melt ascent and migration. The interaction of hydrous melt-peridotite (Rapp et al., 1999; Prouteau et al., 2001) produces these veins which also contain xenoliths derived from different depths in mantle (e.g. Downes et al., 2004; Powell et al., 2004). Melts derived from such veins are strongly silica-undersaturated, but when these melts react with the orthopyroxene in the host peridotite, partially siliceous olivine-rich mafic melts form (Hirschmann et al., 2003; Pilet et al., 2008).

Anhydrous and hydrous metasomatic veins are interpreted as being produced during progressive differentiation of primary hydrous alkaline magmas at high pressure. The comparison between the products formed by fractional crystallization of Ne-normative or Hy-normative initial compositions (hawaiite or picropic basalt) shows that all hydrous liquids generated by differentiation are generally similar. For this reason, it has been suggested that the development of amphibole-bearing metasomatic veins in lithospheric mantle may be related to the

change of the initial melts from Ne-normative to Hy-normative (Pilet et al., 2010). However, percolation and differentiation of a liquid produced at a low degree of partial melting of a mantle source enriched slightly or similar to MORB supports the possibility of hydro-metasomatic veins being interpreted elsewhere as a possible source for alkaline magmatism.

Niu (2008, 2009), and Humphreys and Niu (2009) suggested that H₂O and CO₂-rich primary melts migrating upward would metasomatize the base of the lithosphere with a melt-rich level above the low velocity zone (LVZ). High-pressure equilibrium crystallization experiments in hydrous, CO₂-containing basanitic melts show that mainly cpx-dominated, anhydrous mineral assemblages such as olivine ± garnet ± orthopyroxene in varying amounts occur in the host rock because of percolation processes in the lithosphere (e.g. Harte et al., 1993; Pilet et al., 2010). However, amphibole crystallized predominantly in the residual melts due to varying composition of the melts with more percolation of melts. Finally, hydrous crystallizing assemblages in addition to amphibole may contain various secondary phases such as clinopyroxene ± plagioclase ± phlogopite ± orthopyroxene ± garnet, and rutile and apatite. The compositions of amphiboles derived from metasomatic veins are similar to those of amphiboles that crystallized in FC experiments. In this case, the metasomatic veins are interpreted to have developed during the differentiation of the primary or derivative melts in the lithospheric mantle (Pilet et al., 2010). In addition, the solidus temperatures of amphibole-bearing metasomatic veins (e.g. Pilet et al., 2008) are considered to be lower than those of the anhydrous assemblages (garnet + clinopyroxene ± olivine) forming the surrounding peridotite matrix.

6.5. Geodynamic Implications and cause of mantle melting

Boundaries between Africa, Arabia and Eurasian plates are irregular in the Mediterranean region and geodynamics of the region is determined by the relative movements of the plates. Bitlis suture constitutes the border between Anatolia and Arabia plates and a thrust in the northern part of the Amanos constitutes the border between Anatolia and Africa (Chorowicz et al., 1994). The Dead Sea Fault Zone, developed due to the movements of Africa, Arabia and Eurasia plates, starts from the Red Sea and Maraş extends to the triple junction and the Bitlis suture zone (Fig. 1). Trans-tensional movements along the Dead Sea fault zone cause lithospheric extension (Lustrino and Sharkov, 2006). The alkali volcanism of the İskenderun Gulf exists on the plate

boundaries of Africa-Arabia and lies along the active left-lateral northeast-southwest trending Karatas-Osmaniye Fault Zone (Yurtmen et al., 2000).

In the south of Turkey, the trace element concentrations of the mafic extrusive rocks of the İskenderun Gulf are explained by the metasomatic nature of the source. Although the samples have high Rb/Sr and low Sm/Nd ratios, the Sr-Nd isotopic signature ($^{87}\text{Sr}/^{86}\text{Sr} = 0.7031\text{--}0.7036$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.5218\text{--}0.5129$) points rather towards a depleted source, suggesting a young age for the metasomatism. The geochemical characteristics of the rocks imply hydrous metasomatized veins and the mixing of these veins with peridotites, and low solidus temperatures are sufficient for the melting of such a source. These types of metasomatized mantle components containing amphibole-bearing vein assemblages have also been proposed to be the main source of the alkaline basaltic suites located further south in north-western Syria and in Israel, where hornblendite mantle xenoliths have been reported (e.g. Downes et al., 2004). In addition, experimental studies on the pressures over 3 GPa suggested that there is a significant change in the upper mantle water storage capacity (dropping from $\sim 0.3\text{--}0.4$ wt% to 180 ppm) due to pargasite instability (Green et al., 2010, 2011). The depth that corresponds to this pressure (~ 100 km) indicates the lithosphere-asthenosphere boundaries of a mature oceanic lithosphere as well as the non-cratonic continental lithosphere in the region. Ma et al. (2013) suggested that hydrous interstitial melts present at the boundary of the lithosphere-asthenosphere are responsible for the seismic low-velocity zone observed under many mature ocean basins and in much of Northwest and Western Arabia (Ma et al., 2013). Upward migration of hydrous melts during localized lithospheric extension in the region provides a reasonable mechanism for the evolution of metasomatic veins which produced the alkaline volcanism of the İskenderun Gulf due to decompressional melting caused by trans-tensional tectonism in the region.

7. Conclusions

The alkaline rocks of the İskenderun Gulf with low silica content are classified as basanite and alkali basalt, being sodic and nepheline-normative in composition. Fractionation vectors calculated theoretically from trace element data indicate that clinopyroxene and olivine are the main mineral phases in crystallization assemblages of the rocks. The Sr, Nd and Pb isotopic characteristics of volcanic rocks are similar to those of other occurrences with OIB affinity in the region and NW Syria. The Sr-Nd isotopic data and the assimilation and fractional crystallization (AFC) models suggest that alkali basalts were slightly contaminated by the upper crust material. High TiO_2 and Al_2O_3 contents combined with radiogenic isotope and trace element characteristics show the effect of metasomatized veins in the formation of OIB-type magmas. Trace element modeling shows variable degrees of mixing between amphibole-bearing peridotite and peridotite-derived melts and the involvement of metasomatic veins containing at least 40% amphibole-bearing peridotite in the source of the İskenderun Gulf alkaline rocks. The alkali basalts and basanitic rocks are generated by different extent of mixing of these sources. The decompressional melting due to trans-tensional tectonic in metasomatic veins is the major process for formation of the alkaline rocks of the İskenderun Gulf.

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