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# **Epithermal paleosurfaces**

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Abstract Many active volcanic-hydrothermal and geothermal systems are characterized by distinctive surface and nearsurface landforms and products, which are generated during discharge of a spectrum of fluid types under varied conditions. Remnants of most of these products are preserved in some of their less-eroded, extinct equivalents: epithermal deposits of high-sulfidation (HS), intermediate-sulfidation (IS), and lowsulfidation (LS) types. Steam-heated alteration occupying vadose zones and any underlying silicified horizons formed at paleogroundwater tables characterize HS, IS, and LS deposits as do hydrothermal eruption craters and their subaerial or shallow sub-lacustrine breccia aprons and laminated infill. Although rarely recognized, HS, IS, and LS systems can also contain finely laminated, amorphous silica sediments that accumulated in acidic lakes and mud pots and, exclusive to HS systems, in hyperacidic crater lakes. In contrast, silica sinter and more distal carbonate travertine are hot spring discharge products confined mainly to LS and IS settings, as both form from near-neutral-pH liquids. Hydrothermal chert deposition and sediment silicification can take place in shallow, lacustrine rift settings, also largely restricted to LS and IS deposits. These surface and near-surface hydrothermal products are typically metal deficient, although mercury concentrations are relatively commonplace and were formerly exploited in places. Nonetheless, sinters, hydrothermal eruption craters, and silicified lacustrine sediments may contain anomalously high precious metal values; indeed, the last of these locally constitutes low-grade, bulk-tonnage orebodies. The dynamic nature of epithermal paleosurfaces, caused by either syn-hydrothermal

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Richard H. Sillitoe aucu@compuserve.com aggradation or degradation, can profoundly affect deposit evolution, leading to either eventual burial or telescoping of shallower over deeper alteration  $\pm$  precious metal mineralization. Formational age, tectonic and climatic regime, hydrothermal silica content and texture, and post-mineralization burial history combine to determine the preservation potential of paleosurface products. Proper identification and interpretation of paleosurface products can facilitate epithermal precious metal exploration. Proximal sinters and hydrothermal eruption craters may mark sites of concealed epithermal mineralization, whereas paleogroundwater table silicification and steam-heated blankets can be more widely developed and, hence, less diagnostic. Epithermal precious metal deposits may immediately underlie paleosurface features but are commonly separated from them by up to several hundred vertical meters, especially in the case of IS deposits. Furthermore, the tops of concealed, particularly IS epithermal orebodies in any particular district, irrespective of whether or not paleosurface features are preserved, can also vary by several hundred vertical meters, thereby imposing an additional exploration challenge. Precious metal contents of paleosurface products are unreliable but nonetheless potentially useful guides to concealed deposits. However, sub-paleosurface geochemical anomalism, particularly for arsenic and antimony, may indicate proximity to subjacent ore.

**Keywords** Epithermal deposits · Precious metals · Paleosurfaces · Hot springs · Exploration

# Introduction

Epithermal precious metal deposits form in the shallow parts of volcanic fields, including associated volcano-sedimentary basins, typically at paleodepths of <1 km. Hence, it is unsurprising that many of them are accompanied by surface and near-surface hydrothermal manifestations albeit generally



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devoid of economic precious metal concentrations. This fact has been appreciated for more than a century, as documented by numerous early discussions of likely relationships between epithermal deposits and hot spring activity (e.g., Lindgren 1933; Schmitt 1950; White 1955). Although the genesis of epithermal deposits can be geologically brief, perhaps as little as a few tens of thousands of years (Simmons and Brown 2006; Rowland and Simmons 2012), the volcanic and/or sedimentary land surfaces overlying the deposits at the time of formation are commonly dynamic and can undergo appreciable change, thereby profoundly influencing their surface and near-surface manifestations and even formational history.

Epithermal deposits are commonly subdivided into highsulfidation (HS), intermediate-sulfidation (IS), and lowsulfidation (LS) end-member types on the basis of their mineralogic characteristics (Hedenquist et al. 2000; Sillitoe and Hedenquist 2003; Table 1), although it is recognized that mineralogic evolution and complexity make it difficult to neatly fit some deposits into this end-member classification scheme. These three epithermal types also tend to occupy distinct volcano-sedimentary settings, characterized by different hydrologic regimes (Henley and Ellis 1983). The HS deposits typify andesitic to dacitic volcanic edifices-stratovolcanoes, dome complexes, and any associated maar volcanoes and underlying diatremes (Sillitoe and Bonham 1984), commonly in the shallow parts of porphyry copper systems-whereas LS deposits are characteristic of rift settings in which bimodal (basalt-rhyolite) volcanism and fluvio-lacustrine sedimentation are commonplace (Sillitoe 1993; Sillitoe and Hedenquist 2003). The IS deposits, mineralogically intermediate between the HS and LS types, can form in either volcanic centers or volcano-tectonic rifts. Both HS and IS deposits and LS and IS deposits can be closely associated and even transitional to one another (Sillitoe 1999; Hedenquist et al. 2000; Sillitoe and Hedenquist 2003; Camprubí and Albinson 2007), whereas HS and LS deposits normally appear to be incompatible. Epithermal deposits of LS and, to a much lesser extent, IS type in close proximity to remanent paleosurface features were initially assigned to a separate hot spring category (e.g., Berger and Eimon 1983; Nelson 1988), but are now generally treated as the shallowly formed parts of standard epithermal systems (Hedenquist et al. 2000).

In this practically oriented overview, the well-known surface manifestations of active hydrothermal systems (e.g., Hochstein and Browne 2000) are briefly summarized first in order to provide present-day analogues for epithermal paleosurface features. The principal paleosurface features associated with HS, IS, and LS epithermal deposits are then described and discussed in their overall volcanic and geomorphologic contexts (Table 2). Particular emphasis is placed on the resultant hydrothermal products and their origins—using detailed studies of active analogues wherever possible, any associated metal and metalloid anomalism, consequences for deposit-scale evolution, and preservation potential. Finally, a number of exploration implications and issues are considered.

Paleosurface features, as discussed herein, comprise hydrothermal landforms and associated surface accumulations as well as hydrothermal alteration zones generated between paleosurfaces and underlying paleogroundwater tables. In active epithermal systems, pressure gradients are controlled by the hydrostatic head that, in turn, is determined by the groundwater table or piezometric surface (Fournier 1983; Henley 1985; Simmons 2002). The distance between paleosurfaces and paleogroundwater tables is variable, ranging from close to zero in low-relief areas to as much as several hundred meters in steep terrain, especially where arid climatic conditions prevail. The alteration contributions of supergene weathering, also controlled by paleo- and present-day groundwater tables, are largely excluded from this overview, although it is recognized that hypogene and supergene processes can be transitional in the shallow epithermal environment and, in places, difficult to separate unambiguously.

Table 1 Summary characteristics of epithermal deposit types

Epithermal type	High-sulfidation (HS)	Intermediate-sulfidation (IS)	Low-sulfidation (LS)
Main mineralization styles	Steep and shallowly inclined replacement bodies, hydrothermal breecias	Veins, stockworks	Veins, stockworks, disseminated bodies
Main proximal alteration types	Silicification, vuggy residual quartz, quartz-alunite	Silicification, quartz-sericite/illite	Silicification, quartz-adularia- illite
Main gangue minerals	Quartz, alunite, barite	Quartz, calcite, manganoan carbonates, rhodonite, adularia	Quartz, chalcedony, adularia
Sulfide abundance	High (10-80 vol.%)	Moderate (5–30 vol.%)	Low (1–5 vol.%)
Sulfidation-state indicators	Enargite/luzonite/famatinite	Tetrahedrite, chalcopyrite, low-Fe sphalerite	Pyrrhotite, arsenopyrite, high-Fe sphalerite
Typical metal signature	Au-Ag-Cu $\pm$ Bi $\pm$ Te	Ag-Au-Zn-Pb-Mn $\pm$ Cu	$Au\pm Ag\pm Se\pm Mo$

Summarized from Einaudi et al. (2003), Sillitoe and Hedenquist (2003)

Table 2	Summary	characteristics	of epithermal	paleosurface	features
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Paleosurface feature	Associated epithermal deposit type	Proximity to hydrothermal upflow zone	Hydrothermal fluid responsible	Main component mineral(s)
Steam-heated zones	HS, IS, LS	Proximal but widespread	$H_2S$ -bearing steam condensate	Opal/chalcedony, alunite, kaolinite, smectite
Groundwater table silicification	HS, IS, LS	Proximal but widespread	H <sub>2</sub> S-bearing steam condensate	Opal/chalcedony
Lacustrine amorphous silica sediments	HS, IS, LS	Commonly proximal	H <sub>2</sub> S-bearing steam and SO <sub>2</sub> - and HCl-bearing magmatic condensates	Opal, cristobalite
Hydrothermal eruption craters and breccias	HS, IS, LS	Commonly proximal	Neutral-pH chloride water	Illite, smectite
Hot spring sinter	IS, LS	Commonly proximal	Neutral-pH chloride water	Opal/chalcedony
Hot spring travertine	IS, LS	Distal	CO <sub>2</sub> -rich water	Calcite, aragonite
Hydrothermal chert	IS, LS	Proximal-to-distal	Neutral-pH chloride water	Opal/chalcedony
Silicified lacustrine sediments	IS, LS	Proximal-to-distal	Neutral-pH chloride water	Opal/chalcedony

Taken from references cited in the text and personal observations

## Active hydrothermal systems

Active hydrothermal systems, some exploited as power sources, are sites of convective and lesser conductive dissipation of heat energy, and many of them are manifested at surface by one or more types of fluid discharge. This gives rise to a variety of distinctive landforms and products, which can be partially preserved in shallowly eroded epithermal precious metal deposits, as discussed further in the following sections.

For the purposes of the following discussion, two endmember types of high-temperature hydrothermal systems may be recognized: volcanic-hydrothermal and geothermal, with the latter subdivided into those associated with high-relief, volcanic and low-relief, rift settings (Henley and Ellis 1983; Hedenquist and Arribas 1999; Hedenquist et al. 2000; Hochstein and Browne 2000; Simmons et al. 2005; Fig. 1). In general, volcanic-hydrothermal systems can generate HS and IS epithermal mineralization, whereas geothermal systems are the sites of IS and LS mineralization.

On the summits and upper slopes of volcanic edifices overlying volcanic-hydrothermal systems, magmatic water vapor with SO<sub>2</sub>, HCl, CO<sub>2</sub>, and H<sub>2</sub>S gases discharges directly to the atmosphere via high-temperature solfataras and fumaroles (Fig. 2a). Condensation of the vapor and its contained gases into groundwater produces highly acidic (pH  $\leq$ 1) and corrosive waters, which can be ponded in hyperacidic crater lakes and mud pots and emitted from hot springs (Hedenquist 1995; Fig. 1a). Warm and cold springs, both acidic and neutral pH, can debouch on the lower volcano slopes, up to ~10 km from the summits (e.g., Giggenbach et al. 1990; Fig. 1a).

Geothermal systems are characterized by relatively reduced, neutral-pH, alkali chloride waters, although the magmatic component can still be appreciable (e.g., Sillitoe and Hedenquist 2003), particularly where systems are developed beneath high-relief volcanic centers. In such situations, the deeper, neutral-pH water fails to attain the surface at the higher elevations and steam separated from the boiling liquid at depth dominates and gives rise to surficial steam-heated features, including fumaroles, mud pots, and steaming ground (Fig. 2b) as well as outflow of acidic water (Fig. 1b). Discharging acidic water can also accumulate in shallow lakes, some occupying hydrothermal eruption craters. The low-pH ( $\sim 2-3$ ), acid-sulfate water responsible for the steamheated alteration results from groundwater absorption of H<sub>2</sub>Sbearing steam derived from boiling of the deeper, neutral-pH geothermal water (White et al. 1956; White 1967; Schoen 1969; Schoen et al. 1974). Lateral flow of both the steamheated and underlying alkali chloride waters, the latter for distances as great as ~10 km, is accompanied by CO2-rich marginal water (Hedenquist 1990), which becomes bicarbonate dominated as  $CO_2$  is lost by effervescence. Distal springs discharge groundwater-diluted chloride, bicarbonate, or hybrid bicarbonate-sulfate waters capable of precipitating silica sinter if not too dilute and cool, with the bicarbonate waters also forming travertine, surficial carbonate accumulations (Fig. 1b).

In marked contrast, geothermal systems hosted within lowrelief terrain commonly discharge neutral-pH, alkali chloride waters along streams and on lake margins and bottoms, although shallow boiling also gives rise to nearby steamheated features above the groundwater table (Fig. 1c). The discharging alkali chloride waters, which can be channeled by the breccia conduits beneath hydrothermal eruption craters, typically form silica sinters (Fig. 1c). The active steam-heated environment is characterized by the same hydrothermal features observed in high-relief geothermal systems (Fig. 2b). The peripheral parts of these low-relief geothermal systems



Fig. 1 Schematic representation of active, high-temperature hydrothermal systems and their principal surface features (inspired by Hedenquist and Arribas 1999; Hochstein and Browne 2000; Simmons et al. 2005). a Volcanic-hydrothermal system above shallow intrusion and dominated by magmatic fluids. b Geothermal system above deep intrusion in high-relief terrain, with ascendant neutral-pH, alkali

chloride water unable to attain the surface except in outflow zones. **c** Geothermal system above deep intrusion in low-relief terrain, with deep neutral-pH, alkali chloride water reaching the surface. Note that steamheated alteration above groundwater tables is far less extensive in **c** than in **a** or **b** 



Fig. 2 Contrasting surface discharges from active systems. **a** Magmatic volatiles, rich in SO<sub>2</sub> and HCl, passively discharging at temperatures of 800–900 °C from a volcanic-hydrothermal system at the Satsuma-Iwojima volcano, a rhyolite dome in Kyushu, Japan (see Hedenquist et al. 1994; Kazahaya et al. 2002). **b** Steaming ground at Cathedral Rocks above a paleogroundwater table represented by the level of Frying Pan Lake (6-m deep, pH 3.5, 55 °C), Waimangu geothermal system, Taupo Volcanic Zone, North Island, New Zealand (see Simmons et al. 1993)

also tend to be dominated by weakly acidic, CO<sub>2</sub>-rich waters, which can precipitate travertine on surface discharge (Fig. 1c).

# **Epithermal paleosurface products**

# Steam-heated zones

The reactive H<sub>2</sub>S-bearing steam condensate formed in the vadose zone above groundwater tables causes advanced argillic alteration (Fig. 3) and, in poorly consolidated volcanic material, can produce near-surface cavities (Fig. 4a) and resultant collapse features. At the prevailing low (<~120 °C) temperatures, the advanced argillic assemblage comprises powdery opaline silica, alunite, and/or kaolinite (Fig. 4b), depending on solution pH. The opaline silica, which is diagenetically transformed to microcrystalline quartz (chalcedony; White et al. 1956), forms at pH 2-3, whereas kaolinite and more distal smectite characterize rock volumes subjected to lessacidic conditions (Fig. 3). Minor disseminated pyrite can occur in the proximal portions of steam-heated blankets, but it tends to be promptly oxidized as the foci of steam emission migrate. Steam-heated alteration is everywhere characterized by its low density, typically <2. Actively forming steamheated zones, like that capping part of the Ladolam LS gold deposit, Lihir Island, Papua New Guinea, are represented at surface by steaming ground and corresponding vegetation-kill zones (Carman 2003; Fig. 2b).



Fig. 3 Schematic relationships between a steam-heated zone, lacustrine amorphous silica sediments, groundwater table silicification, silica sinter, hydrothermal eruption crater and breccia, hydrothermal chert, and silicified lacustrine sediments. Note the pH-controlled mineralogic

zoning of the steam-heated zone, temperature-controlled proximal-todistal zoning of the sinter, and sub-groundwater table kaolinization caused by drain-back of acidic water. *GWT* groundwater table



**Fig. 4** Images of steam-heated alteration and paleogroundwater table silicification. **a** Cavity produced by rock dissolution in steam-heated remnant above the advanced argillic lithocap of the Marte porphyry gold deposit, northern Chile (see Sillitoe 1994). **b** Steam-heated alteration exposed in a road cut above the Coipa Norte HS gold-silver orebody at La Coipa, northern Chile. Note its white, friable character. **c** Steam-heated alteration around the outcropping, weakly mineralized upper part of the Tres Reyes vein at the Arcata IS silver deposit, Peru (see Candiotti et al. 1990). **d** Subhorizontal zone of paleogroundwater table silicification (forming the present surface) alongside the Furtei HS

gold deposit, Sardinia, Italy. Note the formerly exploited kaolinite quarries beneath the silicified horizon. The kaolinization was generated by drain-back of acidic water from an overlying, but now-eroded steam-heated zone. **e** Subhorizontal zone of paleogroundwater table silicification and overlying steam-heated alteration peripheral to the La Coipa HS-IS gold-silver district, Chile. **f** Hydrothermal breccia, generated by shallow-seated steam eruption, cutting steam-heated zone, Pascua-Lama HS gold-silver deposit, Chile. Geologist stands just left of contact with non-brecciated rocks

Since boiling of ascendant fluids and upward liberation of H<sub>2</sub>S-bearing steam are features of all types of epithermal systems, HS, IS, and LS deposits can be capped by steam-heated zones (Sillitoe 1993; Hedenquist et al. 2000). These blankets are essentially indistinguishable from one another because the acidic solutions and, hence, the resultant minerals are essentially the same in all cases. Steam-heated zones above many HS and some IS deposits tend to be thicker than those above LS deposits because groundwater tables are typically deeper in high-relief volcanic edifices than in topographically subdued rift settings. Indeed, LS deposits may be characterized by only thin veneers of steam-heated alteration or may lack it entirely as, for instance, in lacustrine settings (Fig. 3). Since the thickness of steam-heated zones is influenced by the depth of groundwater tables, they also tend to be thicker where prevailing climatic regimes are arid rather than pluvial. Aridity also decreases groundwater recharge and consequent dilution, thereby allowing buildup of the steam condensate. The anoxic atmosphere during the Archean precluded steam-heated alteration because the liberated H<sub>2</sub>S would have remained unoxidized, as exemplified by the North Pole epithermal system, Western Australia (Harris et al. 2009).

Native sulfur, a product of H<sub>2</sub>S oxidation by atmospheric oxygen under acidic conditions, is widespread in steamheated zones but tends to be more abundant in those developed at the tops of HS systems, from which it was formerly exploited in places (e.g., La Coipa, Chile; El Queva, Argentina; Sillitoe 1975). However, much of the native sulfur  $(\pm \text{ iron sulfides})$  at the tops of HS systems, like that formerly exploited from the summit regions of some andesitic-dacitic arc volcanoes (e.g., Aucanquilcha, Chile; Matsuo, Honshu, Japan; White Island, New Zealand) and still mined in artisanal fashion at Kawah Ijen, Java, Indonesia, was probably more commonly deposited from solfataras fed directly by magmatic gas with variable  $SO_2/H_2S$  ratios (Mukaiyama 1970; Giggenbach 1987). Temperature rise (>~115 °C) and consequent fusion of such solfataric sulfur accumulations can lead to sulfur pools and flows (Watanabe 1940; Naranjo 1985; Oppenheimer 1992).

Steam-heated alteration generally constitutes areally extensive blankets, remnants of which are still observed at many epithermal precious metal deposits of HS (e.g., Yanacocha, Peru, Veladero, Argentina, and Pascua-Lama, Chile-Argentina; Harvey et al. 1999; Deyell et al. 2005), IS (Arcata, Peru; Candiotti et al. 1990), and LS types (Wind Mountain, Hycroft [Crowfoot-Lewis], and Florida Canyon, Nevada; Wood 1991; Ebert and Rye 1997; Fifarek et al. 2010). However, the basal remnants of steam-heated zones can have upward-flared, funnel-shaped geometries centered on individual veins, vuggy residual quartz ribs, faults, or fracture zones (e.g., Arcata; Fig. 4c).

Kaolinite alteration is commonly observed as root zones of steam-heated blankets, either as laterally extensive but vertically restricted zones (Fig. 3) or alongside the shallow parts of LS and IS veins. The kaolinite can also line or fill open spaces in the veins themselves. Although this kaolinization can be formed by descent of groundwater tables, it is also due to permeability-controlled, commonly structurally guided drain-back of steam-heated waters, probably as the epithermal systems waned. Kaolinite of this type, beneath paleogroundwater table silicification, was formerly exploited alongside the Furtei HS gold deposit in Sardinia, Italy (Fig. 4d). At Mt. Muro in Central Kalimantan, Indonesia, late-stage kaolinite developed alongside and within some of the auriferous veins as the IS system waned (Simmons and Browne 1990).

## Groundwater table silicification

The groundwater table, marking the subhorizontal bases of steam-heated zones, can undergo massive silicification (Sillitoe 1993; Hedenquist et al. 2000). The silicified zones range from a few to several tens of meters in thickness and are normally subhorizontal unless subjected to later tilting (Figs. 3 and 4d, e). The silica is initially precipitated as replacement opal, but progressively transforms to chalcedony as a result of recrystallization. Limonite impregnation and staining developed by supergene oxidation of pyrite  $\pm$  marcasite are commonplace.

Much of the silica precipitated at groundwater tables is dissolved in the vadose zone by steam condensate, typically from volcanic glass contained in tuffs. The condensate flows laterally at the groundwater table or within perched aquifers, following local hydraulic gradients. This implies the likelihood of greater silica transport and deposition, at distances up to several kilometers, in high-relief areas than in rifts or other topographically subdued volcanic settings. The silicified horizons are commonly hydrothermally brecciated, with the clasts variably re-cemented by opaline silica or derivative chalcedony. The brecciation is attributed to sealing of the impermeable silicified horizons and eventual buildup of pressures and rupturing to form shallow-seated steam eruptions, which can also cause brecciation of overlying steam-heated blankets (Mongillo and Allis 1988; Fig. 4f).

In several epithermal districts, groundwater table silicification can be traced for several kilometers along their apparent margins, although it has been largely eroded or simply never formed above most of the known orebodies, as observed near the Furtei HS gold deposit (Fig. 4d), in the La Coipa HS-IS gold-silver district (Fig. 4e), in the Fresnillo IS silver district, Mexico (Albinson 1988; Sawkins 1988), and in the Goldbanks LS gold district, Nevada (Stone et al. 2000; Ellis and Stroup 2015). At the Hycroft gold deposit, an opaline silica (opalite) horizon, up to 12-m thick, underlies the extensive steam-heated blanket (Ebert and Rye 1997).

## Lacustrine amorphous silica sediments

Acidic lakes and mud pots can exist in HS, IS, and LS systems where the acidity is typically due to absorption of boiled-off,  $H_2S$ -bearing steam by standing water. However, magmatic condensate, produced by direct input of SO<sub>2</sub>, HCl, and H<sub>2</sub>S from underlying, high-level magma chambers is dominant in the HS environment, most notably in hyperacidic (pH $\leq$ 1) crater lakes, such as those perched high on the Ruapehu, North Island, New Zealand, Poás, Costa Rica, and Kawah Ijen composite volcanoes (e.g., Rowe et al. 1992; Christenson and Wood 1993; Delmelle and Bernard 1994), as well as some lakes within volcanic dome complexes. Hyperacidic crater lakes are considered to overlie potential sites of HS epithermal ore deposition (Hedenquist 1995).

Sediments suspended in the water columns and accumulating on the floors of these acidic lakes and mud pots are dominated by fine-grained opal-CT and opal-C (cristobalite) along with one or more of gypsum, barite, anatase, native sulfur, and pyrite: the ultimate products of rock dissolution by hot, lowpH fluids (Brantley et al. 1987; Delmelle and Bernard 1994; Hedenquist and Taran 2013). The so-called silica residue, described by Rodgers et al. (2002, 2004), is a similar, smallvolume product that accumulates in acidic pools and mud pots in the LS geothermal systems of the Taupo Volcanic Zone, North Island, New Zealand, Alunite and kaolinite can also accumulate as sediments in somewhat less-acidic lakes and mud pots. The amorphous silica-dominated sediments are typically finely laminated (Fig. 5a) and can display soft-sediment deformation features, some caused by material sloughed from steep and unstable lake margins. Native sulfur occurs as molten accumulations on the floors of some acidic lakes and is present locally as interbeds and impregnations in the chemical sediments (Kusakabe et al. 1986; Delmelle and Bernard 1994; Fig. 5a).

These amorphous silica sediments have not been widely recognized in epithermal deposits, and crater lake sediments appear to be unknown. Nonetheless, several remanent patches of chemical sediments of this type abut the advanced argillicaltered dacitic domes at the Yanacocha gold deposit (JW Hedenquist, written commun, 2015) and elsewhere, although in some places they may be difficult to distinguish from lacustrine sediments that have been subjected to intense steamheated alteration and, hence, are dominated by opaline silica.

# Hydrothermal eruption craters and breccias

Hydrothermal eruption craters are most readily recognized by the localized presence of lacustrine mudstones and/or the crudely bedded breccia aprons that surround them. The breccias are commonly polymictic, poorly sorted, angular to subangular, and matrix supported (Fig. 5c); can contain clasts of surface and subsurface hydrothermal products, such as



**Fig. 5** Images of lacustrine amorphous silica sediments and hydrothermal eruption breccias. **a** Finely laminated, lacustrine amorphous silica sediments in the breached crater lake at Jogoku-dani in the Tateyama volcano, Honshu, Japan (see Kusakabe et al. 1986). Yellow coloration of certain layers is due to native sulfur impregnation. **b** Small hydrothermal eruption crater filled with boiling water and surrounded by eruption breccia composed of black mudstone clasts, Osorezan, Honshu, Japan (see Aoki 1992). Note the yellow color of the water due to amorphous arsenic sulfide flocculent. **c** Crudely bedded, polymictic, hydrothermal eruption breccia above the Pascua-Lama HS disseminated gold-silver deposit, Chile

sinter and epithermal vein material in IS and LS systems; and lack juvenile magmatic components. The craters themselves can range in size from several (Fig. 5b) to hundreds of meters and are generally filled by laminated sediments, ranging in composition from amorphous silica in hot, acidic lakes to mudstones in cold lakes. The largest of ten Holocene craters beneath and alongside Lake Rotokawa in the Taupo Volcanic Zone is ~1.5 km in diameter (Krupp and Seward 1987). In the case of large eruptions, the breccia aprons can extend for up to ~2 km outboard of the craters, and sub-crater brecciation can extend downward for several hundred meters (Collar and Browne 1985; Hedenquist and Henley 1985). Nonetheless, distinction between amorphous silica sediments filling a small hydrothermal eruption crater that has lost its breccia apron to erosion from those deposited in a mud pot or other restricted depression is nigh impossible.

Hydrothermal eruptions are commonly caused by rapid subsurface buildups of fluid pressure, transferred from depth via a non-condensable gas phase (Hedenquist and Henley 1985), sometimes beneath impermeable barriers (Fournier 1983, 1985; Hedenquist and Henley 1985; Nelson and Giles 1985) but commonly without such impediments (Browne and Lawless 2001). However, external triggers such as faulting, lake drainage, seismic activity, and/or magma intrusion contributing CO<sub>2</sub>, as in the case of the Waiotapu geothermal system in the Taupo Volcanic Zone (Nairn et al. 2005), can also play important roles.

A series of Holocene hydrothermal eruption craters formed under shallow-water conditions near the shore of the caldera lake at Osorezan, Honshu, Japan, with one of them still filled by H<sub>2</sub>S-rich, neutral-pH, boiling chloride water that is actively depositing sulfur, orpiment, and gold (Aoki and Thompson 1990; Izawa and Aoki 1991; Aoki 1992; Fig. 5b). A sub-lacustrine, bedded native sulfur deposit formed in the largest hydrothermal eruption crater at Lake Rotokawa as a result of bacteriogenic oxidation of H<sub>2</sub>S (Krupp and Seward 1987). Hydrothermal eruption craters and associated breccia remnants mark epithermal paleosurfaces at several localities, and the breccias are particularly widespread at the Hycroft LS (Ebert and Rye 1997) and Pascua-Lama HS deposits (Fig. 5c). The breccias may be interbedded with sinter at LS deposits, as at McLaughlin, California (Lehrman 1986), and Milestone in the Silver City district, Oregon (Rytuba and Vander Meulen 1991b). The same situation is represented by the siliceous breccia capping Bodie Bluff in the Bodie LS gold district, California, and the breccia above the Golden Promise LS gold vein in the Republic district, Washington, which both contain sinter and LS vein fragments (Silberman and Chesterman 1991; Fifarek et al. 1996). The hydrothermal eruption crater that once overlay much of the blind Favona LS vein gold deposit in the Waihi district, North Island, New Zealand, was at least 800-m long and 300-m wide, judging by the size of the underlying vent breccia, which also contains both sinter and vein clasts (Torckler et al. 2006; Fig. 6). The localized, inward-thickening mudstone sequence that overlies the Yamada LS vein system at Hishikari, Kyushu, Japan, is also interpreted as a hydrothermal eruption crater, although any associated breccia deposits have been eroded (Izawa et al. 1993). A similar situation is described from the Broken Hills LS gold-silver deposit, North Island, New Zealand (Rabone 2006).

#### Hot spring sinter

Silica sinter forms aprons or terraces where hot ( $\leq 100$  °C), ascendant, near-neutral-pH, amorphous silica-saturated fluids



**Fig. 6** Section showing the spatial relationship between the top of the geologically blind LS epithermal veins and hydrothermal breccia filling the vent below a now-eroded eruption crater at the Favona LS gold-silver deposit, Waihi district, North Island, New Zealand (taken from Mauk et al. 2006)

discharge at the surface (Fig. 3), a situation that is only possible where groundwater tables intersect the surface as, for example, along lake shores or river banks; hence, the widespread association of sinter with lacustrine and fluvial sedimentary strata (e.g., Rytuba and Vander Meulen 1991a; Yahata et al. 1994; Wallace et al. 2008). Sinter also forms mounds around geyser vents, which can be marked by chimney-like silica constructions attaining heights of up to several meters (Fig. 7a). Sinter and steam-heated alteration can occur nearby one another and may be considered as lateral equivalents (Fig. 3). Amorphous silica, opal-A, is precipitated as the hot water flows across the discharge aprons or down the terraces, with distinctive pool sinter forming where the water ponds (Fig. 7b). Microbial activity besides cooling and evaporation plays a prominent role in sinter formation (e.g., Jones et al. 2001). The presence of amorphous silica sinter indicates subsurface reservoir temperatures in excess of ~200 °C, assuming the liquid is saturated with quartz (White 1973; Fournier 1985).

The non-crystalline opal-A transforms through poorly crystalline opal-CT and opal-C to moganite (monoclinic silica) and then microcrystalline quartz (chalcedony) in ~50, 000 years, with a concomitant increase in density, decrease in both water content and porosity, and destruction of microtextures (White et al. 1988; Herdianita et al. 2000; Rodgers et al. 2004; Lynne et al. 2007). Hence, epithermal sinters, typically Pliocene or older, are predominantly composed of chalcedony. The aging process appears to be accelerated by the presence of organic and other impurities, elevated depositional and post-depositional temperatures, and postdepositional burial (Herdianita et al. 2000).

Most sinter accumulations are associated with LS epithermal deposits because it is the epithermal type that forms

Fig. 7 Images of hot spring sinter and travertine. a Geyser vent mound at the southeastern extremity of the Cerro Negro vein system, Deseado massif, Patagonia, Argentina. b Pool sinter displaying syneresis cracks, Milos island, Greece. Image is ~40 cm across. c Geyserite from Orakeikorako geothermal system, Taupo Volcanic Zone, North Island, New Zealand. Image is ~20 cm across. d Chalcedonic sinter with bedding-parallel cavities (bubble-mat texture), Verbena, Drummond Basin, Queensland, Australia. e Silicified filamentous bacterial streamer fabric on bedding plane of chalcedonic sinter indicative of direction of water flow over sinter apron, Verbena, Drummond Basin, Queensland, Australia. f Travertine, showing layering and manganese and iron staining, Wau IS gold district, Papua New Guinea. Image is ~40 cm across



in rift settings where outcropping groundwater tables are more likely (Sillitoe and Hedenquist 2003). However, some IS epithermal systems generated in pull-apart rift basins also include sinter, as exemplified by Fruta del Norte, Ecuador (Leary et al. 2007). In contrast, there are no well-authenticated examples of sinter associated with HS epithermal deposits, probably for the simple reason that it does not form because silica polymerization and precipitation are inhibited in acidic (pH <3–4) fluids, typical of HS systems (Fournier 1985). Nonetheless, as described above, amorphous silica does form colloids that accumulate as finely laminated sediments in acidic lakes.

Sinter recrystallized to chalcedony is normally distinguishable by its non-planar, finely laminated or banded texture (Fig. 7d) and multi-colored (white, yellow, green, tan, redbrown, gray, and/or black) character, typically ascribed to iron, manganese, and mercury impurities. Also diagnostic is the presence of columnar microbial structures perpendicular to the laminae (coniform texture); irregular, lenticular cavities parallel to the laminae (bubble-mat texture; Fig. 7d); nodular and other forms of geyserite formed in evaporative splash zones of geysers (Fig. 7c); filamentous microbial forms occurring as bedding-parallel streamers indicative of water flow direction (Fig. 7e); oncoids that grow in turbulent pools; and/ or plant root, stem, and leaf molds in both growth and flowaligned positions; as well as other textures (e.g., Lynne 2012). Nonetheless, in float or drill core and where rock outcrop is poor, sinter that is recrystallized to chalcedony and underwent additional silica introduction can sometimes be tricky to tell apart from colloform-banded LS epithermal vein chalcedony.

In reasonably well-preserved sinter occurrences, a series of depositional environments may be defined, from proximal, high-temperature (>70 °C), near-vent zones typified by various types of abiotic geyserite, through mid-slope aprons, terraces, and pools characterized by stromatolitic growth and thermophilic bacterial mats, to distal, cool (<35 °C), and plant-rich wetland (marsh) facies (e.g., Walter et al. 1996; Lynne 2012; Channing and Edwards 2013; Fig. 3). Syn- or post-depositional erosion of sinter aprons and terraces results in clastic facies (e.g., Steamboat Springs, Nevada; White et al. 1964). Sinter can commonly be interbedded with fluvial, lacustrine, or volcaniclastic material, which itself may undergo complete or partial silicification (e.g., White et al. 1964; Campbell et al. 2003). A variety of biota, including stromatolites, filamentous microbes, diatoms, and plants, has been

described in detail from several sinters, ranging in age from Paleozoic (Trewin 1993; Walter et al. 1998) through Jurassic (Guido et al. 2010) to Recent (Jones et al. 2001), although their preservation potential tends to decrease progressively with increasing age (de Wet and Davis 2010).

Several LS epithermal gold-silver deposits occur beneath or alongside contemporaneous siliceous sinter remnants, including steep vein systems at National and Fire Creek, Nevada (Vikre 1987; Milliard et al. 2015), McLaughlin (Lehrman 1986), Cerro Blanco, Guatemala (White et al. 2010), and El Dorado, El Salvador (Richer et al. 2009); disseminated mineralization and deeper bonanza-grade veins at Ivanhoe, Nevada (Wallace 2003); and, as far as is known, disseminated mineralization alone at Hasbrouck Mountain and Buckhorn, Nevada (Graney 1987; Jennings 1991). At all these localities, the sinter is underlain by and/or interbedded with lacustrine sedimentary and volcaniclastic rocks that were also partially silicified and subjected locally to steam-heated alteration. The major Cerro Vanguardia LS vein gold deposit in the Late Jurassic Deseado massif of Patagonia, Argentina, was also likely generated beneath a fluvio-lacustrine setting and can be traced along strike into extensive areas of preserved paleosurface features, including proximal-to-distal sinter accumulations (Guido and Campbell 2014).

# Hot spring travertine

Travertine aprons or terraces are constructed as a result of  $CO_2$  degassing where relatively low-temperature,  $CO_2$ -rich waters attain the surface. As in the case of sinter, this is typically where the groundwater table intersects the surface. However, since sinter and travertine are the products of completely different chemical processes, as discussed earlier, they are not normally closely associated in epithermal systems (Fig. 3); however, modern-day exceptions are known (Campbell et al. 2002; Smith et al. 2011). Travertines are commonly fed via extensional faults and fractures marked by fissure-ridge deposits: outwardly dipping, bedded travertine flanking fissures filled by vertically banded travertine; however, tufa cones form where the controlling fissures underlie unconsolidated sediments, as on lake floors (Hancock et al. 1999).

Travertine is characterized by a variety of facies, ranging from micritic to coarsely crystalline and from hard and dense (Fig. 7f) to porous and friable. The facies and mineralogic composition (calcite and/or aragonite) are controlled by water composition and temperature, CO<sub>2</sub> saturation and degassing rate, biotic activity, and presence of growth inhibitors (e.g., Mg/Ca ratio; Jones and Renaut 2010). Well-banded travertine can sometimes prove difficult to distinguish from epithermal carbonate vein material (Fig. 7f). However, travertine facies can contain distinctive carbonate fabrics, ranging from abiotic, shrub-like crystal forms at proximal sites to microbially influenced more distally, where stromatolites, filamentous microbial mats, and plant remains are diagnostic (e.g., Renaut et al. 2002). Travertine also commonly contains or is interbedded with iron and manganese oxides (Fig. 7f).

Travertines typically lack a close spatial relationship—and in many cases, no relationship at all—to epithermal precious metal deposits, but can occur up to several kilometers distant where they mark outflow sites of relatively cool, CO<sub>2</sub>-rich waters. An example is provided by the tufa mounds (commonly referred to as travertine) within the moat sediments of the Creede ash-flow caldera, Colorado, which may mark the distal outflow of the Creede IS vein silver deposit (Bethke and Hay 2000). The topographically low, southeastern part of the Cerro Negro LS vein system in the Deseado massif is dominated by travertine, although there is widespread hydrothermal silicification of fissure-ridge/mound and apron travertines and stromatolitic mounds formed in the transition between subaerial and sub-lacustrine settings (Guido and Campbell 2012).

## Hydrothermal chert

Chert can form where hydrothermal fluids undergo cooling as they debouch on the floors of shallow lakes, with active examples reported from the East African rift valley of Kenya (Renaut and Owen 1988). Small siliceous chimneys or spires, with or without syngenetic chert horizons (described as sublacustrine sinter), also formed above hot spring vents on the floors of Lake Taupo in the Taupo Volcanic Zone and Yellowstone Lake, Wyoming (de Ronde et al. 2002; Shanks et al. 2007). In principle, transitions can occur between hydrothermal chert deposits and the distal wetland facies of sinter aprons (Channing and Edwards 2013).

Hydrothermal chert, originally deposited as silica gel, can be of high purity where clastic input during its accumulation was minor, and at Stará Kremnička, Slovakia (Fig. 6b), for example, it was exploited as a raw material for manufacture of ferrosilicon alloys. There, the chert is believed to have precipitated from the distal, lower-elevation outflow from the Kremnica LS vein gold deposit, 8 km to the north (Lexa and Bartalský 1999). The chert contains fossilized marsh plantssome in growth positions, snails, frogs, and turtles, testifying to extremely shallow- and cool-water depositional conditions (Lexa and Bartalský 1999). Thinly laminated, pyritic chert also forms thin beds in the inferred hydrothermal eruption crater above the Yamada LS gold veins at Hishikari (Izawa et al. 1993) and similar material, displaying soft-sediment deformation features, occurs at the Rawhide low-grade, bulktonnage LS gold deposit, Nevada (Black et al. 1991).

# Silicified lacustrine sediments

In rift environments conducive to LS epithermal deposit formation, silicification can be a subaqueous, syn- to diagenetic process that accompanies lacustrine sedimentation. As a result, lacustrine sediments are wholly or partly transformed to opaline silica, which eventually is recrystallized to chalcedony. The textures of the sediments, particularly fine lamination, and any contained biogenic material are commonly well preserved.

Unequivocal distinction of some sub-lacustrine, subaerial, and subsurface silicic deposits is not straightforward, even in the case of recent examples, and becomes even more difficult as a result of aging and consequent diagenetic changes (Jones et al. 2007). Silicified fine-grained lacustrine sediments and hydrothermal chert, which can be closely associated (Fig. 3), are both typically laminated, causing them to be commonly confused with one another and, potentially, also with lacustrine amorphous silica sediments. The presence of biogenic material, including plant molds, in both limnic chert and silicified sediment can further complicate their ready distinction and even their misidentification as sinter. However, silicified sediments typically have higher residual titanium contents than either limnic chert or sinter, although this compositional difference does not help to distinguish subaqueous from strata-bound subsurface silicification, which can only be accomplished on the basis of geologic context.

Silicified lacustrine sediments tend to be more widespread and abundant than either limnic chert or sinter in volcanosedimentary rift settings, as exemplified by the Late Jurassic rift basins of the Deseado massif. The up to ~50-m-thick, strata-bound silicification in the Ivanhoe gold district took place during lacustrine sedimentation, albeit interrupted by subaerial interludes characterized by sinter deposition (Wallace 2003). However, the silicified horizon was originally interpreted as a paleogroundwater table feature, implying formation after the lake had disappeared (Bartlett et al. 1991).

# Paleosurface metal contents

In contrast to the precious and base metals, which are largely immobile in a vapor at low temperatures, mercury can be transported effectively (White 1981; Barnes and Seward 1997; Christenson and Mroczek 2003), leading to its precipitation as cinnabar in the steam-heated environment. Cinnabar can also be deposited in sinters and associated opalite, either along with stibnite and other minerals from the liquid phase (e.g., Steamboat Springs, Ivanhoe, Buckskin Mountain in the National district, and Puhipuhi, North Island, New Zealand; White et al. 1964; Wallace 2003; Hampton et al. 2004; Vikre 2007) or from gases once the groundwater table has fallen. Less commonly, cinnabar impregnates mudstone laminae in hydrothermal eruption craters, as observed at Hishikari (Izawa et al. 1993) and now actively depositing at the surface of the Ngawha geothermal system, North Island, New Zealand (Ellis and Mahon 1966; Weissberg et al. 1979; Davey and van Moort 1986; Barnes and Seward 1997).

Cinnabar was formerly mined at Ngawha, from some steam-heated zones in both HS (e.g., Paradise Peak, Nevada; Sillitoe and Lorson 1994) and LS (e.g., Steamboat Springs; White et al. 1964) districts and from sinters and associated silicified rocks linked to LS deposits (e.g., McLaughlin, National, Ivanhoe, and Goldbanks districts and Puhipuhi (Lehrman 1986; White 1986; Vikre 1987, 2007; Bartlett et al. 1991; Stone et al. 2000; Fig. 8a). However, by far the largest deposits, containing corderoite (Hg<sub>3</sub>S<sub>2</sub>Cl<sub>2</sub>) and other chloride-bearing species besides cinnabar, are hosted by opalite horizons in moat sediments of the McDermitt caldera complex, Nevada and Oregon (Rytuba and Glanzman 1979).

Although steam-heated alteration, groundwater table silicification, lacustrine amorphous silica sediments, sinter, and travertine are commonly devoid of appreciable precious- and base metal concentrations (e.g., White and Heropoulos 1983), at least at the time of initial formation, high precious metal and metalloid contents have been recorded. For example, sinters alongside the widely studied Champagne Pool, a Holocene hydrothermal eruption crater at Waiotapu in the Taupo Volcanic Zone, contain gold (up to 540 g/t) and silver (up to 745 g/t) along with thallium and mercury adsorbed onto orange-colored, amorphous arsenic and antimony sulfide precipitates (Weissberg 1969; Hedenquist and Henley 1985; Hedenquist 1986; Jones et al. 2001; Pope et al. 2005). Siliceous muds associated with a small sinter terrace along the shore of Lake Rotokawa, also in the Taupo Volcanic Zone, contain amorphous arsenic and antimony sulfides, containing elevated Au, Ag, Tl, Hg, Ga, and W; gold averages ~1 g/t (Weissberg 1969; Krupp and Seward 1987). Similar arsenic and antimony precipitates, containing up to 6500 g/t Au, some of it in telluride form, along with late-stage cinnabar occur in mudstone filling the hydrothermal eruption craters at Osorezan (Aoki and Thompson 1990; Izawa and Aoki 1991; Fig. 5b). The geochemistry of paleogroundwater table silicification is poorly characterized, although arsenic and mercury are weakly anomalous in such material at Furtei (Fig. 4d).

Sub-lacustrine hydrothermal chert is commonly anomalous in a variety of elements but not the precious metals: As, Sb, and Hg at Stará Kremnička (Lexa and Bartalský 1999) and As, Cs, Hg, Mo, Sb, Tl, and W at Yellowstone Lake (Shanks et al. 2007). Nonetheless, silicified lacustrine sediments can contain strata-bound, disseminated, precious metal mineralization, as exemplified by the Hollister LS gold-silver deposit at Ivanhoe (Bartlett et al. 1991) and San Cristóbal IS silver-zinc-lead deposit, Bolivia (Buchanan 2000; Phillipson and Romberger 2004), although many such precious metal occurrences are low-grade and remain subeconomic (e.g., Rytuba and Vander Meulen 1991a; Stone et al. 2000). Part of this sediment-hosted mineralization at San Cristóbal could be truly exhalative in origin (Buchanan 2000; Phillipson and Romberger 2004), raising the possibility of transitions between sub-lacustrine epithermal and shallow-marine

Fig. 8 Images of metal and metalloid concentrations in paleosurface products and steamheated overprints. a Old Buckskin Mountain mercury workings in silicified epiclastic sediments and sinter, National LS gold district, Nevada (see Vikre 2007). b Goldbearing LS vein in lacustrine mudstone overprinted by steamheated alteration, Seta, Hokkaido, Japan. Note the porous nature of the vein quartz, resulting from carbonate and/or adularia dissolution. c White, powdery opaline silica (left side) developed during the steam-heated overprint of the alteration halo to the Paradise Peak HS gold-silver deposit, Nevada. The partly brecciated, vuggy residual quartz (right side), host to the orebody. was largely unaffected (see Sillitoe and Lorson 1994). d Cinnabar and native sulfur deposited in fractures in vuggy residual quartz during the steamheated alteration overprint of the Paradise Peak HS gold-silver deposit, Nevada



volcanogenic massive sulfide precious metal deposits, the latter exemplified by Eskay Creek, British Columbia (Roth et al. 1999).

Precious metal mineralization can also overprint sinter in LS and IS deposits (e.g., McLaughlin; Sherlock et al. 1995) as a result of upward progradation of the underlying epithermal systems. Alternatively, steam-heated zones can overprint preexisting epithermal mineralization, leaving sufficient precious metals in the LS veins (e.g., Seta, Hokkaido, Japan; Fig. 8b) or siliceous HS ribs (e.g., Paradise Peak and Borealis, Nevada; Sillitoe and Lorson 1994; Eng 1991; Fig. 8c) for them to attain ore grade. At Hycroft, however, precious metals were reportedly stripped during formation of the steam-heated blanket, but reprecipitated in its structurally controlled roots below the then-deeper groundwater table (Ebert and Rye 1997), presumably because admixture of descendant acid-sulfate and ascendant alkali chloride waters can be an effective goldprecipitation mechanism (Reed and Spycher 1985).

Manganese and iron oxides associated with travertine are capable of scavenging a variety of metals but generally to subeconomic levels. However, a manganese oxide horizon beneath travertine at Golconda hot springs, Nevada, was formerly exploited for its tungsten content (Hewett et al. 1963). Unusually, borates (ulexite and borax) are present in some travertine deposits on the Puna plateau of northwestern Argentina (Sillitoe 1975; Alonso and Viramonte 1990).

# Effects of paleosurface modification

# Volcanic and geomorphologic settings

Modification of epithermal paleosurfaces and, as a direct consequence, the positions of associated paleogroundwater tables, during epithermal deposit formation can be greatly influenced by the regional-scale climatic and tectonic evolution as well as by local volcanic, structural, and geomorphologic events (Sillitoe 1993; Simmons 2002; Simmons et al. 2005).

Aggradation and progressive groundwater table ascent commonly result from the accumulation of volcanic and sedimentary material in regional- or local-scale depocenters (Fig. 9a). In contrast, erosional degradation and related groundwater table depression tend to be more common during tectonically induced uplift and attendant erosion and valley incision (Fig. 9b), climate change causing aridification (e.g., Ebert and Rye 1997), and on topographically prominent volcanic edifices. However, groundwater table descent can also be triggered almost instantaneously by volcano-related events,



Fig. 9 Schematic sections showing effects of syn-hydrothermal aggradation and degradation in epithermal systems. **a** Burial of sinter by siliciclastic sediments and/or volcaniclastic rocks caused paleogroundwater table (PGWT) ascent from positions 1 to 2, which may have contributed to bonanza-grade development and, potentially,

such as caldera formation and sector collapse, and by catastrophic drainage of lake-filled craters, calderas, and other topographic depressions (Simmons 2002; Simmons et al. 2005).

#### Aggradation effects

Syn-hydrothermal paleosurface aggradation can take place by accumulation of volcanic and/or sedimentary material of either local or distant provenance. Volcanic rocks can bury active epithermal systems in any volcanic setting, whereas sedimentation is most likely in the topographic depressions that characterize the extensional LS epithermal environment, but can also occur in dome complexes, ash-flow calderas, and maar volcanoes, especially under relatively pluvial climatic conditions.

Syn-hydrothermal aggradation may be relatively commonplace in the epithermal environment, but is difficult to document unambiguously unless two or more stacked sinter horizons are present, as at Hasbrouck Mountain (Graney 1987) and Verbena in the Drummond Basin, Queensland, Australia (Cunneen and Sillitoe 1989). This difficulty stems from the fact that few epithermal deposits have been discovered beneath largely barren volcanic or sedimentary cover. Even where such cover exists, it needs to be shown that its deposition was synchronous with at least the late stages of hydrothermal activity rather than accumulating later after the epithermal deposit was fully formed.

The Fruta del Norte IS gold-silver deposit provides a welldocumented example of siliciclastic sedimentation over the top of a still-active epithermal system in a pull-apart basin setting (Leary et al. 2007; Fig. 10). The Fruta del Norte vein-stockwork was generated beneath a paleosurface defined by a >1-km-long sinter, which was progressively buried beneath >200 m of conglomerate and subordinate interbedded sandstone and felsic tuff (Leary et al. 2007). Confirmation that the sedimentation overlapped temporally with at least the



veins

b

Overprinted

steam-heated zone above PGWT (2)

eventual demise of the epithermal system. **b** Valley incision caused descent of the paleogroundwater table from positions 1 to 2, thereby subjecting the sinter and rocks above paleogroundwater table 2 to steam-heated conditions; however, the sinter was little affected because of its resistant, silicic character

waning stages of hydrothermal activity is provided by the presence throughout the entire conglomerate sequence of a steep, linear body of chalcedony replacement, containing



Fig. 10 Geologic section of the Fruta del Norte IS gold-silver deposit, Ecuador, showing the locations of the vein-stockwork system, overlying sinter, and silicification of the syn-hydrothermal sedimentary cover rocks (simplified from Leary et al. 2007 and unpublished data)

pyrite and marcasite, anomalous levels of arsenic and antimony, and low-order gold values (Leary et al. 2007; Fig. 10). The 40-m-thick ignimbrite flow that overlies the interpreted hydrothermal eruption crater above the Yamada veins at Hishikari is smectite altered and could also have been deposited during at least the waning stages of the epithermal system (Ibaraki and Suzuki 1993; Izawa et al. 1993).

Syn-formational burial of epithermal systems, as documented at Fruta del Norte and possibly also Hishikari, could potentially promote two competing effects. First, progressive accumulation of the overlying, water-saturated sedimentary pile could act as a partial seal, thereby favoring hydrothermal fluid entrapment and consequent precious metal grade enhancement. Second, it might eventually smother the hydrothermal system and contribute to its demise. Elsewhere, however, fine-grained fluvial and lacustrine sediments accumulating on top of epithermal systems can act as permeable hosts to low-grade epithermal deposits, as exemplified by the Hollister gold-silver orebody at Ivanhoe (Bartlett et al. 1991) and San Cristóbal silver-zinc-lead deposit.

# **Degradation effects**

Fault-related uplift or valley incision within and nearby epithermal systems results in lowering of base level and drop of the groundwater table within the upper parts of the systems. As a result, steam-heated zones, generated above groundwater tables, can overprint previously formed paleosurface features, such as sinter and paleogroundwater table silicification as well as underlying mineralized and unmineralized rocks (Fig. 9b). Normal faulting and progressive footwall uplift caused overprinting of quartz-adularia-illite alteration by steamheated assemblages at the Te Kopia geothermal system, Taupo Volcanic Zone (Bignall et al. 2004; Rowland and Simmons 2012). Such superposition of steam-heated alteration has also been documented at the Hycroft (Wallace 1987; Ebert and Rye 1997), Buckskin Mountain (Vikre 2007), and Florida Canyon (Fifarek et al. 2010) LS gold deposits and Arcata and Fresnillo IS silver deposits (Candiotti et al. 1990; Simmons 1991). At Arcata, the kaolinite-rich, steam-heated alteration overprinted the adularia-rich halo to the Tres Reyes vein (Fig. 4c) as well as causing dissolution of the carbonate gangue in shallow parts of the vein, leaving abundant open cavities. Although only steam-heated remnants are now present, fluid inclusion geobarometry documents synhydrothermal groundwater table descent of ~400 m at Fresnillo (Simmons 1991), substantially more than the 150 m at Florida Canyon (Fifarek et al. 2010) and >18 m at Hycroft (Ebert and Rye 1997). At the Cerro Millo HS prospect, Peru, a stacked series of silicified horizons has been interpreted in terms of at least 200 m of syn-hydrothermal paleogroundwater table descent (Hennig et al. 2008).

The siliceous sinter and associated silicified epiclastic deposits atop Buckskin Mountain remained largely unaffected by the paleogroundwater table descent, except for the probable introduction of cinnabar (Fig. 8a), whereas the immediately underlying rocks were transformed to steam-heated alteration assemblages (Vikre 1985, 2007). Steam-heated alteration can also overprint precious metal mineralization, hosted by either HS structures or IS and LS veins. Thus, at Paradise Peak, the steam-heated overprint introduced mercury and native sulfur to the hydrothermally brecciated, vuggy residual quartz body, which constituted the gold-silver orebody (Fig. 8d), while transforming the already-altered, flanking rocks to powdery opal-CT and opal-C (Sillitoe and Lorson 1994; Fig. 8c). Similar superposition of steam-heated alteration over auriferous siliceous ledges also appears to have taken place in parts of the Borealis HS district (e.g., Eng 1991). At the Seta prospect, an LS system associated with sinter and lacustrine sediments (Yahata et al. 1994; Yajima et al. 1997), the steam-heated alteration overprinted the shallowest quartz veins, causing them to become porous and friable (Fig. 8b).

More severe degradation of hydrothermal systems can be induced by flank or sector collapse of composite volcanoes, during which gravitational removal of up to several cubic kilometers of volcanic rock can take place in a geologic instant (Sillitoe 1994). The gravitational instability leading to flank or sector collapse commonly results from hydrothermal alteration and associated elevated fluid pressures in the summit regions and cores of the volcanoes (López and Williams 1993; Cecchi et al. 2005; Delmelle et al. 2015), but can be assisted by other processes, including volcanic and seismic activity and faulting. The hydrothermal products, including near-surface, steam-heated as well as deeper, advanced argillic assemblages that induced the edifice weakening can be observed in the resulting debris avalanche deposits (e.g., John et al. 2008). Sector collapse, massive hydrologic perturbation (e.g., volcanic eruption; Simmons et al. 2005), and other causes of abrupt groundwater table descent can cause reductions in confining pressure sufficient to produce hydrothermal brecciation and eruption as well as underlying precious metal mineralization. For example, sector collapse of a relatively small trachyandesitic stratovolcano of Pleistocene age on Lihir Island appears to have transformed incipiently developed porphyry copper-gold mineralization at depth in the volcanic edifice into the giant, near-surface, disseminated Ladolam LS gold deposit as a direct result of the decompression and consequent brecciation and fluid expulsion (Moyle et al. 1990; Sillitoe 1994; Carman 2003). Catastrophic drainage of Pleistocene glacial lakes at Yellowstone lowered the paleogroundwater table, triggering formation of at least ten hydrothermal eruption craters up to 450 m across (Muffler et al. 1971). However, as pointed out by Simmons (2002), it remains to be demonstrated that erosion by either valley

incision or pedimentation is sufficiently rapid to trigger precious metal deposition, as claimed by Bissig et al. (2002, 2015) for central Andean HS gold-silver deposits.

# **Preservation potential**

# Age control

Most paleosurface features, like epithermal deposits in general (e.g., Kesler and Wilkinson 2009), tend to be relatively young, commonly Miocene or more recent, because of their susceptibility to erosional removal. Where Precambrian, Paleozoic, or Mesozoic paleosurface features are preserved, they are parts of LS or IS epithermal systems that formed beneath topographically low, extensional environments. The Devonian sinters of the Drummond Basin (Cunneen and Sillitoe 1989; White et al. 1989) and Late Jurassic sinters, travertines, and silicified lake beds of the Deseado massif (Guido and Campbell 2011), both extensional back-arc regions, are classic cases in point. Notwithstanding the existence of HS deposits of Proterozoic and Paleozoic age (e.g., Hope Brook, Newfoundland, Canada and Temora, New South Wales, Australia; Thompson et al. 1986; Dubé et al. 1998), pre-Miocene steam-heated zones in this environment are unreported because of their ease of erosional removal from volcanic centers.

### **Climatic control**

As a generalization, arid and semi-arid regions of the world display much better preserved paleosurface features than those with pluvial climatic regimes, which explains why many of the examples of steam-heated alteration and sinter in association with epithermal deposits occur in the Great Basin of Nevada, central Andes, and Deseado massif of Patagonia. Indeed, most of the large, shallow-level HS epithermal deposits appear to be confined to the central Andean region. Paleosurface features normally only survive in Southeast Asia and the southwestern Pacific region if they are exceptionally young, as in the case of the <1-Ma and still hydro-thermally active Ladolam deposit (Moyle et al. 1990; Carman 2003).

# **Compositional control**

Many landforms on epithermal paleosurfaces are composed of soft, friable materials, as exemplified by steam-heated zones, hydrothermal eruption craters and breccias, and fossil hyperacidic lakes and mud pots. Hence, these products, as well as any associated intermediate argillic (illite-smectite) alteration zones are rapidly eroded, particularly in the pluvial climatic regimes characteristic of tropical regions, and are generally poorly preserved even in relatively young epithermal systems unless they underwent either silicification or prompt burial. For example, hydrothermal eruption craters <1800 years old in the Taupo Volcanic Zone lack topographic expression and are recognized exclusively on the basis of their distinctive and relatively restricted breccia aprons (Browne and Lawless 2001). The ephemerality of hyperacidic crater lake sediments in the HS environment may be attributed not only to their low erosional resistance but also to the extremely poor preservation potential of volcano summit regions in general.

In contrast, sinter, paleogroundwater table silicification, sub-lacustrine chert, and silicified lacustrine sediments offer much greater erosional resistance and commonly give rise to impressive examples of topographic inversion. Although formed in topographic lows, sinter, sub-lacustrine chert, and silicified lacustrine sediments can constitute topographic prominences. Arguably, the most impressive example is the late Miocene sinter and interbedded epiclastic deposits capping Buckskin Mountain, which accumulated in an intermontane basin but now cap the highest point in northern Nevada (Vikre 1987, 2007). Similarly, the Hasbrouck Mountain sinter underpins the eponymous isolated hill (Graney 1987) and the Ikiryu sinter in Kyushu, Japan, occupies a prominent, ridgetop position (Nakanishi et al. 2001). Even the erosional resistance of Holocene sinters can give rise to elevated topography (e.g., Tuhunaatara sinter, Taupo Volcanic Zone; Campbell et al. 2003). Nonetheless, many proximal sinters are largely represented by boulder fields and thinner, distal wetland facies are commonly completely destroyed (Channing and Edwards 2013). Use of the term "silica cap" for such prominent silicic features is discouraged because it fails to adequately distinguish the different silicic paleosurface landforms from one another and from deeper, strata-bound silicification.

# **Burial control**

The simplest means of preserving paleosurface features is to quickly bury them beneath post-mineralization volcanic or sedimentary cover. Rapid syn-hydrothermal burial by sedimentary rocks accounts for the complete preservation of the Fruta del Norte IS deposit (Leary et al. 2007; see above). However, most epithermal deposits underwent at least minor erosion before concealment beneath post-mineralization units, resulting in few known examples of paleosurface features that were definitively preserved by post-mineralization volcanism or sedimentation. Arguably, the best example is provided by the Late Jurassic LS epithermal systems of the Deseado massif, which were buried and preserved beneath Cretaceous and Cenozoic continental and marine passive-margin successions (Giacosa et al. 2010). This situation is well displayed by the Cerro Negro LS vein district where any paleosurface features of hydrothermal origin were eroded prior to concealment of the shallow-level, western part of the Eureka West ore shoot beneath volcano-sedimentary rocks (Shatwell et al. 2011). Post-mineralization displacement on the normal fault that localized the Eureka West vein may have instigated the required erosion. The resultant fault scarp is flanked by talus-like breccia, containing abundant quartz vein clasts, which largely covered the outcropping vein prior to deposition of the overlying volcano-sedimentary strata.

# **Exploration implications**

## Identification of paleosurface products

The most obvious requirement for proper appreciation and, hence, successful exploration of little-eroded epithermal prospects is correct recognition of the paleosurface features present and their distinction from superficially similar, but either more deeply formed or supergene weathering products.

Steam-heated alteration needs to be distinguished from potentially ore-bearing, advanced argillic lithocap assemblages formed beneath paleogroundwater tables in HS epithermal systems (Sillitoe 1995a), which should be relatively straightforward given their distinctive textural features. The low density, presence of powdery opaline silica and alunite, and absence of residual vuggy quartz and coarsely crystalline alunite are particularly diagnostic (Table 3).

By the same token, it has to be decided if tabular silicic bodies, commonly morphologically prominent because of topographic inversion, represent chalcedonic sinter, groundwater table silicification, lacustrine amorphous silica sediments, sub-lacustrine chert beds, silicification of unconsolidated lacustrine sediments, silicified subsurface sedimentary/volcanic horizons (including jasperoid after carbonate rocks), or shallowly dipping, HS replacement ribs or crustiform IS/LS veins (Table 4; Fig. 11). Supergene sulfide oxidation causes limonite impregnation and staining of many of these silicic bodies, thereby further hindering their discrimination. The typically low precious metal contents of steam-heated alteration, groundwater table silicification, sub-lacustrine chert, and many sinters can be diagnostic, but can also lead to premature abandonment of epithermal IS and LS prospects if their true, near-surface formational sites, commonly above any precious metal concentrations, are not fully appreciated. Many of the silicic paleosurface products are characterized by an important biogenic component, although this is uncommon in the case of lacustrine amorphous silica sediments and paleogroundwater table silicification (Table 4).

Alunite and kaolinite are also widespread minerals in the surficial supergene environment and are best distinguished from their hypogene epithermal counterparts with a hand lens rather than by means of a portable spectrometer. Powdery, steam-heated and porcelanous, supergene alunites are easy

Table 3 Di	stinguishing features of 1	nain alunite types					
Alunite type	Associated epithermal deposit type	Position	Color	Distinctive texture	Style	Relationship to epithermal mineralization	Comments
Steam heated	HS, IS, LS	Vádose zone, above groundwater table	White	Finely crystalline, friable, powdery, low density	Pervasively developed replacement blankets, up to 10-m thick; up to 70 % of rock volume	Barren unless overprinting pre-existing mineralization	Recessive weathering; low preservation potential, especially under pluvial conditions
Deep hypogene	SH	Below groundwater table	Colorless, white, cream, pink, or yellow-brown <sup>a</sup>	Coarse to fine crystal aggregates with quartz as massive replacements; also alone as pseudomorphs of clasts or henocrysts and limito cavities	Steep or bedding-parallel quartz-alunite bodies, up to several 100-m thick, up to 50 % of rock volume	Sub-ore grade but anomalous in Au and Ag where massive; potentially ore-bearing where developed in hydrothermal hreacias	Resistant, commonly forming prominent outcropping ribs
Supergene	HS, IS, locally LS	In oxidized zones above groundwater tables as well as in upper parts of subjacent sulfide zones; forms during active pyrite oxidation	White, cream, yellow- brown <sup>a</sup> , or uncommonly pale green <sup>b</sup>	Massive, fine-grained, porcelaneous veinlets and patches; locally rock replacement	Typically occuries the youngest veinlet generation; up to 5 % of rock volume	In pyritic or formerly pyritic rocks within or beyond ore; uncommon in HS ore	Mainly developed in arid and semi-arid regions with limited groundwater recharge and dilution
<sup>1</sup> Due to impre	gnation by supergene ja	rosite					

Due to minor amounts of copper in solid solution

Table 4 Distinguishing featu	res of flat-lying silicic product	s in epithermal systems				
Silicic product	Associated epithermal deposit type	Areal extent	Distinctive texture(s)	Distinctive feature(s)	Contained biota	Iron sulfide content, vol %
Groundwater table silicification	HS, IS, LS	Potentially extensive (<∞10 km²)	Massive, partly brecciated, with evidence of replaced lithology	Overlain by steam-heated alteration (if preserved); remanent clastic component (nin, TT)	Commonly none	5-10
Amorphous silica sediments	HS, IS, LS	Relatively restricted (<~1 km <sup>2</sup> )	Finely laminated	Soft-sediment and impact features	Commonly none, but diatoms mossible	0-10
Hot spring sinter	IS, LS	Relatively restricted (<~1 km <sup>2</sup> )	Finely laminated; geyserite, bubble-mat and/or coniform	Varied colors, no alteration halo	Abundant: bacteria, stromatolites, plants,	Typically 0
Hydrothermal chert	IS, LS	Potentially extensive	Massive to finely laminated	Lacks appreciable clastic	poucu Abundant: subaqueous (diatome) and subaerial	0-10
Silicified lacustrine sediments	IS, LS	Potentially extensive	Finely laminated	Remanent clastic component	Abundant: subaqueous (diatome) and subaarial	5-20
Silicified subsurface horizon	HS, IS, LS	Potentially extensive	Typically massive, with evidence of replaced	Overlain by clay alteration; remanent clastic	Commonly none	5-20
Strata-bound jasperoid	HS, IS, LS	Relatively restricted	Variable, partly brecciated	Hosted by carbonate rock	Inherited from host limestone if mesent	530
Strata-bound replacement body	SH	Relatively restricted	Massive, texturally varied, associated vuggy residual	Flanked by quartz-alunite and/or quartz-kaolinite	None	1050
Strata-bound vein	IS, LS	Relatively restricted	yeans Crustiform and colloform banding	Flanked by minor quartz- adultaria or quartz-sericite/ illite alteration	None	Ś

<sup>a</sup> Or derivative limonite composed mainly of jarosite and/or hematite



Fig. 11 Cartoon to show different types of flat-lying, surface and shallow-subsurface silicic products possible in IS and LS epithermal systems. Note that all these products can give rise to topographic

to distinguish from one another (Sillitoe 1993; Table 3). Rocks containing either steam-heated or supergene kaolinite are more difficult to tell apart because both, in contrast to hypogene kaolinite formed in the HS environment, normally lack associated groundmass silicification; however, the supergene variety is likely to be strongly limonitic because it is a result of pyrite oxidation. Drain-back kaolinization, formed by descendant, steam-heated waters as epithermal systems waned, also lacks accompanying groundmass silicification because the silica-bearing solutions are being heated rather than cooled.

#### Location of mineralization

The proximal parts of sinter aprons and terraces formed in topographically low areas, particularly where fault control of hydrothermal upflow is evident, can directly overlie LS or IS epithermal mineralization. Textures, in part imposed by the contained biota, can distinguish between near-vent and distal sinter deposits (e.g., Lynne 2012) even in old sinters, such as those of Late Jurassic age in the Deseado massif (Guido and Campbell 2014). Geyserite (Fig. 7c), providing direct evidence for boiling discharge, is a reliable indicator of proximal sinter (White et al. 1964; Campbell et al. 2015). However, where sinters lie on the lower slopes of or peripheral to volcanic edifices, such as stratovolcanoes or dome complexes, they likely formed in relatively high-temperature outflow zones, which can be distant from the potentially mineralized cores of the systems (e.g., Mac-Ban geothermal field, Philippines; Capuno et al. 2010; Fig. 1b).

Hydrothermal eruption craters and underlying breccias generally form in the proximal parts of epithermal systems and, as in the Taupo Volcanic Zone, typically mark the positions of high-flux, fluid-upflow conduits in which epithermal mineralization can be formed (Hedenquist and Henley 1985; Rowland and Simmons 2012): a situation clearly observed at Hishikari and, based on the subsurface hydrothermal vent breccia, reliably inferred at Favona (Izawa et al. 1993; Torckler et al. 2006; Figs. 3 and 6). Nonetheless, as in the case of sinters, more distal locations are also possible. Where hydrothermal eruption craters have been obliterated, recognition of eruption breccias, applying the criteria described above, including the common but by no means ubiquitous presence of at least a few vein and sinter clasts in the IS and LS environment, is critical. Increased breccia thicknesses and

inversion, resulting in their preservation as prominent surface features. Their proper distinction is one of the important keys to exploration of epithermal prospects

concentrations of large clasts can help to locate the likely sites of craters and their underlying upflow conduits (e.g., Collar and Browne 1985; Hedenquist and Henley 1985).

Where travertines are parts of epithermal systems, they are typically poor indicators of precious metal mineralization because of their characteristic distal positions relative to hydrothermal upflow zones. However, care needs to be exercised because at both Cerro Blanco and the southeastern extension of Cerro Vanguardia (Claudia), travertine and sinter are intimately related in proximity to auriferous LS veins (White et al. 2010; Guido and Campbell 2014).

Steam-heated zones and any underlying paleogroundwater table silicification can be areally extensive, locally >10 km<sup>2</sup>, and create severe problems for explorationists. This is because surface geochemistry, most notably elevated mercury values, seems to have little direct bearing on the precise positions of underlying HS, IS, or LS epithermal mineralization. Such features are difficult to explore unless the underlying epithermal deposit has been partially unroofed by erosion, as in the case of the largely concealed Pascua-Lama HS deposit (Sillitoe 1995b), or remnants of overprinted mineralization are preserved in the steam-heated zone. Reconnaissance drill testing of large steam-heated zones that lack observable evidence for associated mineralization is best guided by an understanding of district-scale fault geometry, based on field mapping, air photograph and satellite image interpretation, and geophysical evidence. Linear, resistivity and/or non-magnetic anomalies may be particularly instructive. Nonetheless, some visually prominent and well-known steam-heated alteration zones, like that at Cuprite, Nevada, which has been used as a remote-sensing test site for the past 40 years (Swayze et al. 2014, and references therein), have so far failed to reveal the location of any underlying epithermal deposit.

# Paleodepth of mineralization

Epithermal mineralization can occur in contact with or only a few meters beneath paleosurface or paleogroundwater table features, as exemplified by many epithermal deposits: HS (e.g., Yanacocha and Pascua-Lama; Harvey et al. 1999; Deyell et al. 2005), IS (e.g., Fruta del Norte; Leary et al. 2007), and LS (e.g., Hasbrouck Mountain, Ivanhoe, and McLaughlin; Graney 1987; Bartlett et al. 1991; Sherlock et al. 1995). Indeed, low-grade, bulk-tonnage HS gold ± silver

deposits invariably occur in close proximity to paleosurfaces, as reliably documented by preserved erosional outliers of steam-heated alteration above most of the largest examples (Sillitoe 1999).

Nonetheless, the ore-bearing parts of some epithermal systems, especially those of IS type, occur 300-500 m and, locally, even more beneath paleosurfaces, as determined using a combination of paleogroundwater table silicification and fluid inclusion geobarometry for a number of IS vein deposits in Mexico, including the Fresnillo district (Albinson 1988; Simmons 1991). Indeed, the existence of veins at depth beneath the paleogroundwater table silicification along the western side of the Fresnillo district was predicted by Sawkins (1988) and substantiated by discovery of the Juanicipio-Valdecañas silvergold ore shoots at a depth of ~500 m (Megaw 2010), using the data and analysis of Simmons (1991). In the National district, the tops of the LS veins are similarly deep, occurring  $\sim 270 \ge$ 440 m beneath the sinter and interbedded silicified epiclastic rocks that cap Buckskin Mountain (Vikre 2007). The Victoria and Teresa IS veins in the Mankayan district, Philippines, must be even deeper because their tops are 150-400 m beneath the present surface (Chang et al. 2011), which, given the absence of paleosurface features, would appear to be well below the paleosurface. These appreciable gaps between paleosurface and/or paleogroundwater table features and underlying precious metal ore-commonly occupied by low-temperature, intermediate argillic alteration (mixed-layer illite-smectite and/or smectite; e.g., Chang et al. 2011)-can lead to insufficiently deep drill testing of epithermal prospects.

Similar barren or poorly mineralized gaps exist in some HS systems, as exemplified by Quimsacocha in Ecuador. There, the subhorizontal ore zone, a lithologically controlled body of highly pyritic, silicified rock, is overlain by >100 m of largely barren vuggy residual quartz cut by minor structures with locally high-grade gold mineralization (Jones et al. 2005) to which must be added an inferred, but now-eroded steam-heat-ed blanket (Fig. 12).

A further complicating factor in the appraisal of epithermal prospects is the fact that the elevation of the tops of veins with respect to contemporaneous paleosurfaces, irrespective of whether or not paleosurface features are preserved, can vary appreciably throughout individual districts, as well illustrated by the Pachuca-Real del Monte, Mexico, and El Peñón, Chile, vein systems (Geyne et al. 1963; Warren et al. 2004; Fig. 13). Hence, although some veins may be at least partially exposed at the present surface, the nearby tops of others can lie at depths as great as 100-200 m or, in the case of Pachuca-Real del Monte, ~400 m (Fig. 13), possibly at least in part a direct result of paleogroundwater table fluctuations. Elsewhere, as demonstrated at Esquel, Argentina, widely separated, outcropping vein segments may connect at a depth beneath higher ground underlain by pre-mineralization cover rocks, implying that the original top of the vein is largely preserved (Sillitoe et al. 2002; Fig. 14). The tops of the blind veins at both Esquel and National are distinctively stibnite bearing (Vikre 1985; Sillitoe et al. 2002), in keeping with the upward concentration in epithermal systems of As, Sb, Tl, and Hg, elements that are mobile at lower temperatures (e.g., Ewers and Keays 1977; Hedenquist et al. 1996). Where all veins in a district are blind, as a result of formation beneath pre-mineralization cover, it is possible that there may have been little if any paleosurface evidence for their presence, at least vertically above them. However, hydrothermal outflow zones may be marked by low-temperature intermediate argillic alteration, travertine, and possibly, even sinter.

The fluvial and/or lacustrine sediments that accumulated syn-hydrothermally above some LS and a few IS systems are largely barren (e.g., Fruta del Norte), which poses a dilemma for explorationists who must determine if such sedimentary sequences, especially those of relatively limited areal extent, could cap blind epithermal deposits. Nonetheless, fluvio-lacustrine sequences elsewhere can contain strata-bound, disseminated orebodies (e.g., Ivanhoe, San Cristóbal). It is important though to bear in mind that bonanza-grade gold  $\pm$ 

Fig. 12 Longitudinal geologic section of the volcanic rockhosted Quimsacocha HS goldsilver-copper deposit, Ecuador, showing the subsurface position of the subhorizontal silicic orebody, which is localized by a tuffaceous horizon (taken from Jones et al. 2005). The hypothetical position of a now-eroded, steam-heated zone is also shown





Fig. 13 Differences in original elevations of ore shoots in the Pachuca-Real del Monte IS silver-gold district, Mexico (from Geyne et al. 1963). Note that the several hundred meter elevation range of ore-shoot tops

reflects differences in metal-precipitation conditions rather than postmineralization fault displacement

silver veins can also occur at depth beneath such low-grade, disseminated mineralization, a fact only appreciated much later in the case of Ivanhoe (Tewalt 1998). This near-surface, disseminated precious metal mineralization needs to be distinguished from similarly strata-bound, but somewhat more deeply formed epithermal orebodies hosted by permeable stratigraphic units, such as the non-welded ignimbrite at the Round Mountain LS disseminated gold deposit, Nevada (Sander and Einaudi 1990), where bonanza gold grades occur in structures below, within, and above the disseminated ore (Tingley and Berger 1985).

#### **Geochemical indicators**

The precious metal contents of sinters may bear no relationship to the potential of any underlying LS or IS epithermal mineralization because, as at Champagne Pool, gold and silver are commonly scavenged by the amorphous arsenic and antimony sulfide species. Since the latter are not everywhere or continuously formed in hot springs, sinter analysis would not appear to be a particularly effective exploration technique. Nonetheless, analysis of 13 late Miocene to recent sinters from throughout Japan revealed that those associated with known gold mineralization had significantly higher average gold contents (up to 827 ppb), commonly located in proximity to hot spring vent areas and accompanied by elevated As, Sb, and Fe values (Nakanishi et al. 2003; Fig. 15). Several sinters related to LS gold deposits elsewhere have also returned broadly similar gold values (e.g., up to 1161 ppb at Buckhorn; Jennings 1991). Arsenic and antimony contents of sinters, like those of gold, can be extremely variable, but are locally extraordinarily high (e.g., up to 12,000 and 3000 ppm, respectively; Hedenquist and Henley 1985; Hayashi 2013). Furthermore, even calcite layers in the mixed silica-carbonate sinters at Savo, Solomon Islands, are enriched in As and Te (Smith et al. 2011). In conclusion, therefore, it would be reasonable to take elevated precious metal and metalloid tenors in proximal sinters as an encouraging sign of subjacent mineralization, although their absence does not necessarily condemn a prospect. Nonetheless, many sinters apparently lack associated precious metal ore at depth (e.g., McGinness Hills, Nevada and Ikiryu; Casaceli et al. 1986; Nakanishi et al. 2001).

Steam-heated zones typically also lack geochemical anomalies for precious metals and the associated element suite, other than mercury, unless they either are overprinted by or overprint precious metal mineralization, as discussed earlier. At the Purén IS silver-gold deposit, on the margin of the La



**Fig. 14** Longitudinal section of the Galadriel-Julia vein at Esquel, Deseado massif, Patagonia, Argentina, showing the subsurface position of much of the LS gold-silver orebody (see Sillitoe et al. (2002) for further

details). Note the presence of the basaltic andesite sill above the geologically blind portion of the Julia ore shoot



Fig. 15 Logarithmic histograms of gold contents of samples from 13 silica sinters in Japan (from Nakanishi et al. 2003). a Sinters associated with known gold deposits. b Sinters without known gold deposits. Note that the sinters associated with known gold mineralization have significantly higher gold contents, although contributions from overprinted auriferous veinlets cannot be ruled out

Coipa district, silver (up to 3 ppm) and  $100 \times \text{Ag/Mo}$  (up to 17) values in steam-heated alteration define both the position and overall trend of the ore-grade, precious metal mineralization located 40 m below (Arribas et al. 2005). These low tenor silver values are inherited from pre-existing mineralization that was overprinted as a consequence of syn-hydrothermal descent of the paleogroundwater table.

The gaps between the tops of blind veins and contemporaneous paleosurfaces can be largely barren in some LS and IS epithermal systems, although low-order geochemical anomalism, typically for arsenic and antimony, may be present, especially on approach to the veins. In the case of the bonanza-grade Quebrada Colorada LS vein in the El Peñón district (Fig. 16), 100 s of ppm As and 10 s of ppm Sb extend for >100 m above its geologically blind top, although gold and silver values are insignificant (Warren et al. 2004). Similarly, the concealed Victoria IS veins, on the periphery of the Far Southeast porphyry copper-gold deposit, have up to 189 ppm As (background 0.2 ppm) at the present surface, 250–350 m above their tops (Chang et al. 2011).

Where fluvio-lacustrine sedimentation is suspected to have accompanied IS or LS epithermal activity, the recommended approach is diligent search for subtle alteration features and accompanying geochemical anomalism, particularly where faults or fractures cut the sedimentary sequences. An instructive example is provided by the structurally localized rib of silicified conglomerate situated 200 m stratigraphically above the Fruta del Norte IS deposit, which returned highly anomalous arsenic (up to several thousand ppm), antimony, and mercury values but only low-order (typically <0.3 g/t) gold values (Leary et al. 2007; Fig. 10).



**Fig. 16** Vertical geochemical gradients in rocks surrounding the Quebrada Colorada LS gold-silver vein in the El Peñón district, Chile (taken from Warren et al. 2004). *Bars* show Au, Ag, As, and Sb values averaged over 50-m intervals for drill core samples within 150 m of the vein; white bands represent median values. Note the arsenic and antimony anomalism above the geologically blind ore shoot, which has its top at ~1815 m above sea level (~80 m below surface)

# **Concluding remarks**

Epithermal systems can be characterized by a variety of paleosurface and immediately subsurface features of great utility in the search for concealed HS, IS, and LS precious metal deposits. These features reflect district-scale hydrology, particularly the paleogroundwater table position, which may lie subsurface, at the surface or, in the location of a lake, above the surface. Distinction between these features—particularly sinter, paleogroundwater table silicification, lacustrine amorphous silica sediments, lacustrine chert, silicified fluviolacustrine sediments, HS silicic ribs, and IS/LS veins—all composed of superficially similar chalcedony as a result of aging and recrystallization, can facilitate understanding of epithermal districts and guide the search for concealed ore (Table 4; Fig. 11).

The shallow parts of epithermal systems undoubtedly still offer great promise for blind precious metal deposits of both bulk-tonnage disseminated and high-grade vein types. When compared to deposits concealed beneath post-mineralization cover, those beneath paleosurface features have the distinct advantage that their locations should be more readily amenable to geologic deduction. However, this advantage is somewhat limited because partial preservation of epithermal paleosurface features tends to be geographically confined to either extensional, commonly back-arc regions, characterized by semi-arid climates and relatively low erosion rates, such as the Great Basin, Mexican Altiplano, Deseado massif, and Drummond Basin, or the highly arid central Andes. Beyond such regions, where rainfall and erosion rates are higher, paleosurface remnants tend to be relatively uncommon, isolated occurrences unless the epithermal systems are either extremely young ( $\leq 1$  Ma) and still at least mildly active (e.g., Cerro Blanco, Hishikari, Osorezan, Ladolam) or underwent prompt burial and relatively recent exhumation (e.g., Republic, El Dorado, Fruta del Norte, Ikiryu). Nevertheless, given the widespread existence of appreciable barren gaps between paleosurface features and tops of IS and LS veins or HS silicic replacement bodies as well as the different elevations of ore shoots in many epithermal districts, the likelihood of concealed subsurface mineralization is high even where paleosurface features have been eroded.

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